

Package of Research Projects **"Diffusion in Zeolites"**

by CNRS (France), DFG (Germany), EPSRC (United Kingdom), NSF (USA)

Extension for the Period from 2006-2009

Project 5

Studying Zeolitic Diffusion by Interference and IR Microscopy

Jörg Kärger

Sergey Vasenkov (Gainesville, University of Florida)

Pavel Kortunov (Exxon-Mobil)

Christian Chmelik

DB Shah (Cleveland University)

Lars Heinke

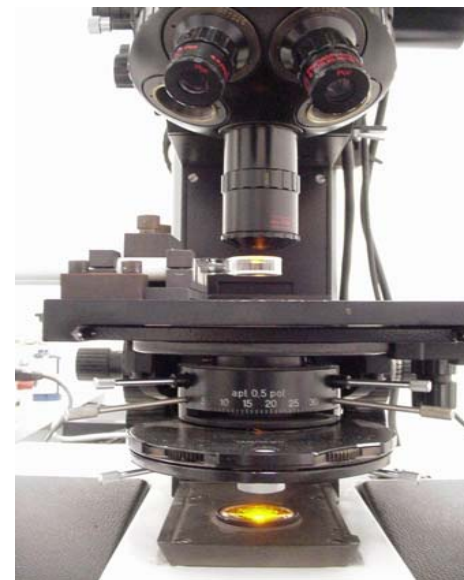
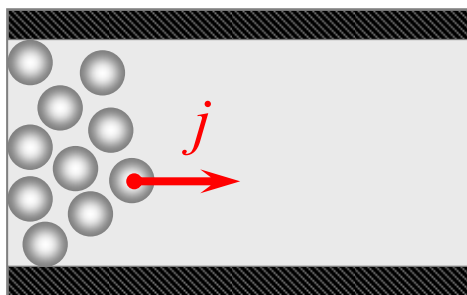
Despina Tzoulaki

What do we measure?

Interference Microscopy (IFM)

