

Diffusion studies by QENS measurements approaching the 'ideal' situation

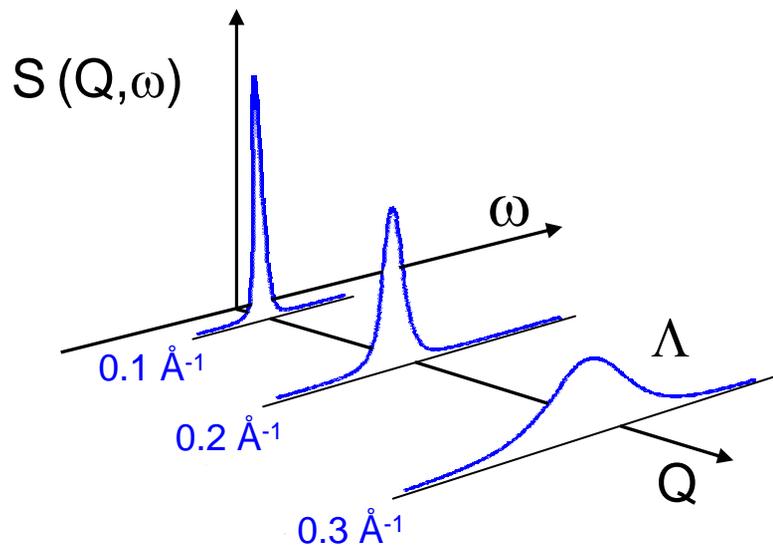
Hervé Jobic

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69626 Villeurbanne

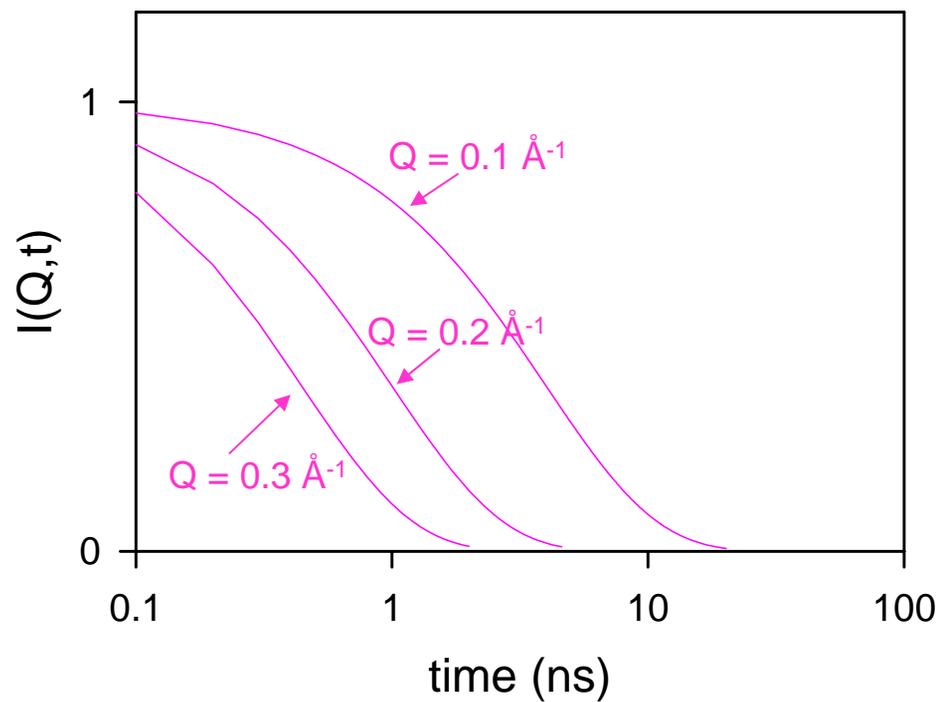


Frankfurt, Oct 2006



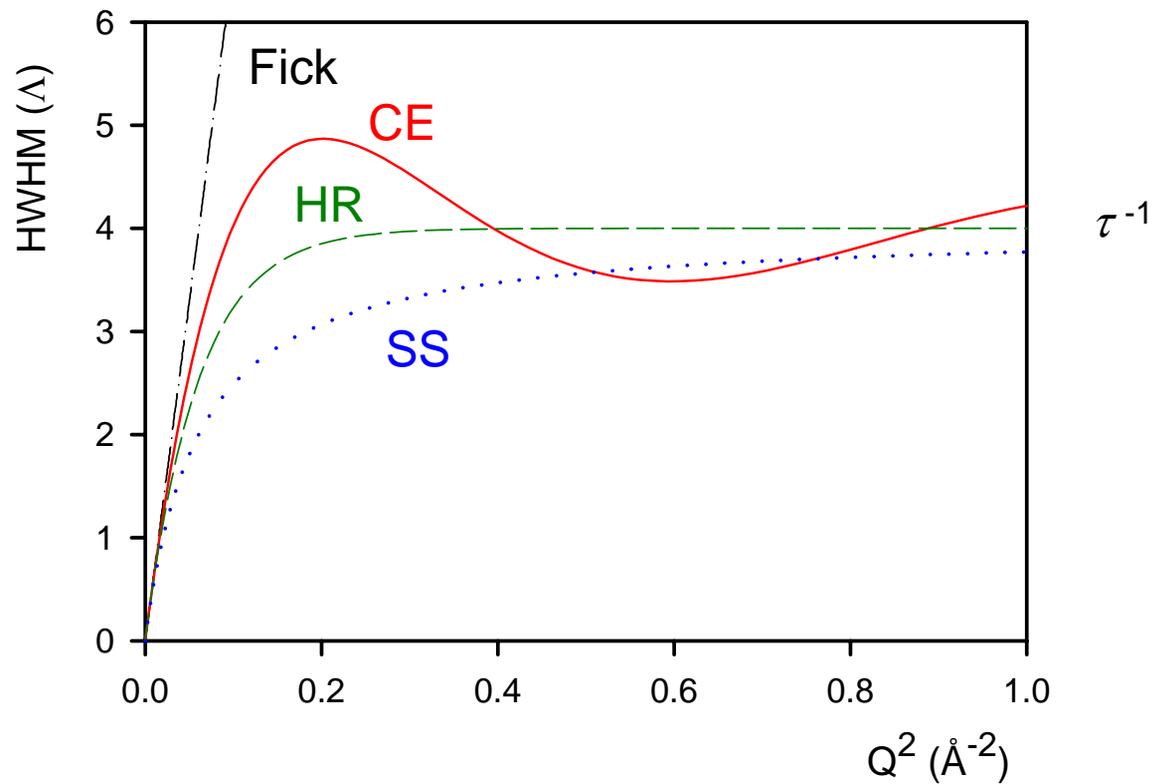


Quasi-elastic
scattering:
QENS
TOF - BS



Neutron spin echo
NSE

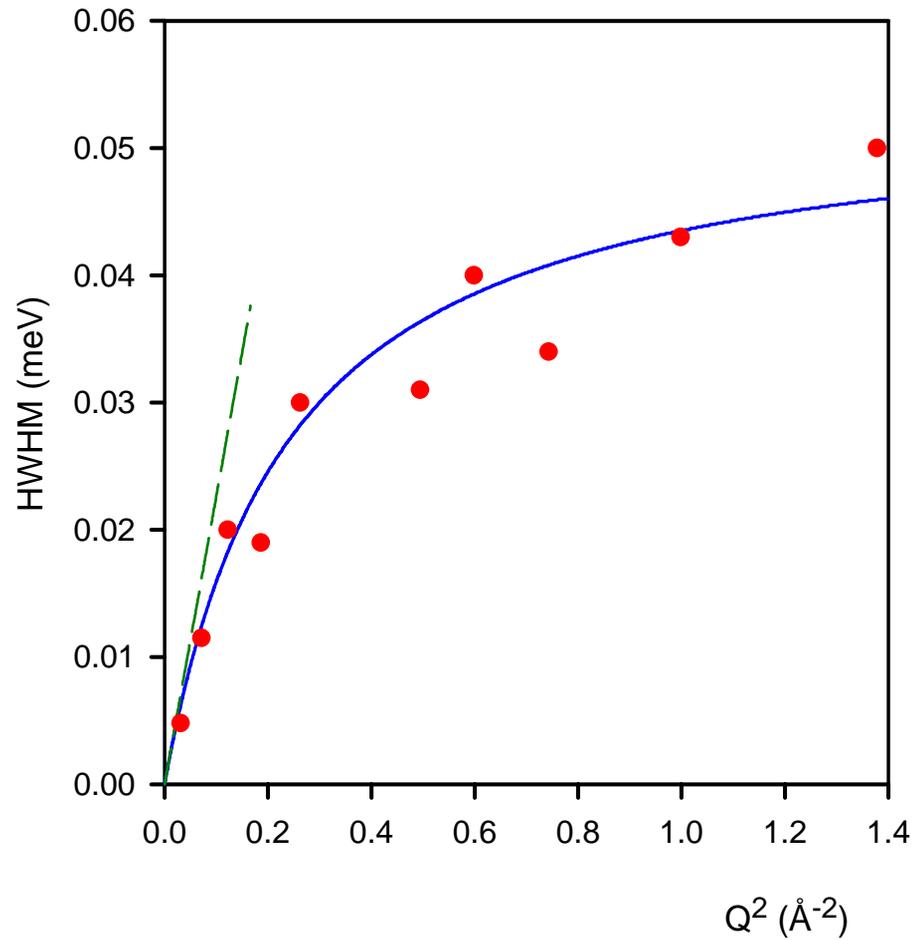
Fick: DQ^2



$$\left. \begin{array}{l} \langle r^2 \rangle^{1/2} = 10 \text{ \AA} \\ \tau \equiv \end{array} \right\} \text{ same } D$$

NH₃ / silicalite

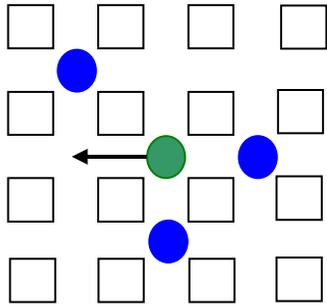
4.3 molecules / u.c. T = 360 K



SS model

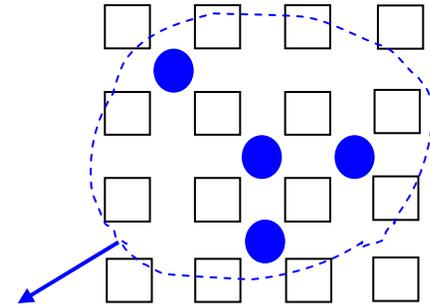
$$\langle d \rangle^{1/2} = 5 \text{ \AA}$$

$$\tau = 12 \text{ ps}$$



$$\frac{\partial G_S(\mathbf{r}, t)}{\partial t} = D_s \nabla^2 G_S(\mathbf{r}, t)$$

hydrogenated molecules



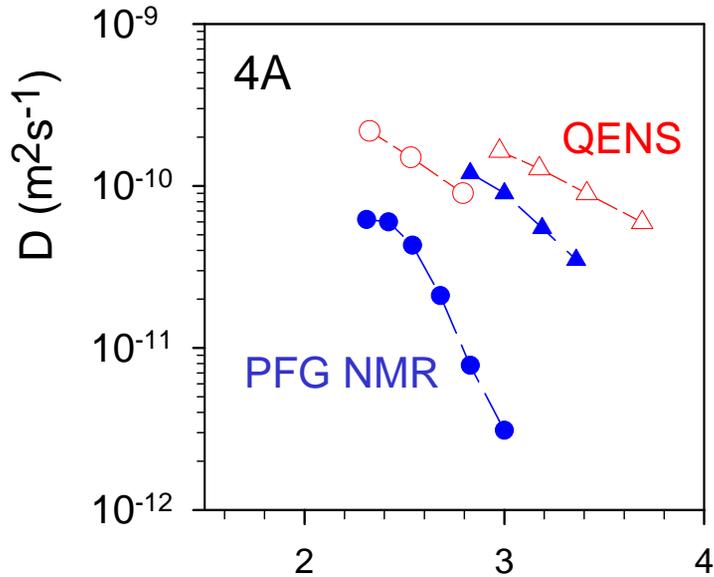
$$\frac{\partial \rho(\mathbf{r}, t)}{\partial t} = D_t \nabla^2 \rho(\mathbf{r}, t)$$

deuterated molecules
+ O₂, N₂, CO₂, SF₆...