$$j = -D_{(T)} \text{grad } c$$

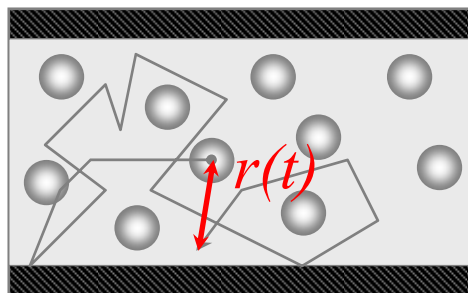
transport diffusion



Pulsed Field Gradient (PFG) NMR

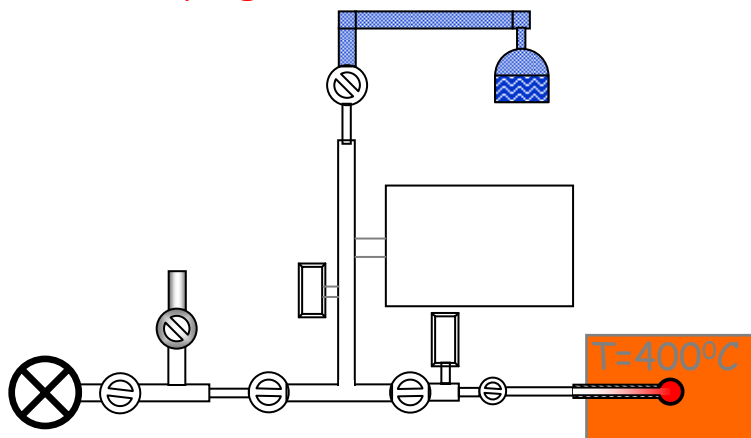
$$\langle r^2(t) \rangle = 2D t$$

self-diffusion

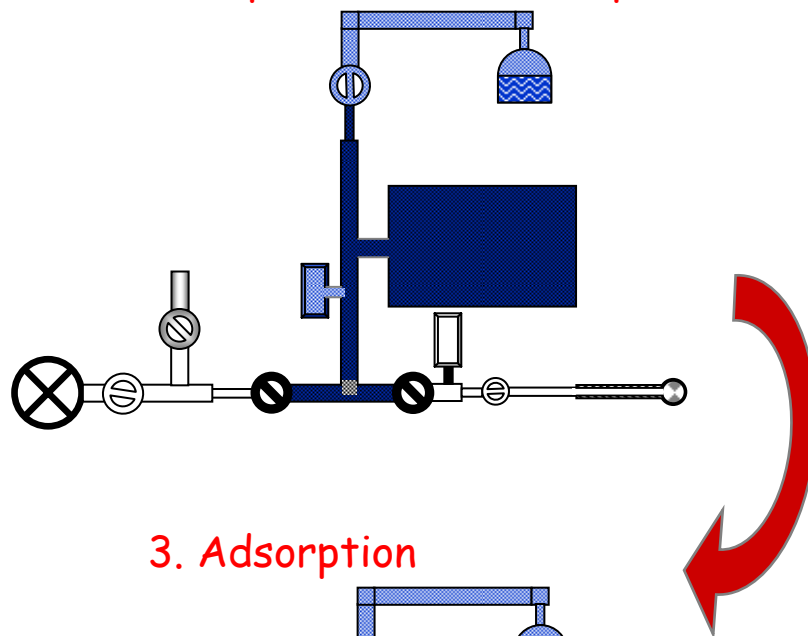


Adsorption system and cycles

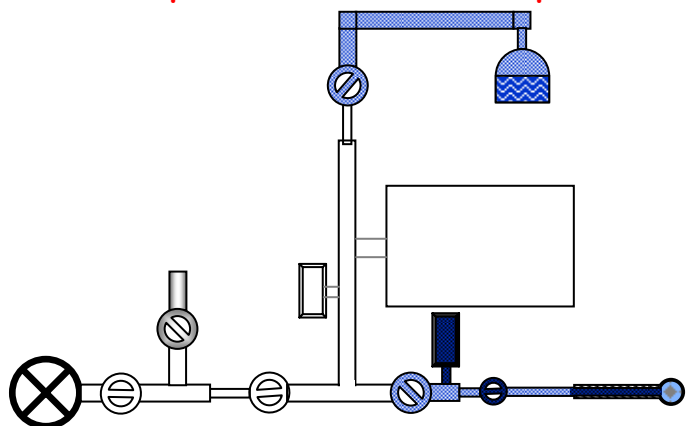