$$D_t = D_0 \frac{d \ln p}{d \ln c} = D_0 \Gamma$$

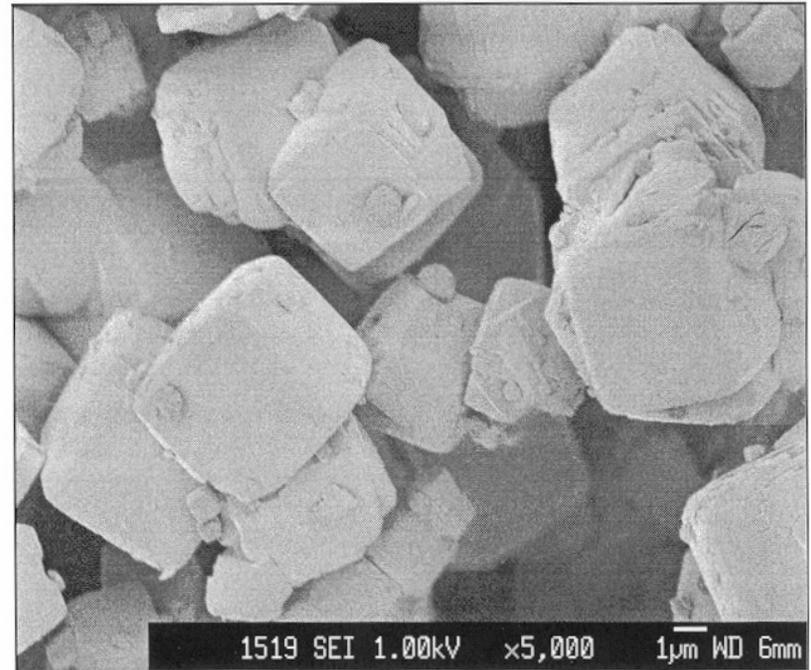
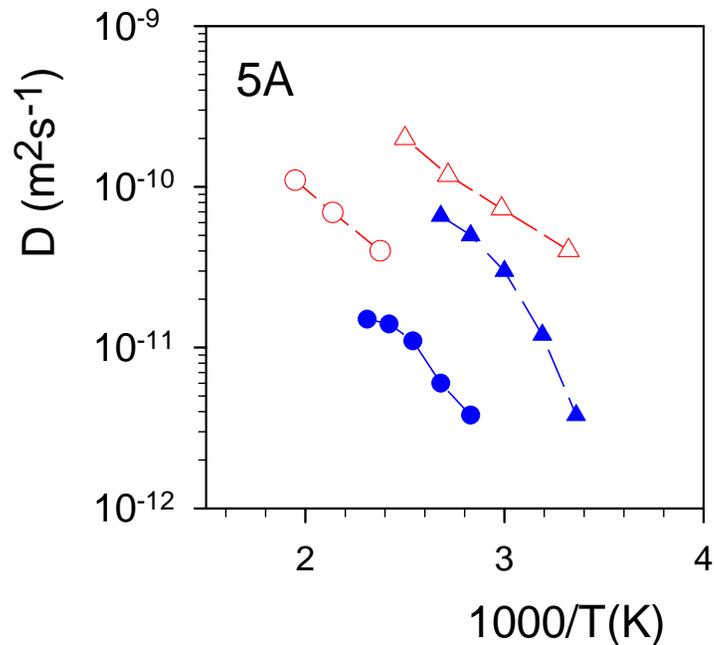


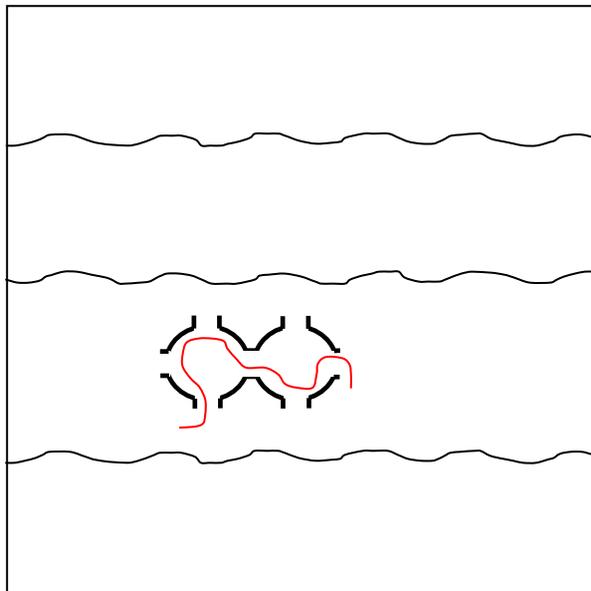
H₂O in type A zeolites



circles: 5 H₂O per α -cage

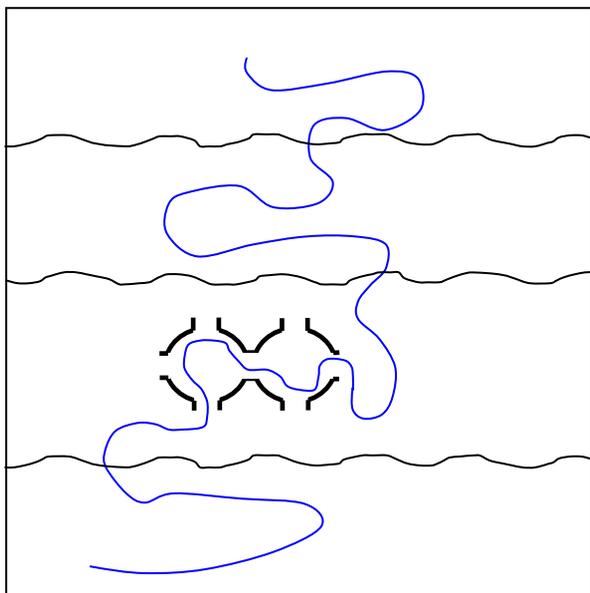
triangles: 15 H₂O per α -cage





QENS

≈ nm

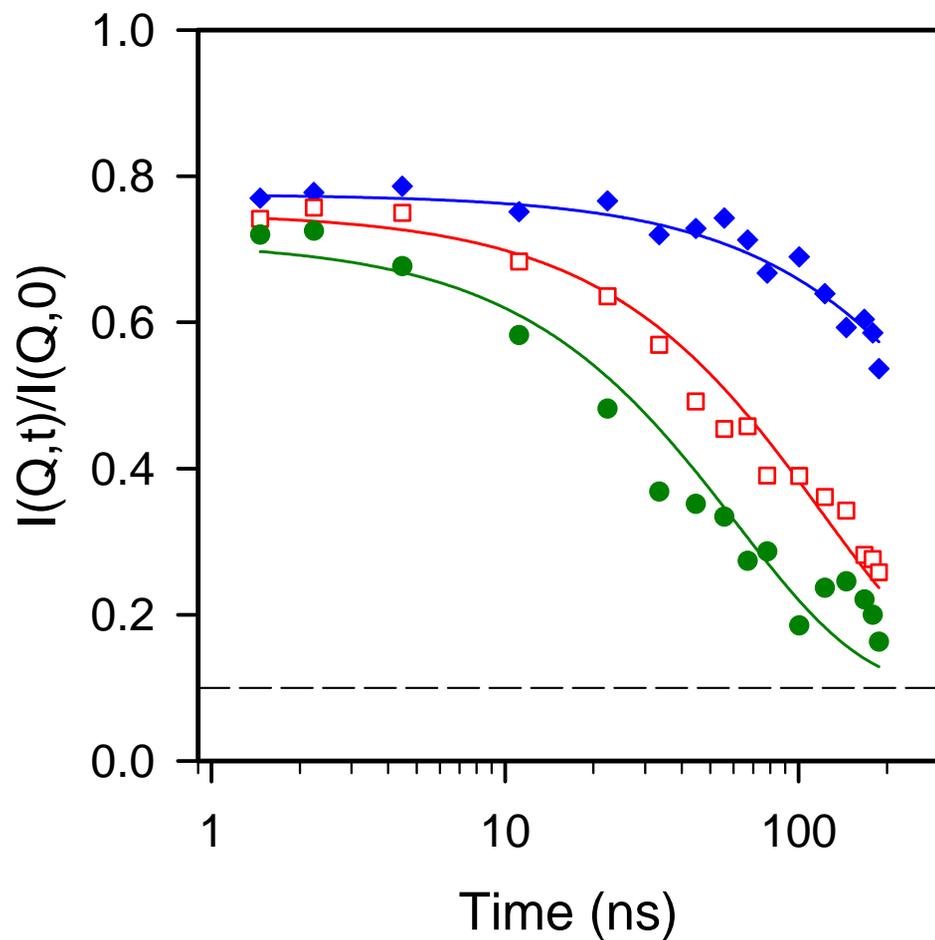


PFG NMR

≈ μm

n-alkanes in NaCaA (5A)

($T = 475$ K, $Q = 0.2 \text{ \AA}^{-1}$)

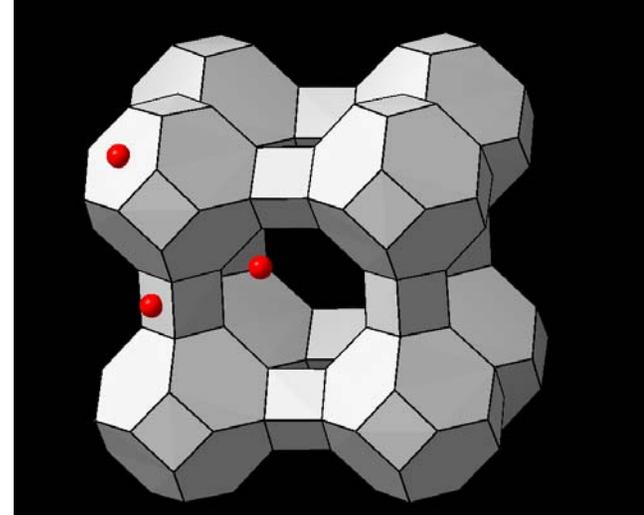


octane

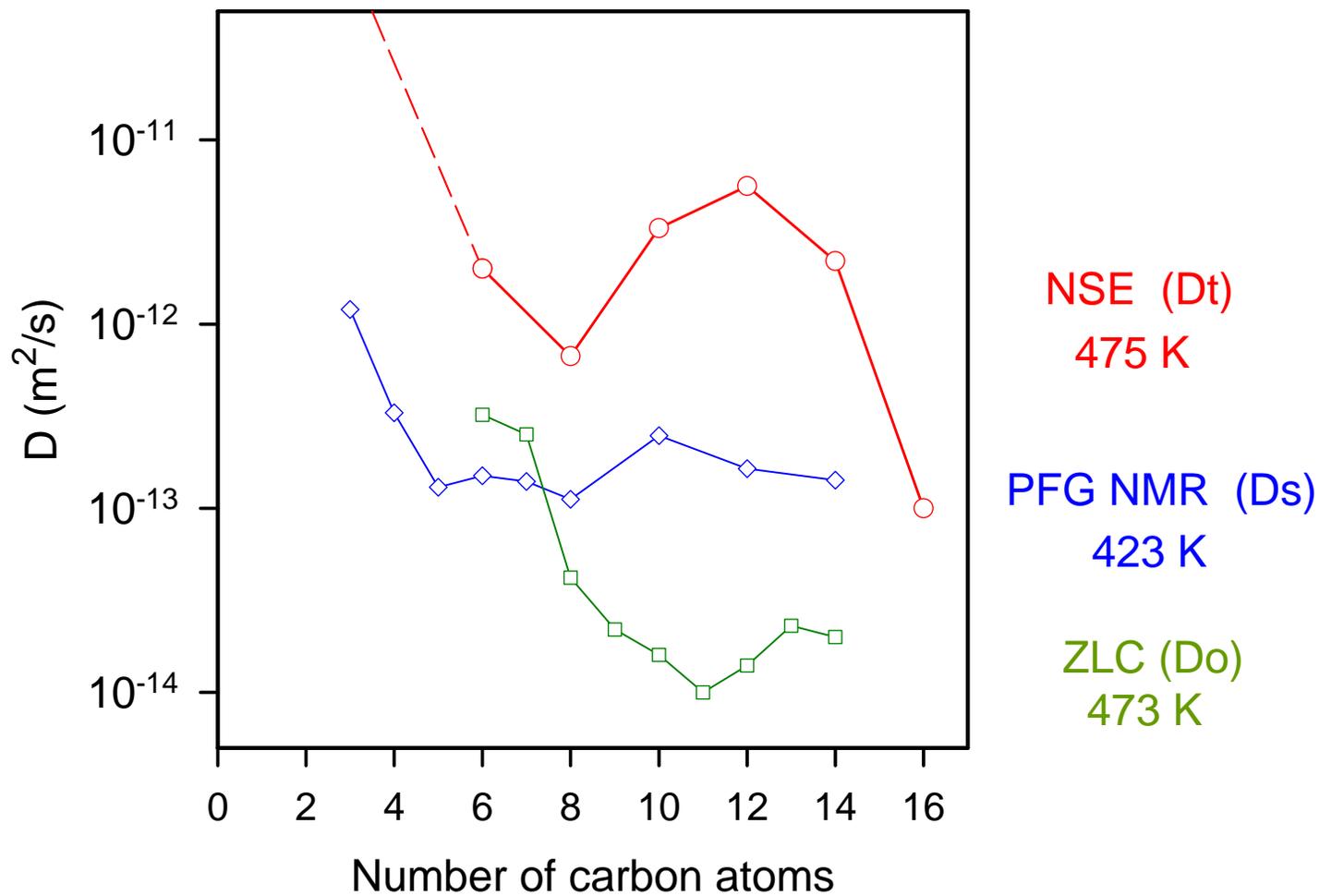
decane

dodecane

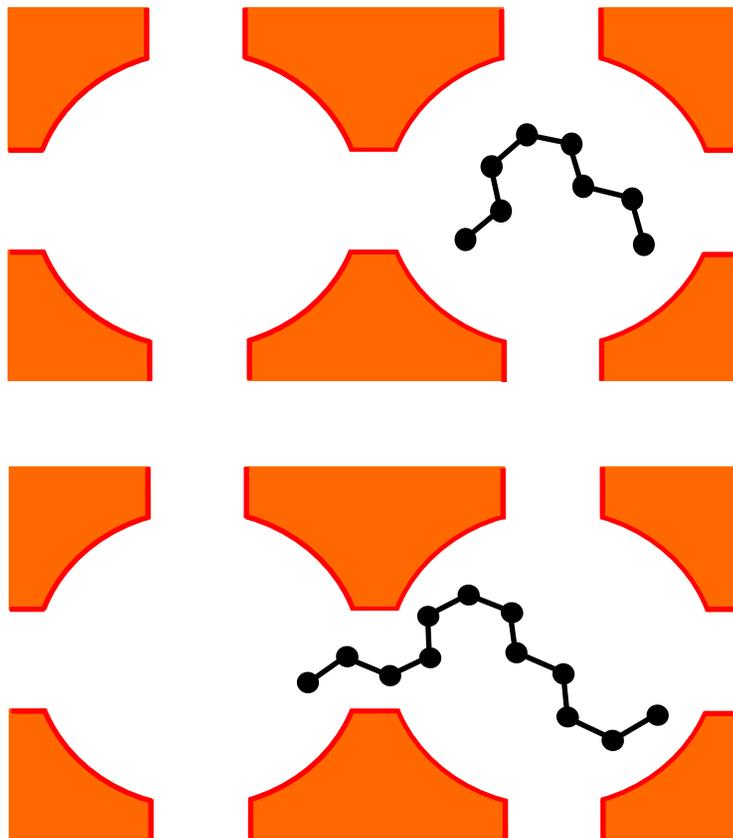
12 C / α -cage



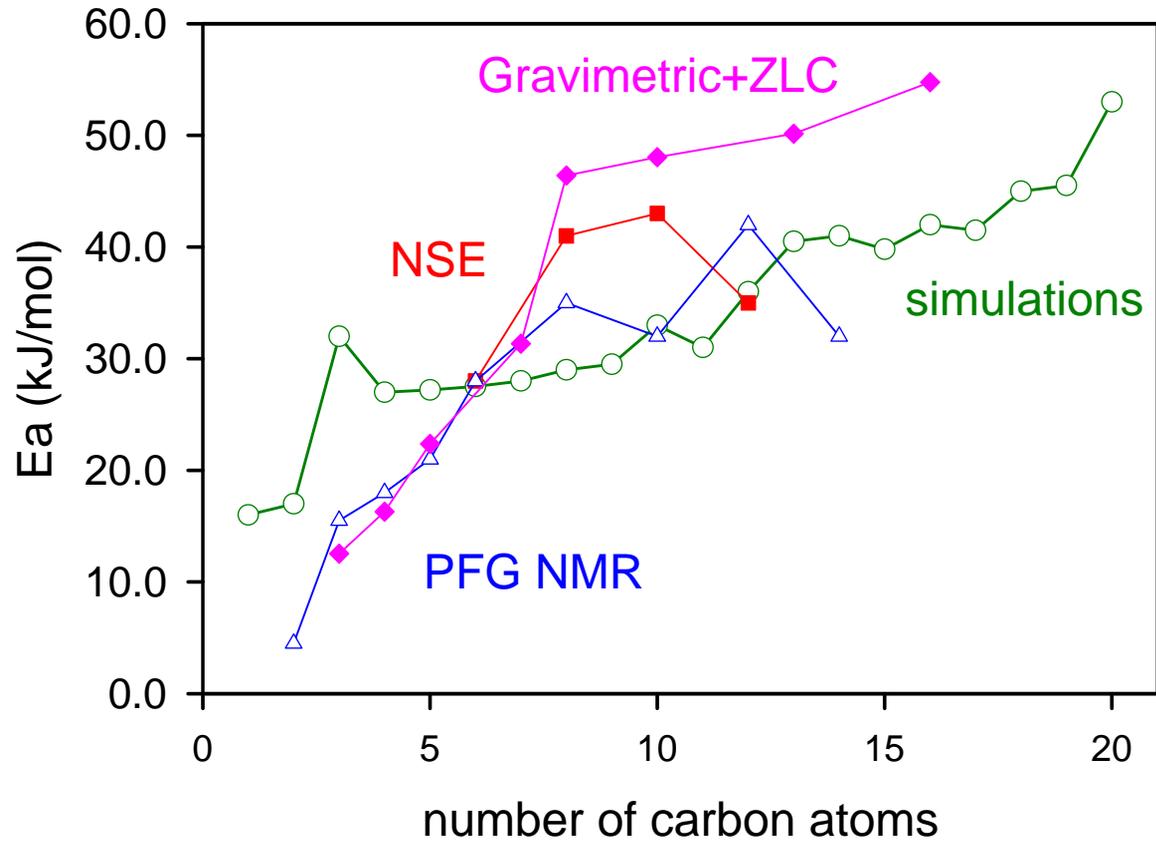
n-alkanes in 5A



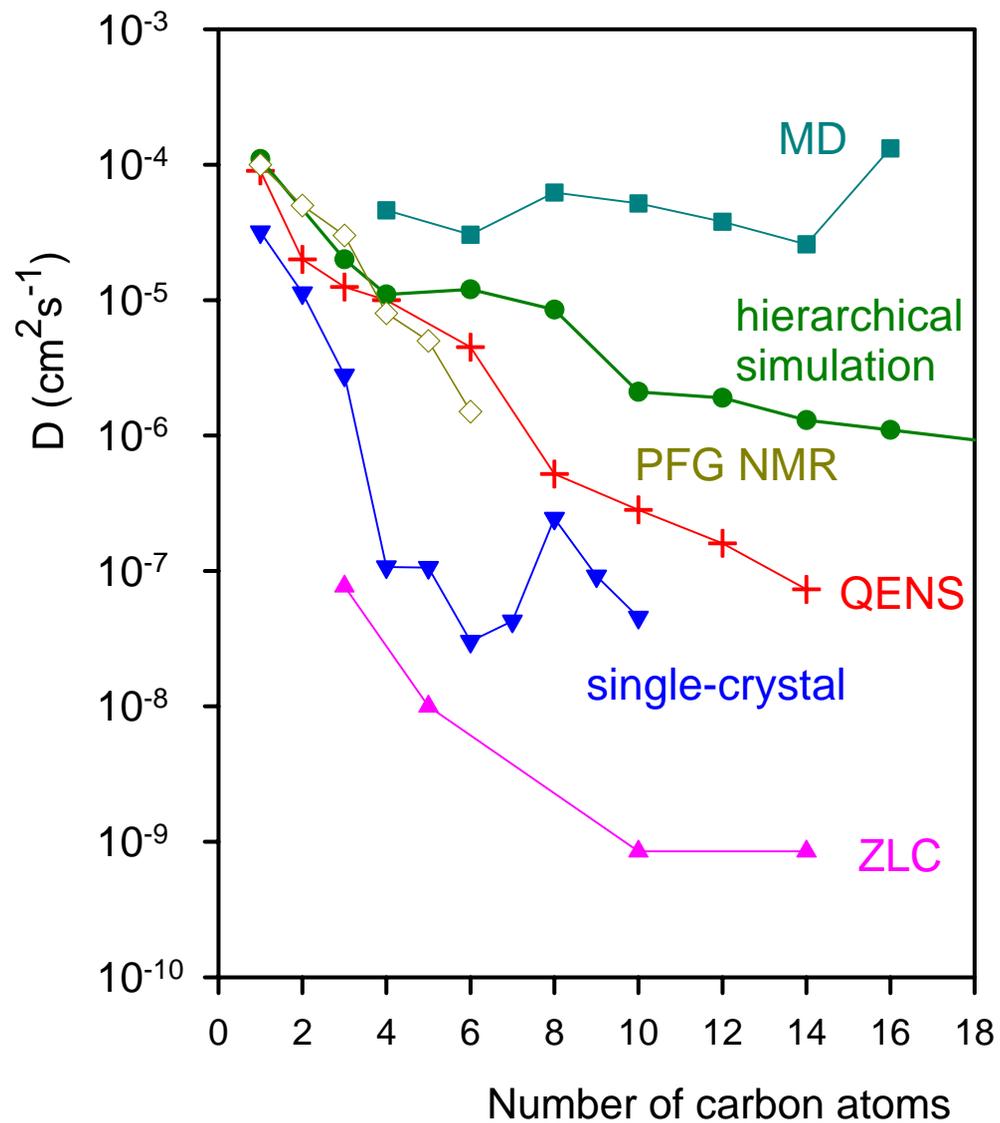
'Window effect'