1. Pumping / Activation



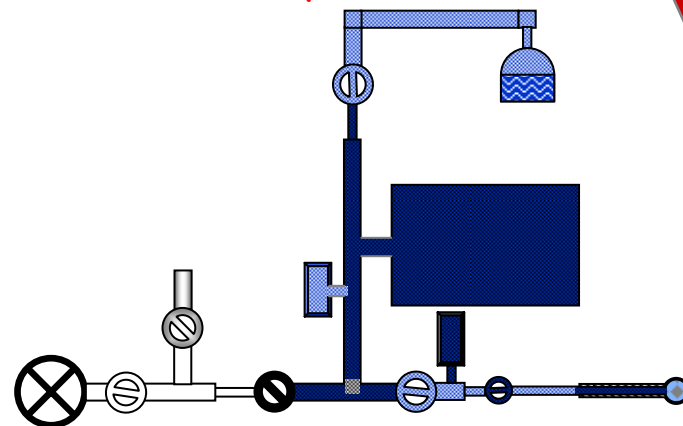
2. Preparation for adsorption



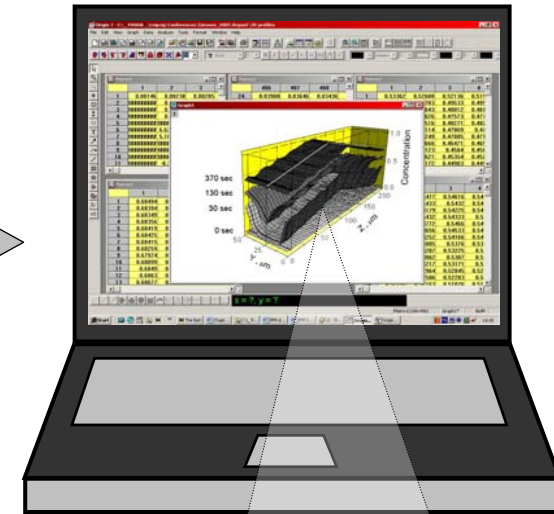
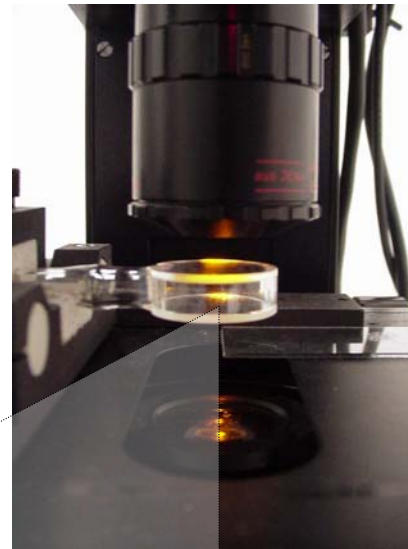
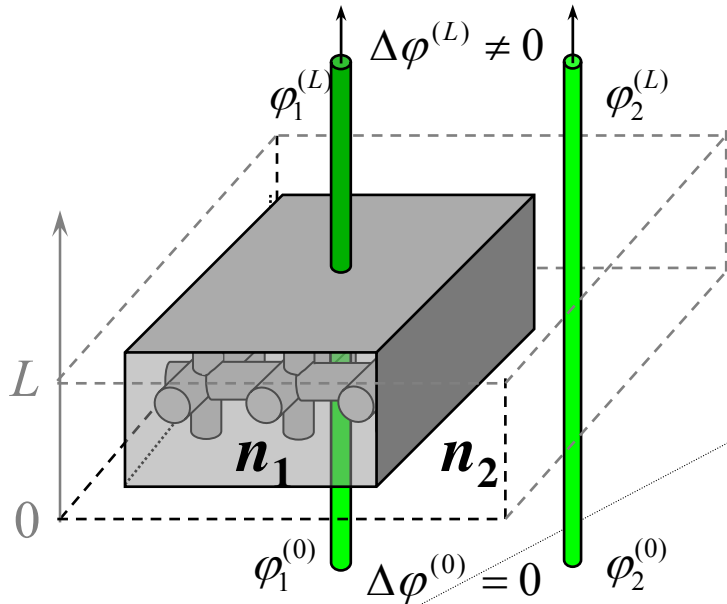
4. Preparation for desorption



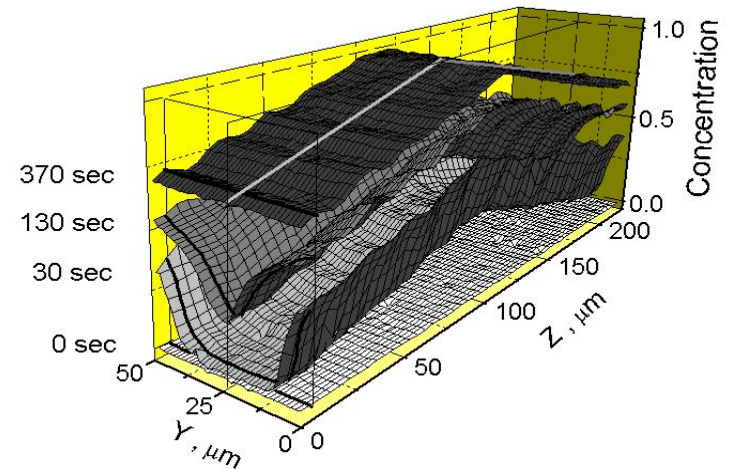
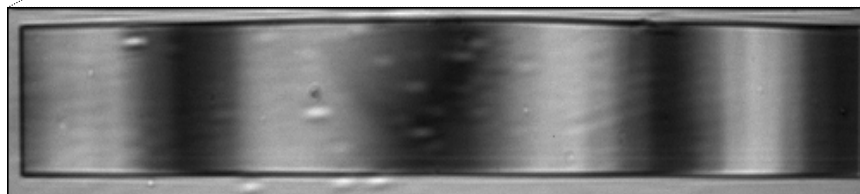
3. Adsorption