E_a *n*-alkanes in 5A



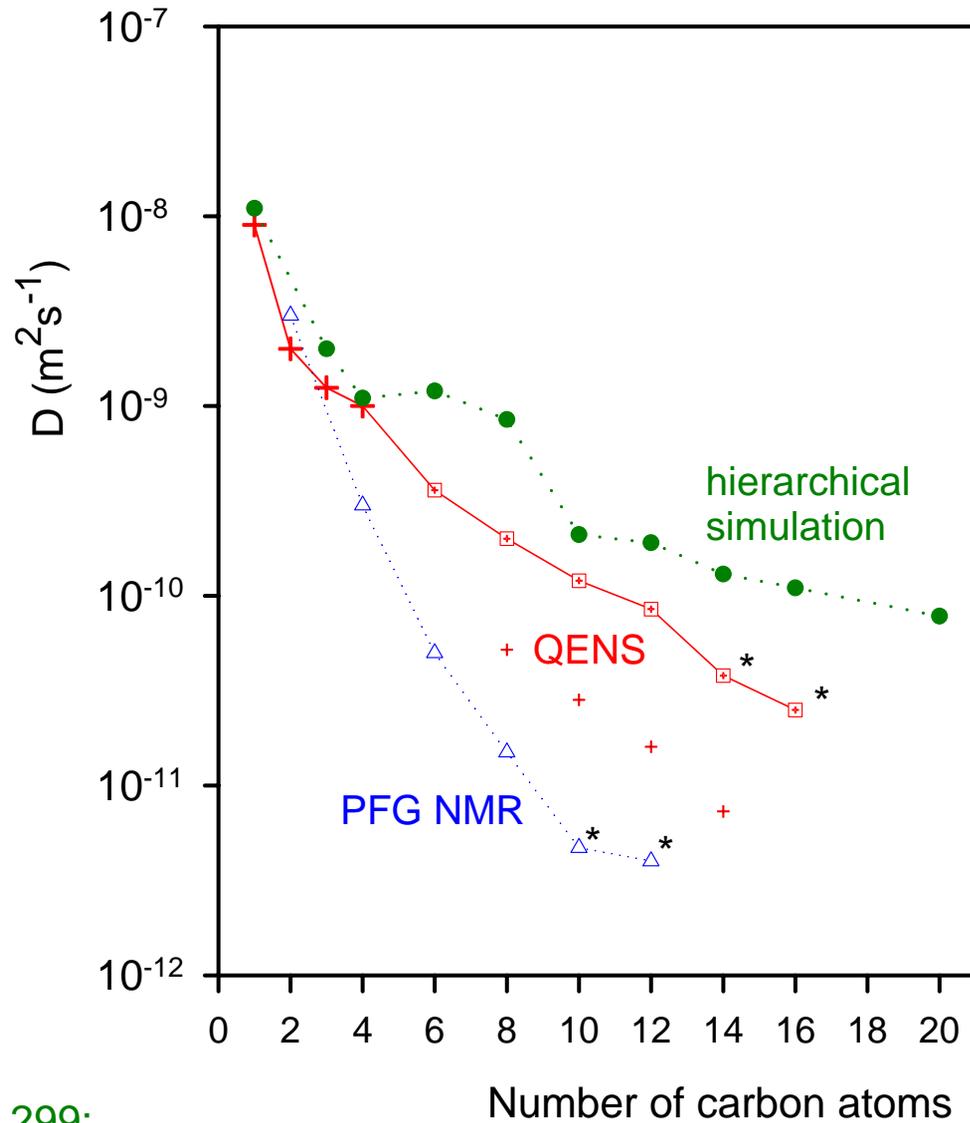
Ds *n*-alkanes / MFI
(T = 300 K)



Ds *n*-alkanes / silicalite
(T = 300 K)

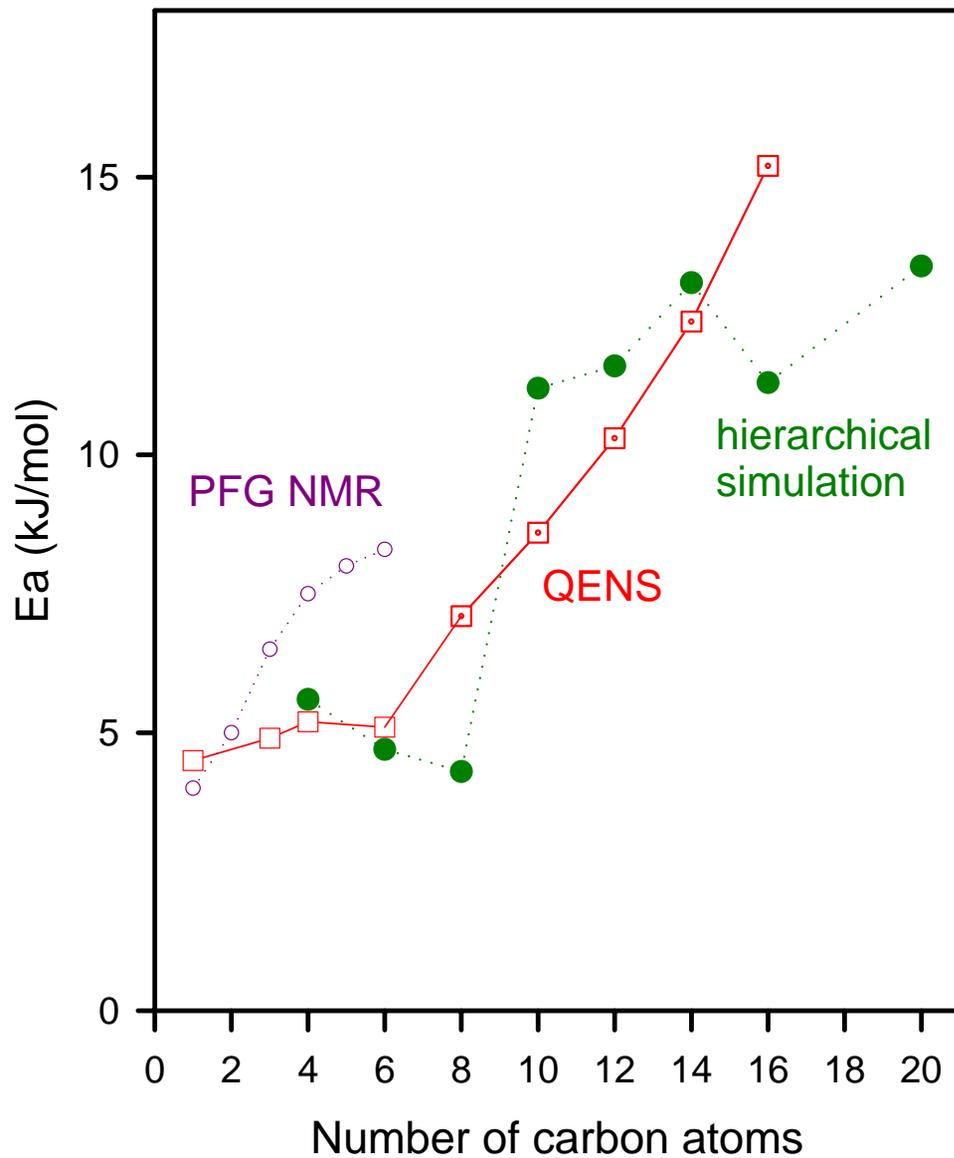


50x160 μm

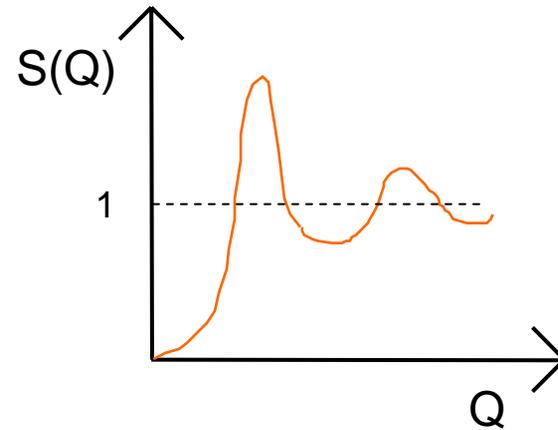
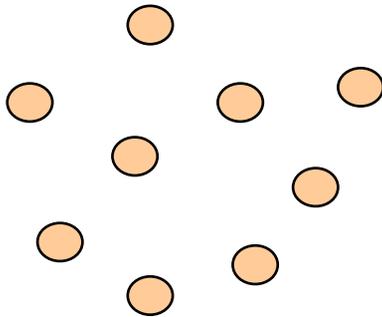
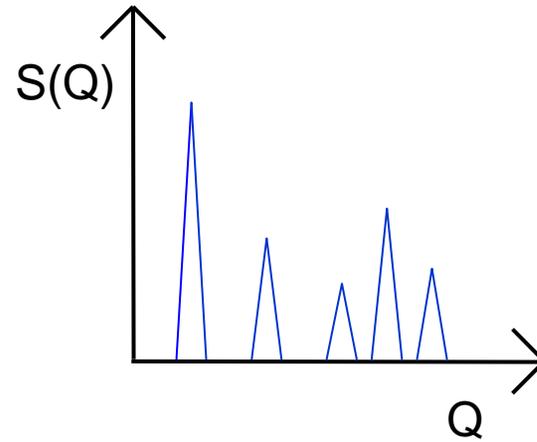
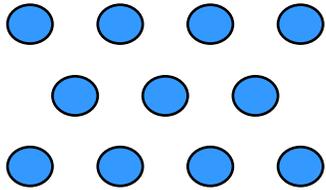