IFM Technique



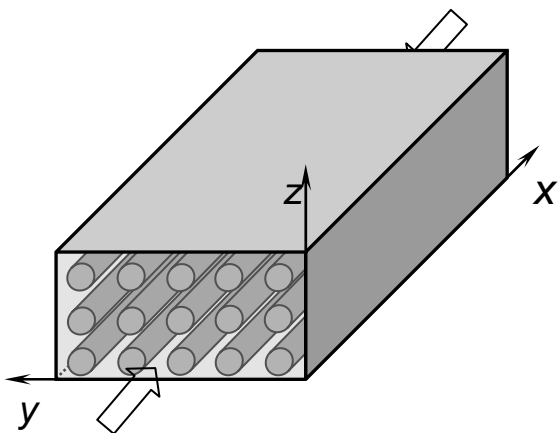
$$\Delta\varphi \sim \Delta n \sim \Delta c$$



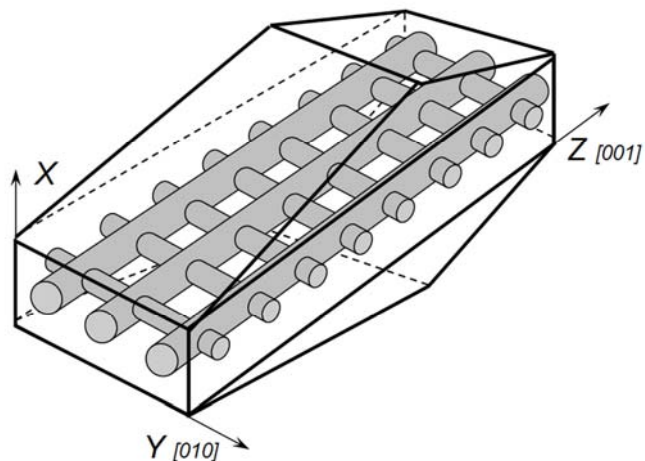
Spatial resolution: $0.5 \mu\text{m} \times 0.5 \mu\text{m}$

Outline

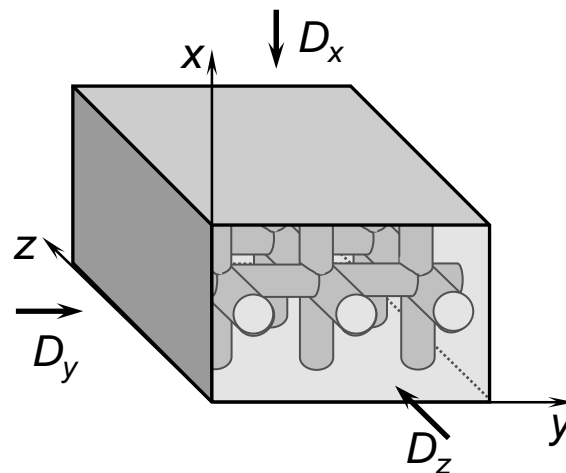
(a) 1-D pore structure of the MOF crystals



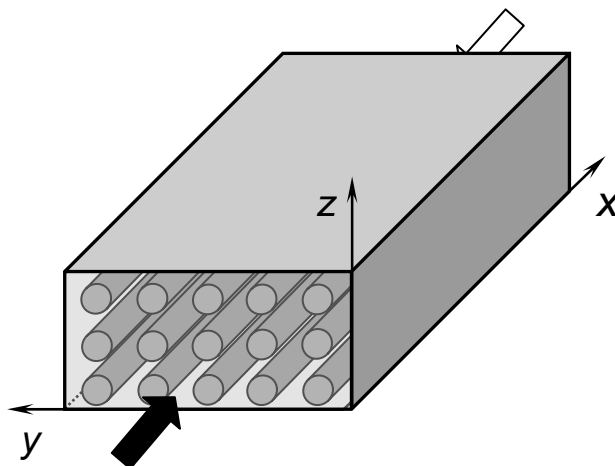
(b) 2-D pore structure of the Ferrierite zeolite



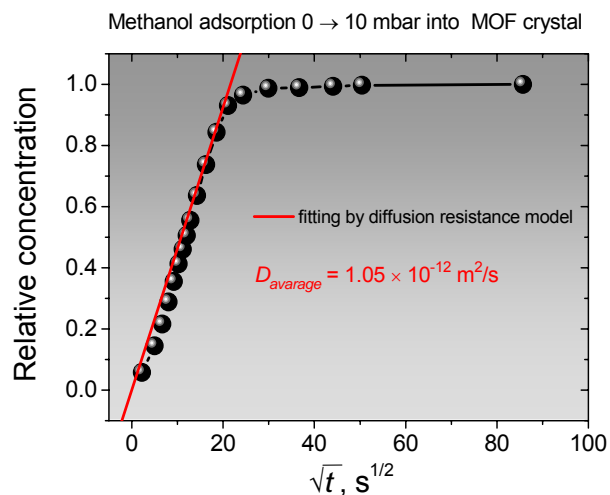
(c) 3-D pore structure of the SAPO STA-7



1-D pore structure of MOF crystal (*)

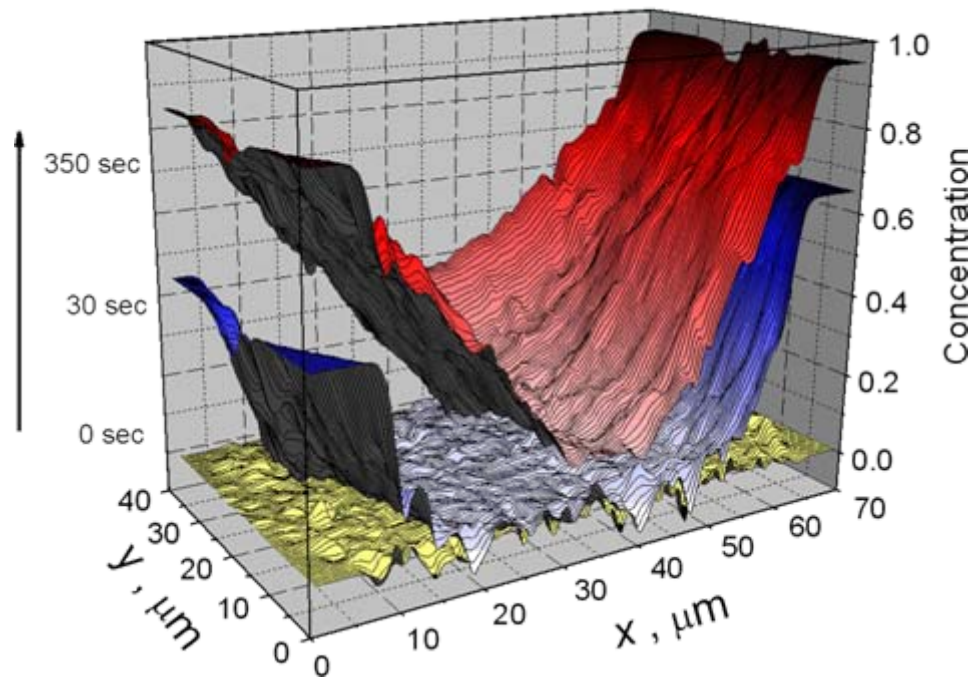


Macroscopic adsorption techniques:



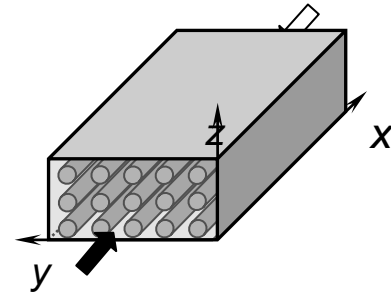
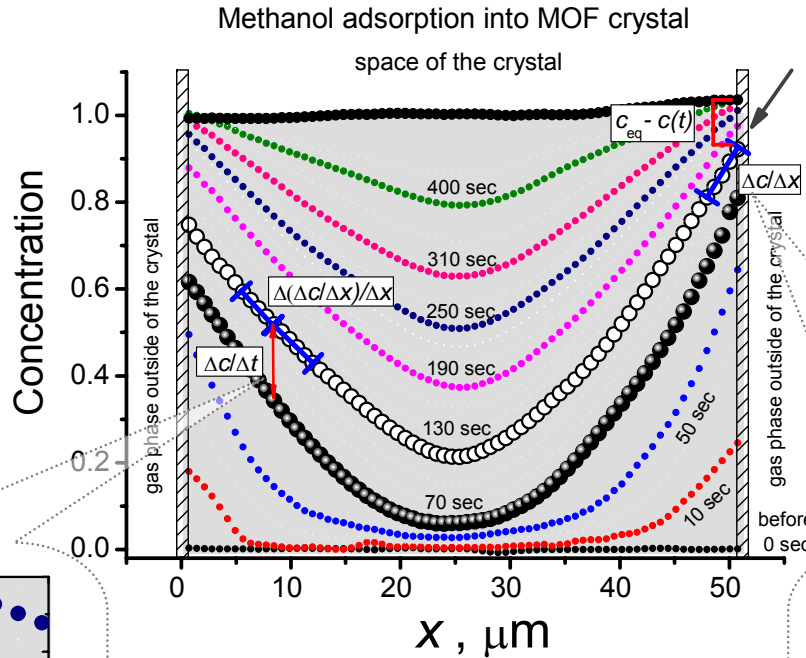
Interference microscopy technique:

adsorption of Methanol (0 → 5 mbar) into MOF crystal

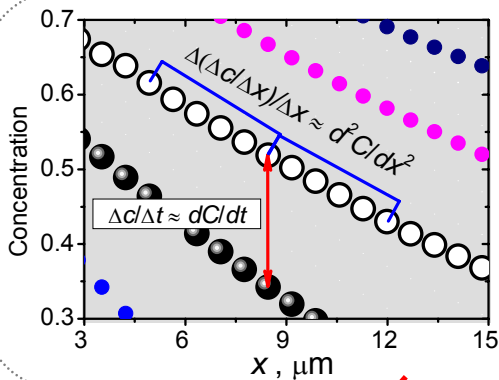


(*) D.N. Dybtsev, H. Chun, S.H. Yoon, D. Kim and K. Kim; *J. Am. Chem. Soc.*, 2004, 126, 32-33

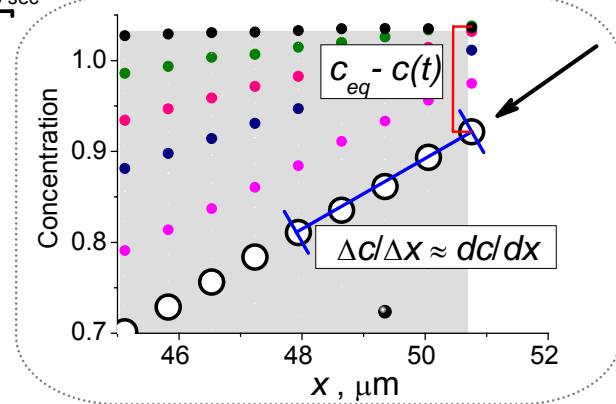
First and second Fick's laws



Local DIFFUSION



Surface PERMEABILITY



$$\frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left[D(c, x) \frac{\partial c}{\partial x} \right] = D(c, x) \frac{\partial^2 c}{\partial x^2} + \frac{\partial D(c, x)}{\partial c} \cdot \left(\frac{\partial c}{\partial x} \right)^2$$