Microp. Mesop. Mater. 90 (2006) 299;
J.Phys. Chem. B 110 (2006) 1964

Ea *n*-alkanes / silicalite



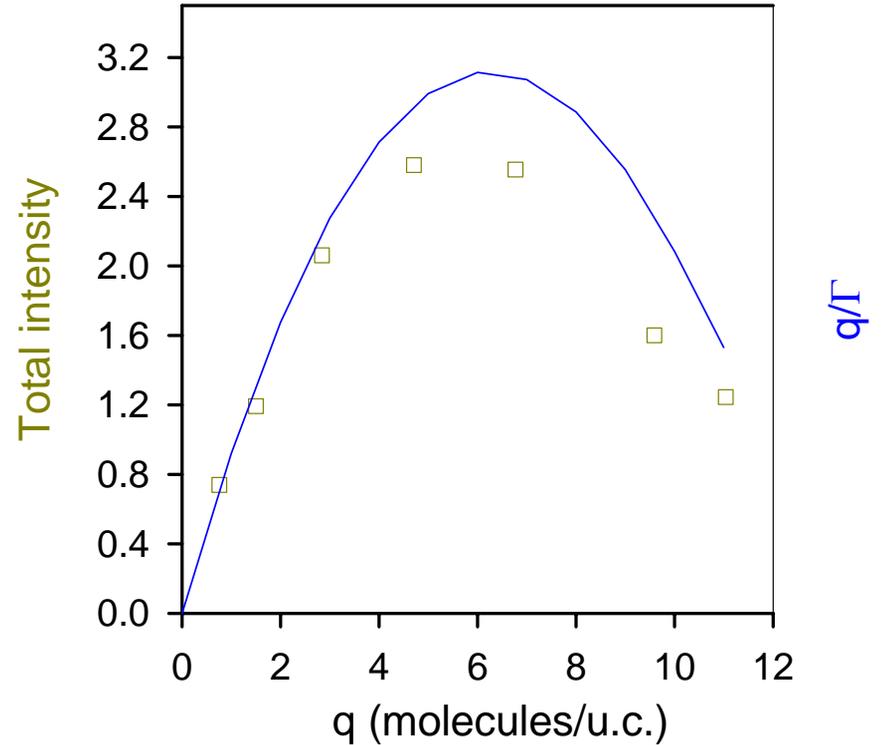
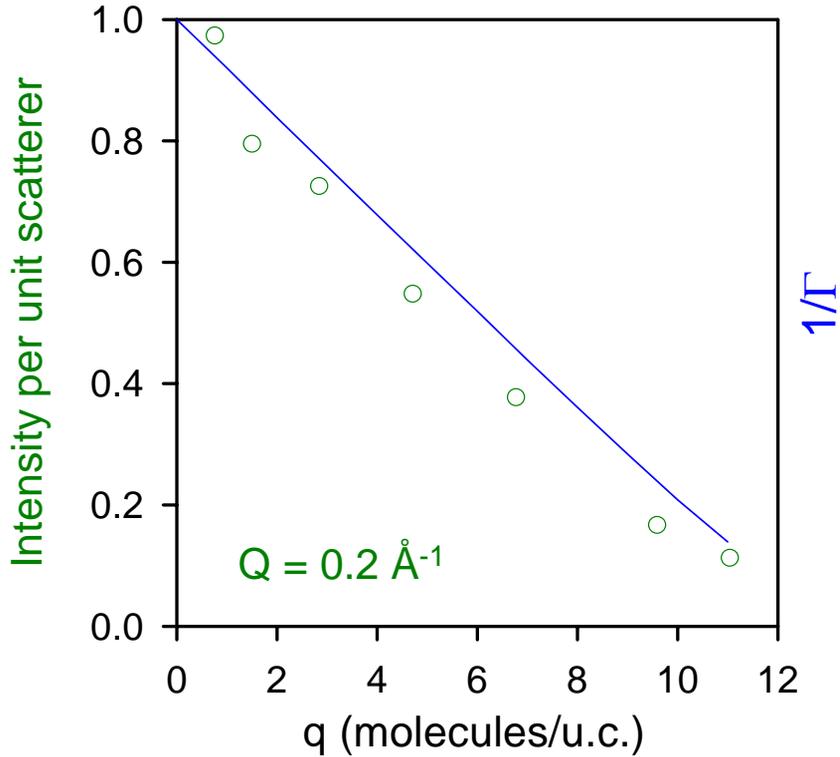
Coherent scattering



$$S(Q) = 1 + \int g(r) \exp(i\mathbf{Q}\cdot\mathbf{r}) d\mathbf{r}$$

$$S(Q) = \int S(Q, \omega) d\omega$$

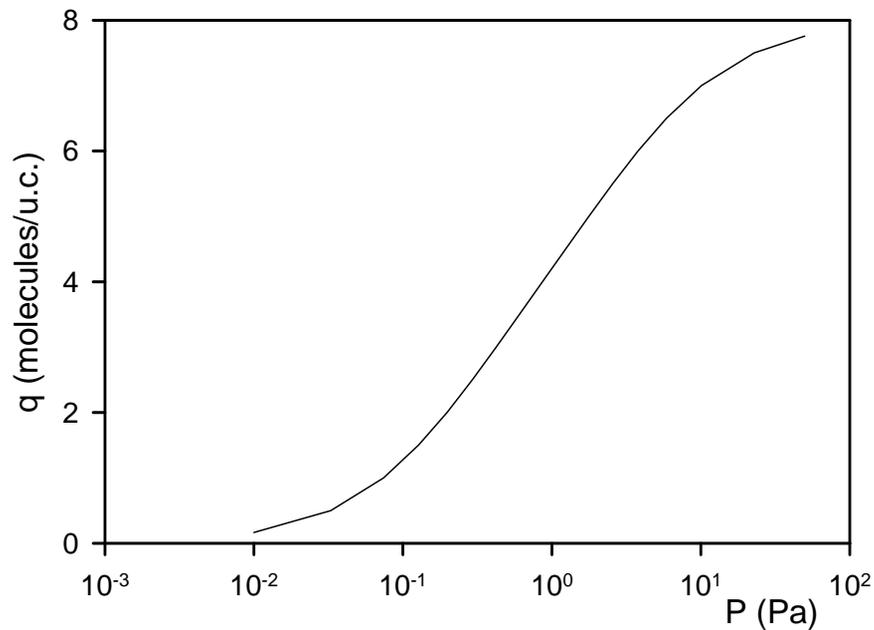
C₂D₆ in silicalite @ 300K



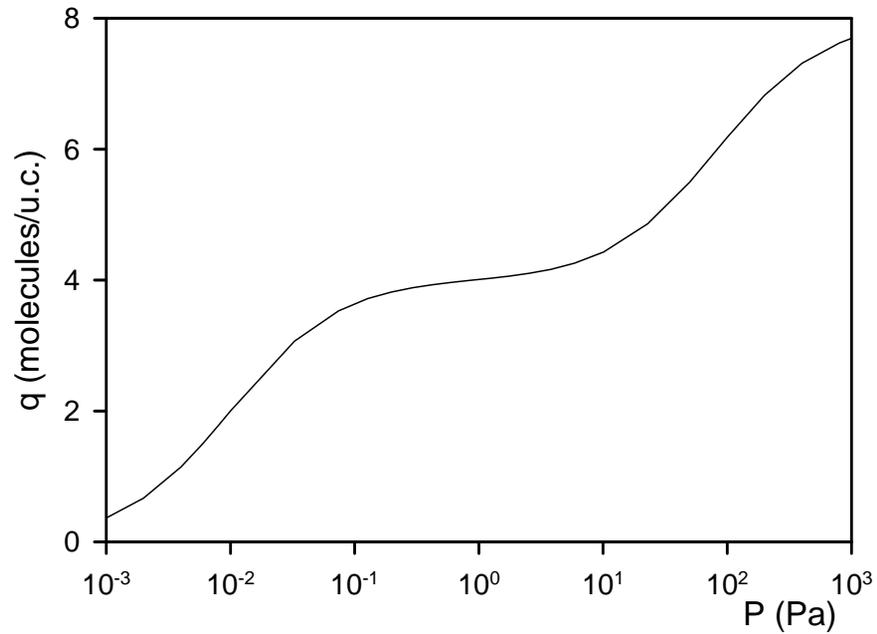
$$S(Q)_{Q \rightarrow 0} = \frac{\langle N^2 \rangle - \langle N \rangle^2}{\langle N \rangle} = \rho k_B T \kappa_T$$

$$\kappa_T = \frac{1}{\rho k_B T} \left(\frac{d \ln q}{d \ln p} \right)$$

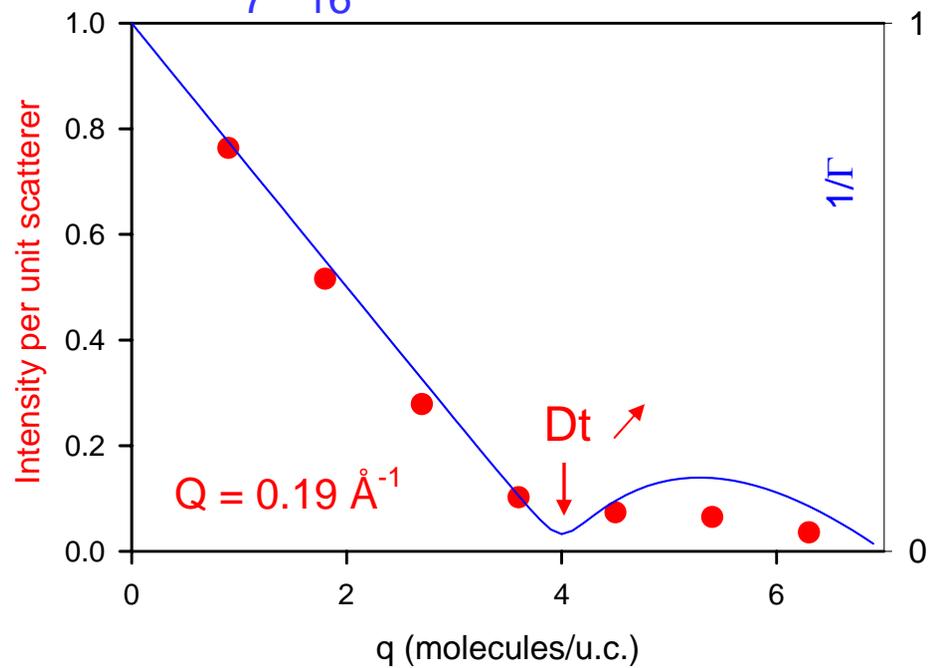
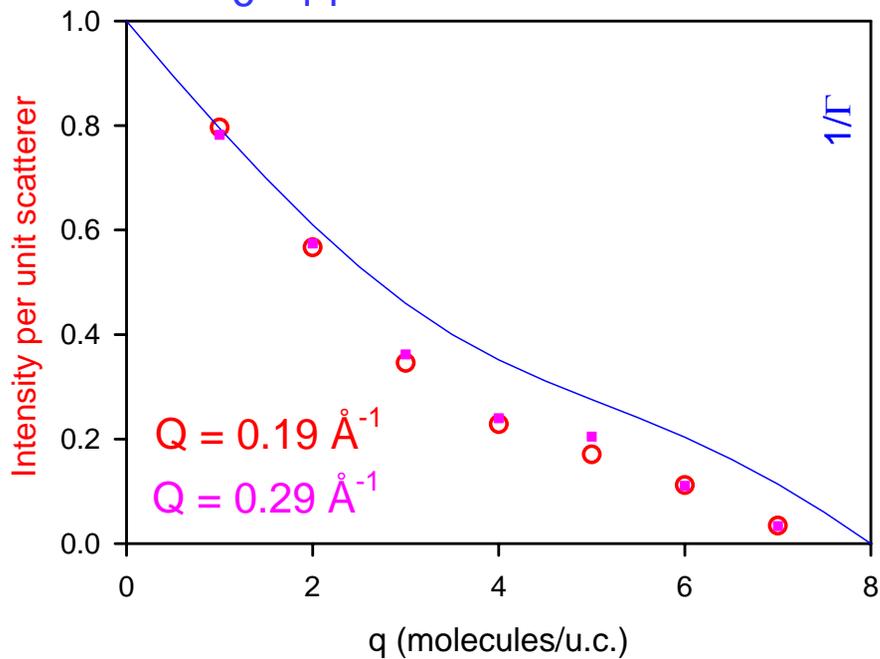
$$S(Q)_{Q \rightarrow 0} = \frac{d \ln q}{d \ln p} = \frac{1}{\Gamma}$$