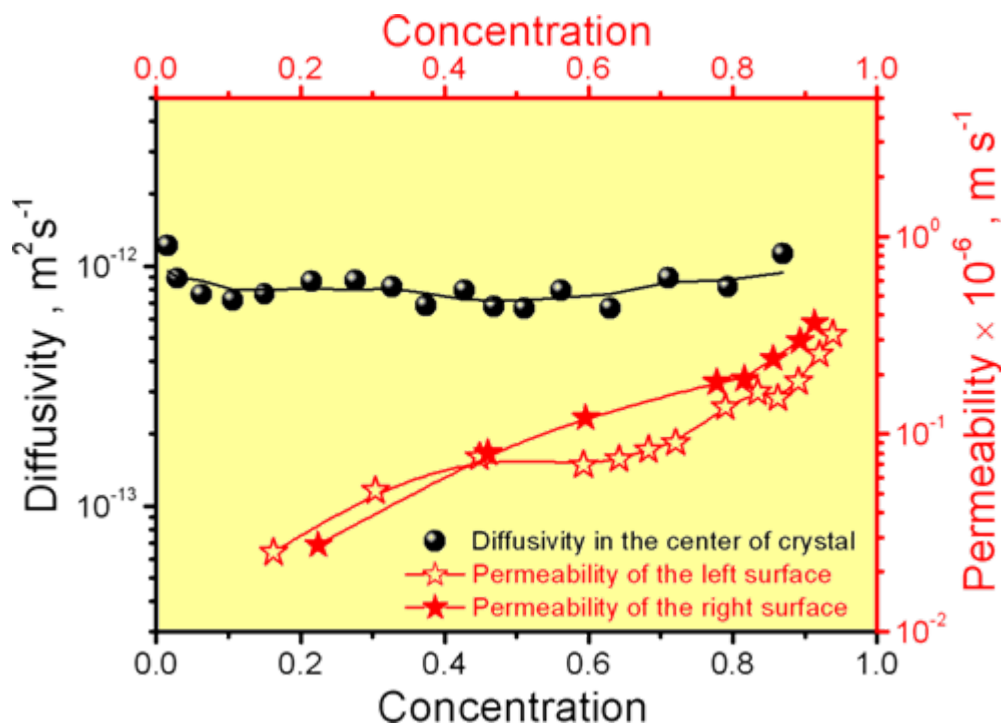
$$j = \alpha (c_{eq} - c(t)) = D(c(t)) \cdot \frac{dc}{dx}$$

1-D diffusion in the MOF crystal

Local DIFFUSION

(microscopic analysis)

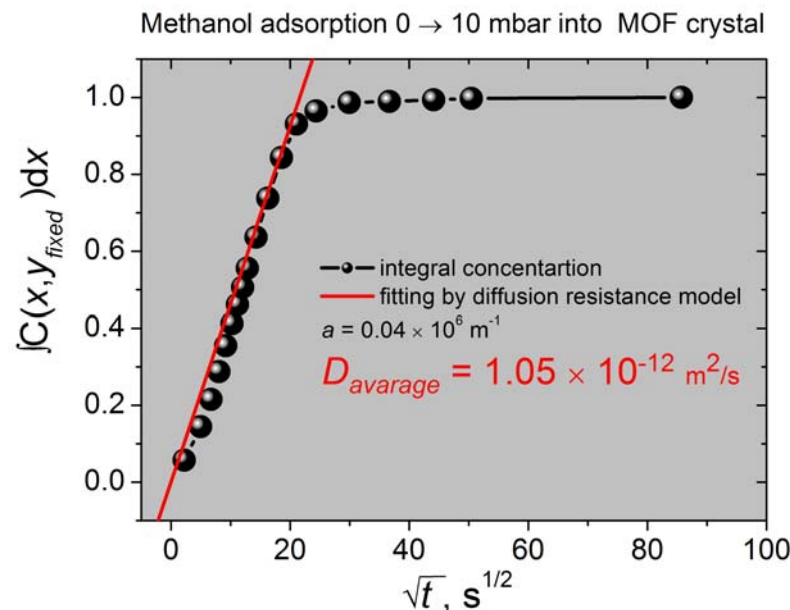
$$\frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left[D(c, x) \frac{\partial c}{\partial x} \right] = D(c, x) \frac{\partial^2 c}{\partial x^2} + \frac{\partial D(c, x)}{\partial c} \cdot \left(\frac{\partial c}{\partial x} \right)^2$$