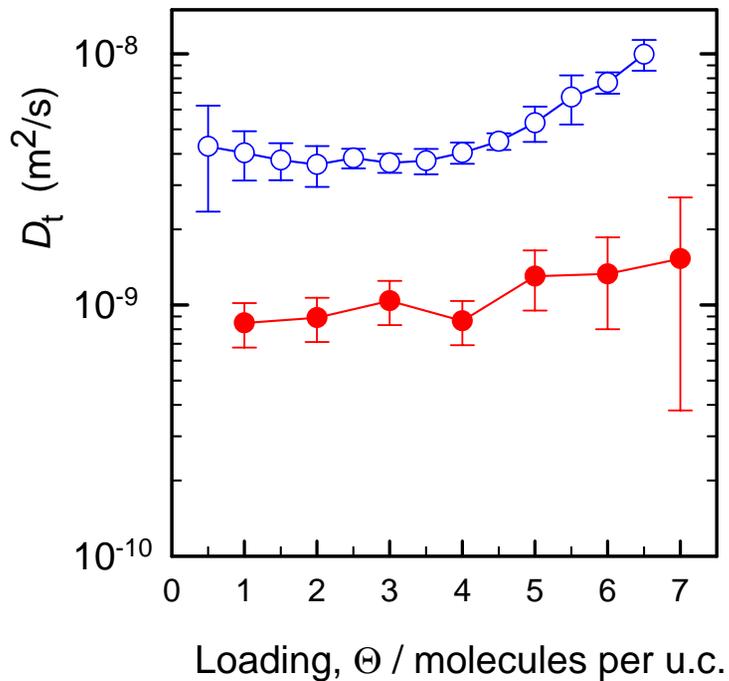
$n\text{-C}_6\text{D}_{14}$ / silicalite @ 300K



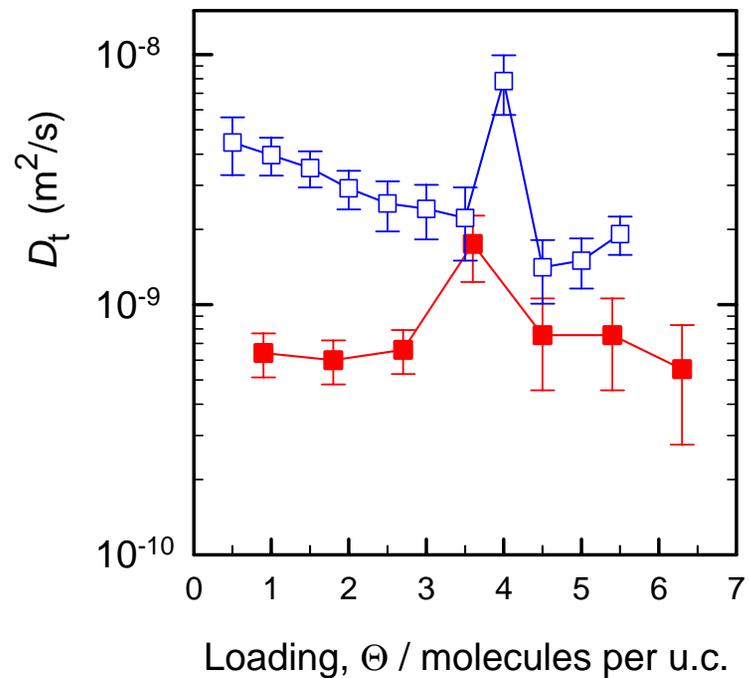
$n\text{-C}_7\text{D}_{16}$ / silicalite @ 300K



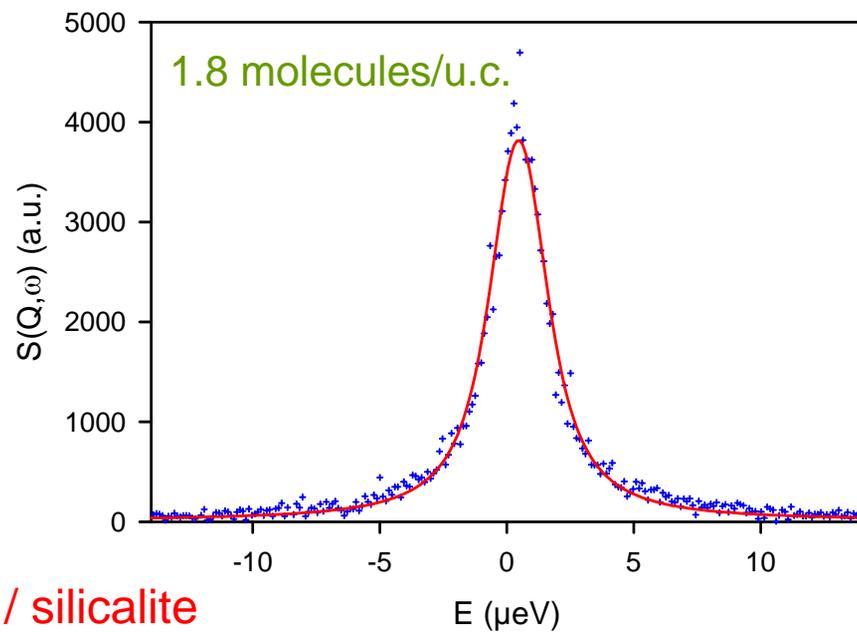
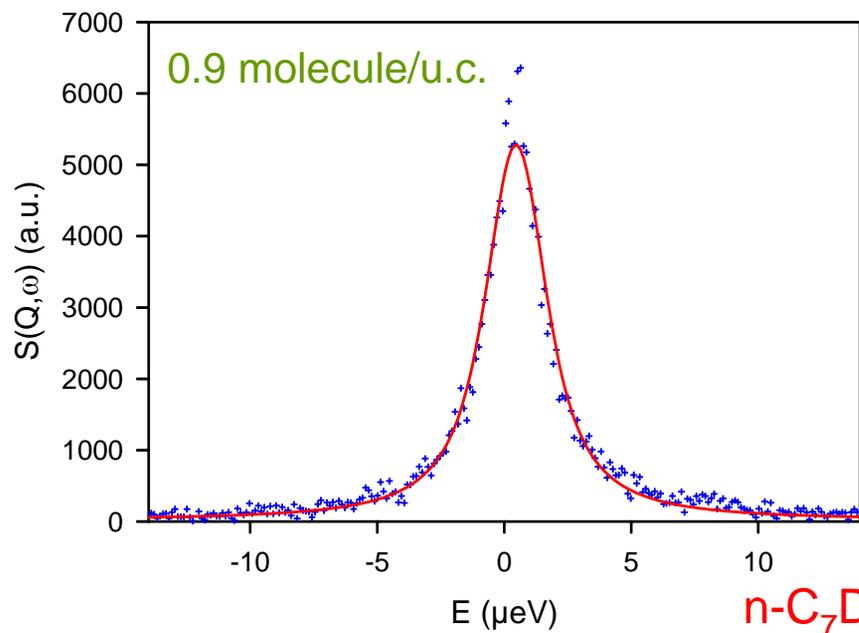
D_t C₆/silicalite @ 300K



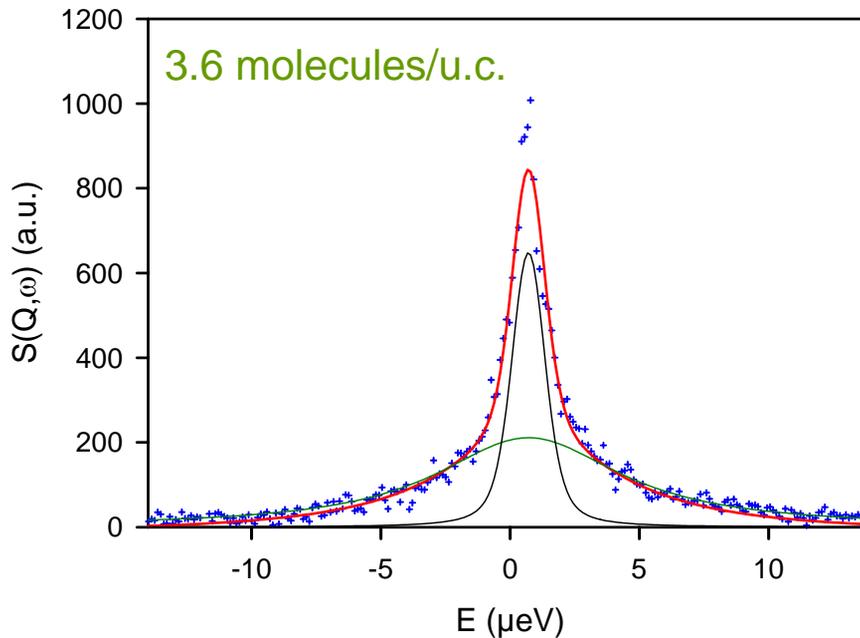
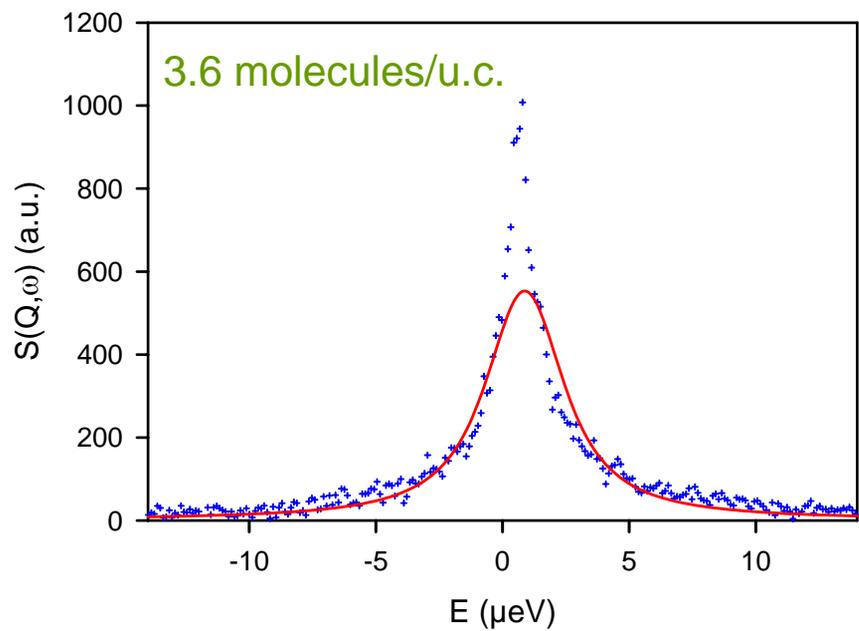
D_t C₇/silicalite @ 300K



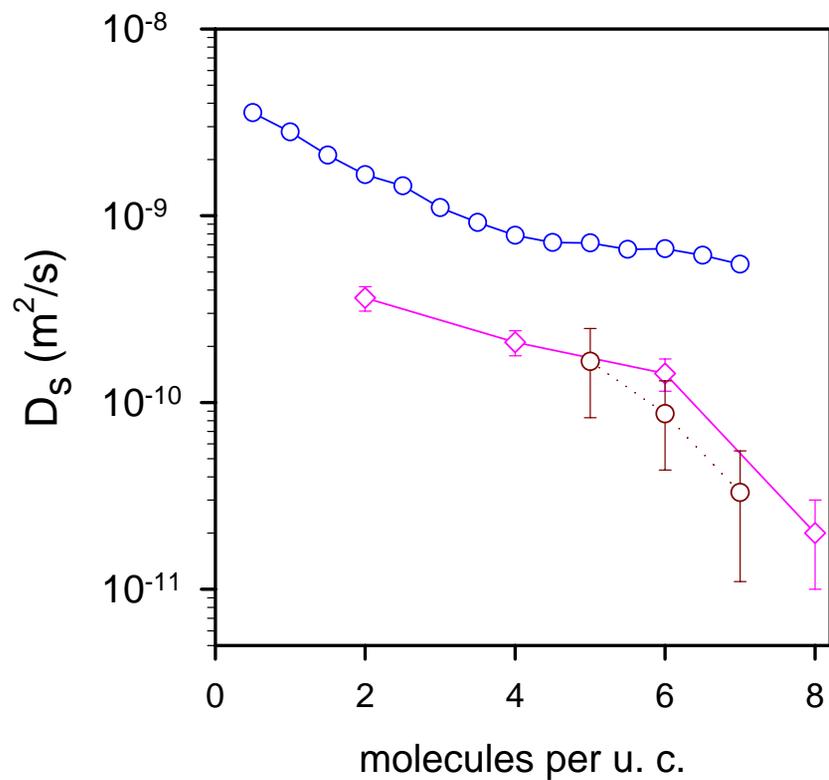
MD
QENS



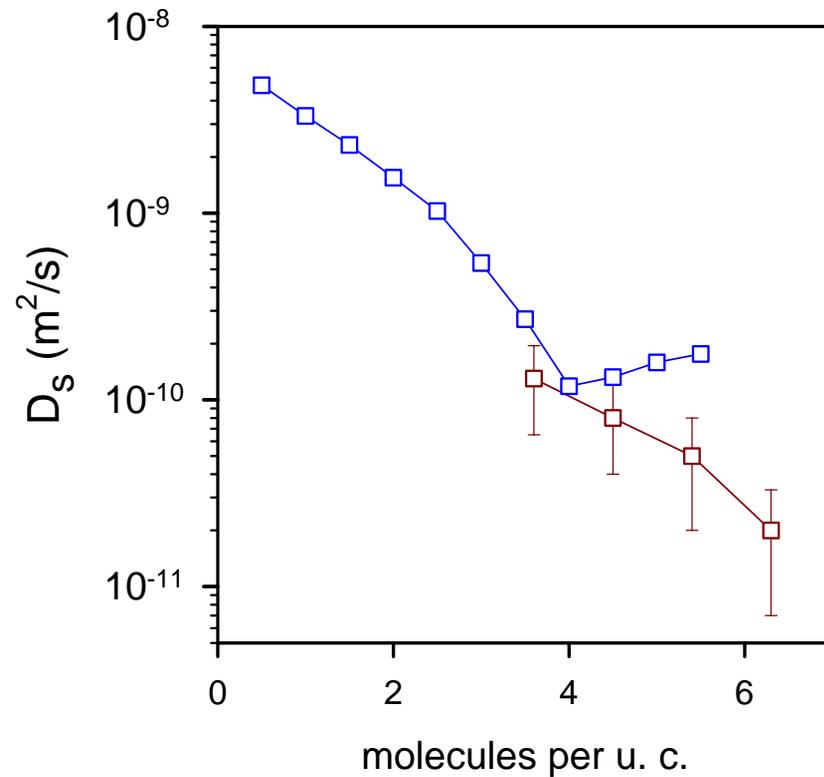
$n\text{-C}_7\text{D}_{16}$ / silicalite
 $Q = 0.19 \text{ \AA}^{-1}$



D_S C₆/silicalite @ 300K



D_S C₇/silicalite @ 300K

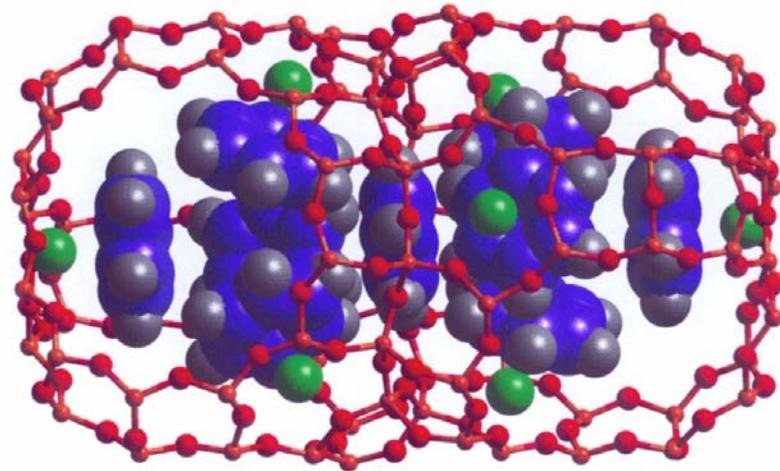
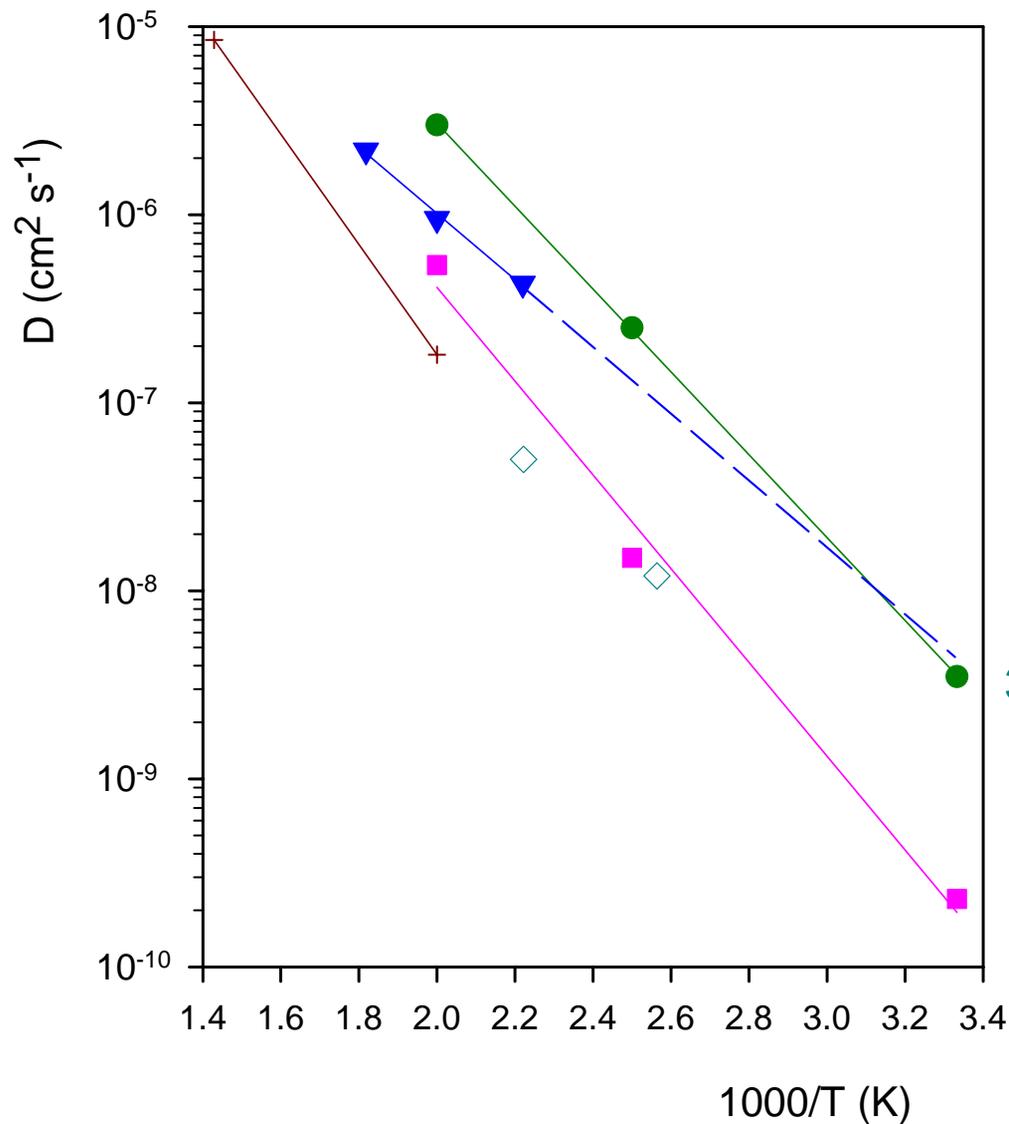


MD

inc QENS

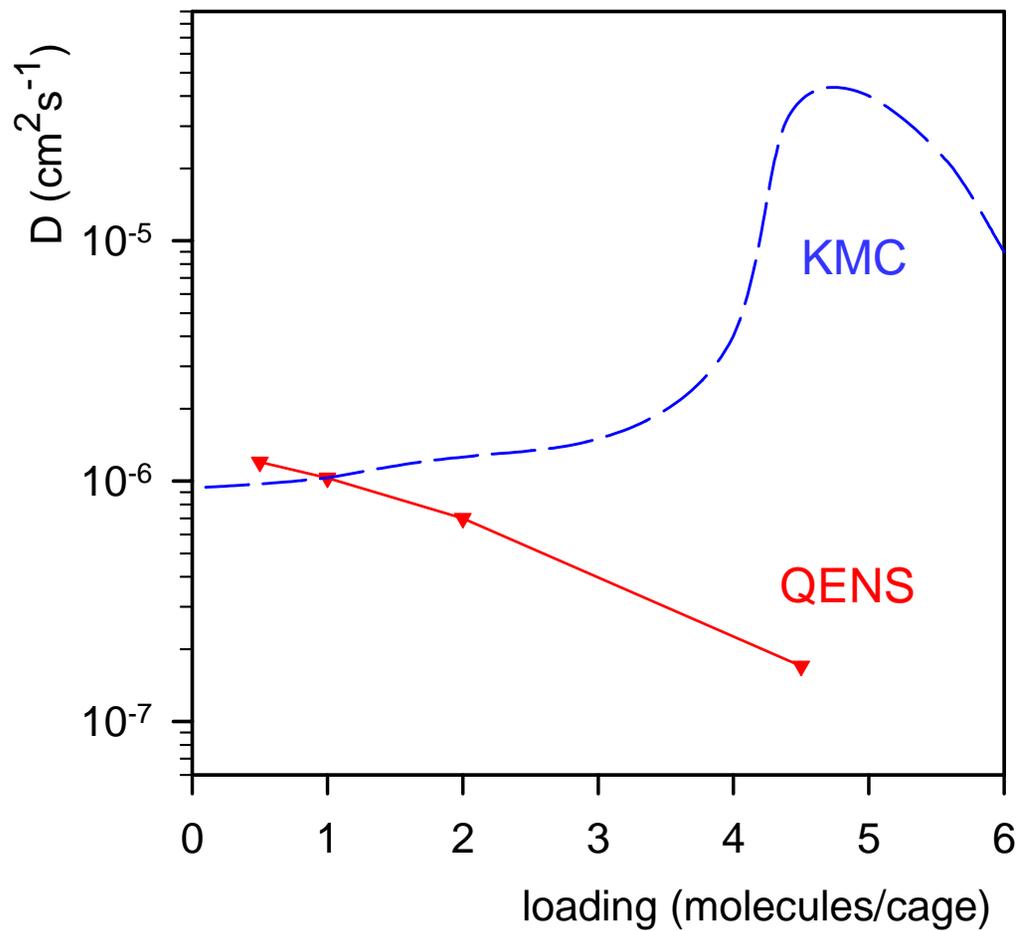
coh QENS

Ds Benzene/NaY

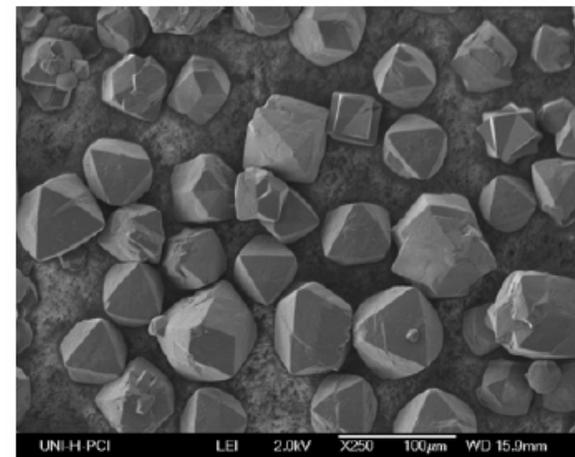
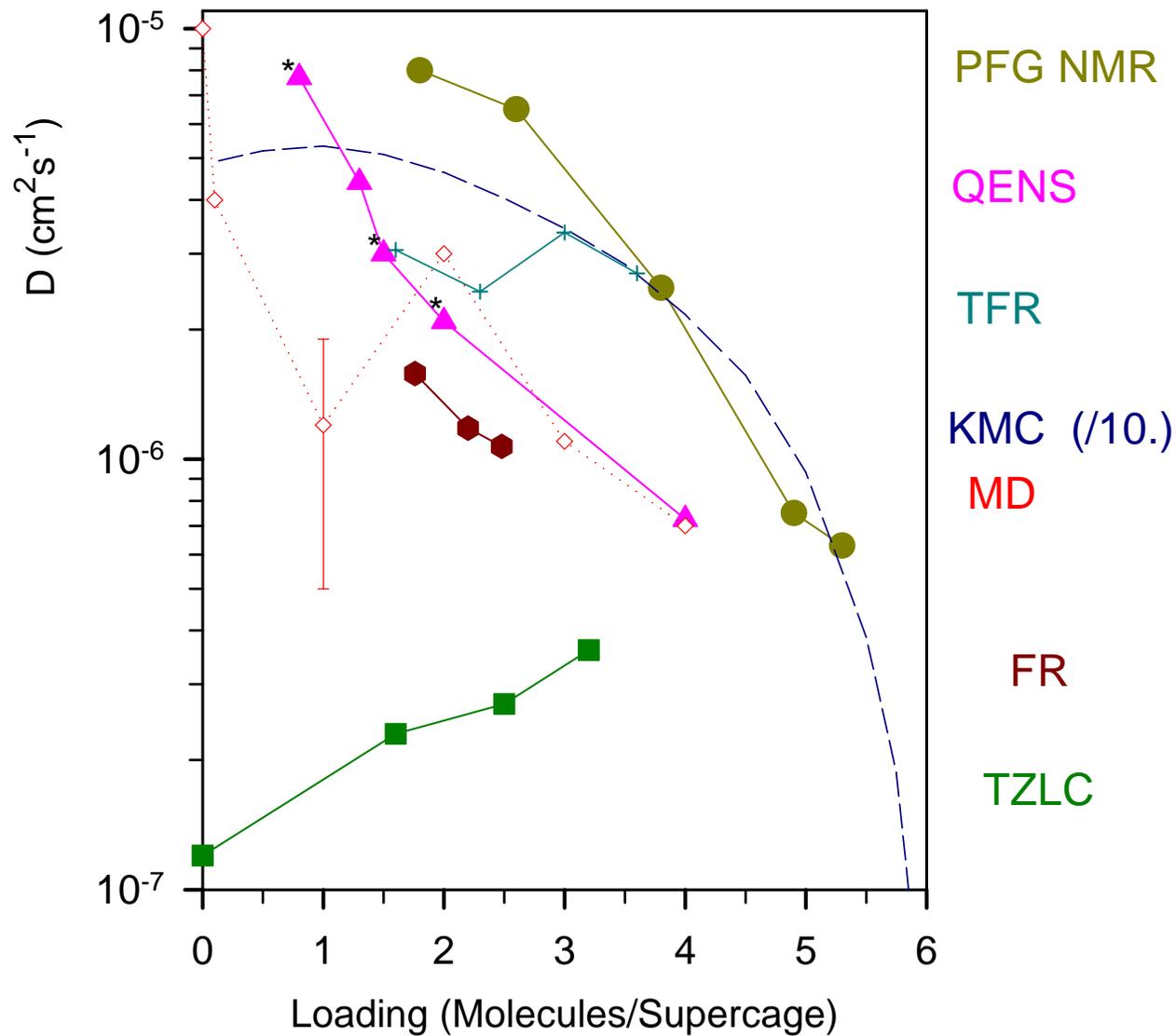


- MD
- KMC
- QENS
2 mol./cage
- SFG
3.5 mol./cage
- CRCD

Ds Benzene/NaY @ 480 K

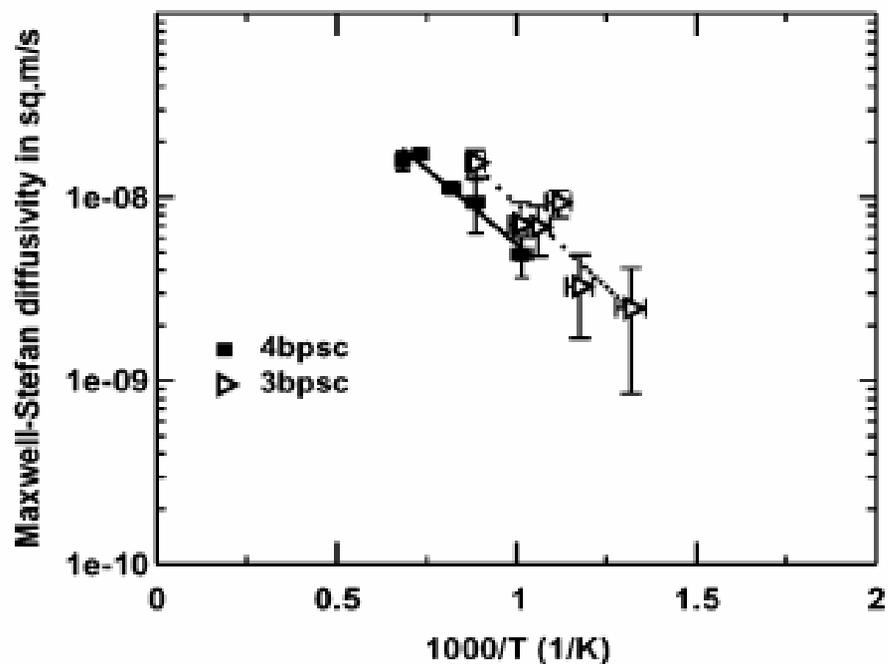


Ds Benzene / NaX (T = 468 K)

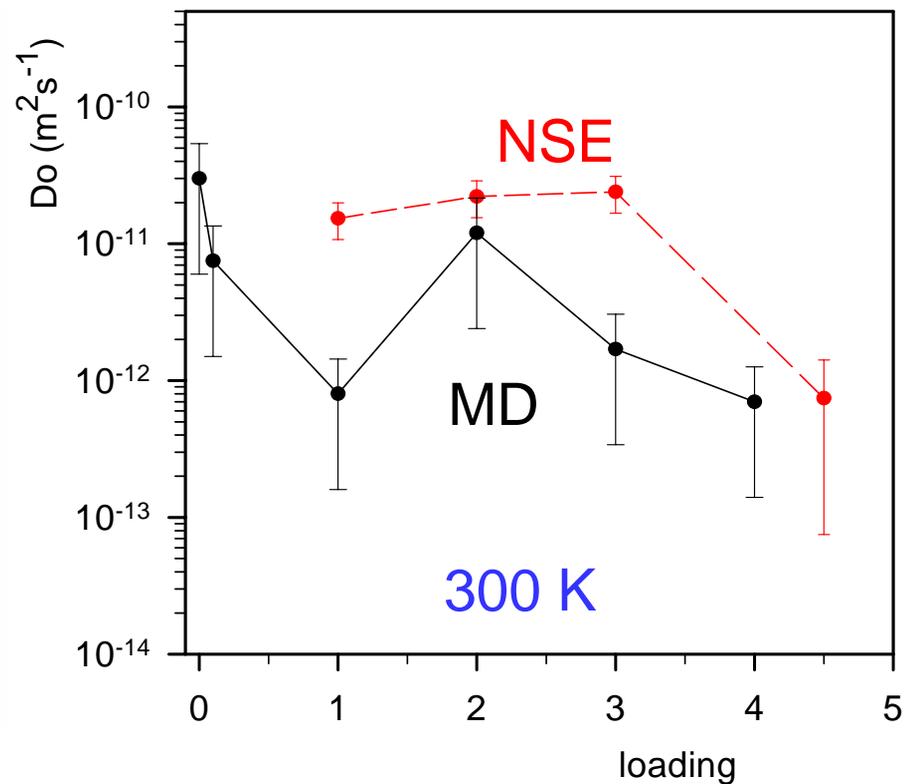


~50 μm

Do Benzene/NaX



JPC B 108 (2004) 17171



Micropor. Mesopor. Mater. 90 (2006) 307

CONCLUSIONS

Neutron scattering: structure & dynamics molecules adsorbed in zeolites.

one can work with small crystals ($< \mu\text{m}$)

Diffusion: $t : 1 \text{ ps} - 1 \mu\text{s}$ }
 $d : 0.1 - 100 \text{ \AA}$ }

Neutron diffusivities are insensitive to internal barriers and approach the simulations performed on ideal structures