Average DIFFUSION

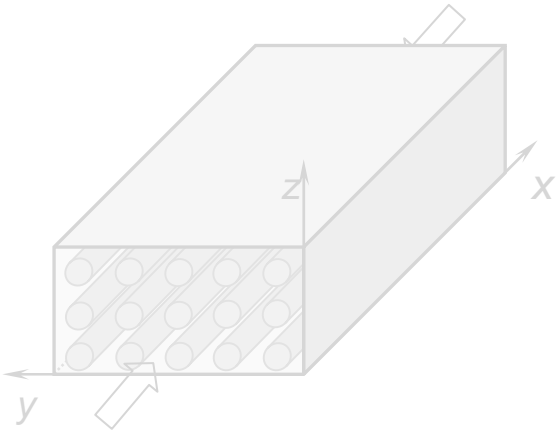
(macroscopic analysis)

$$\frac{c_t - c_0}{c_\infty - c_0} \approx 2a \sqrt{\frac{Dt}{\pi}}$$

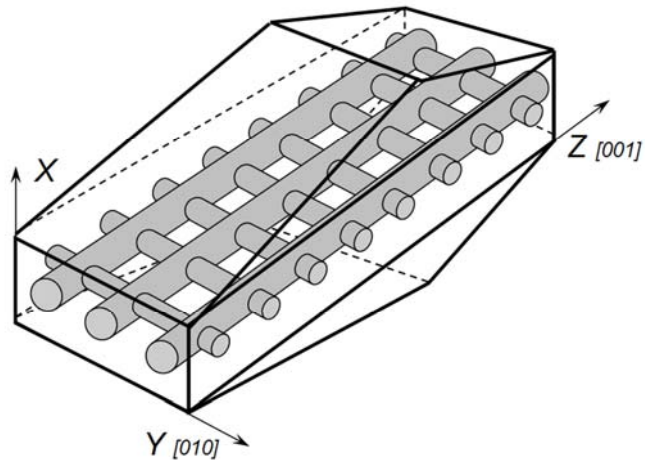


Outline

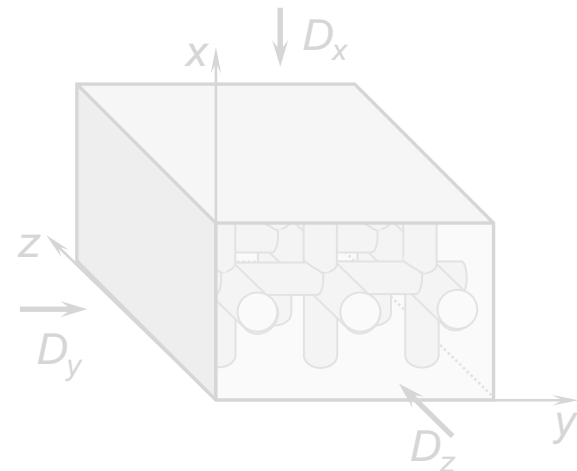
(a) 1-D pore structure of the MOF crystals



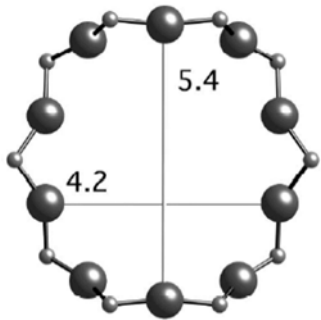
(b) 2-D pore structure of the Ferrierite zeolite



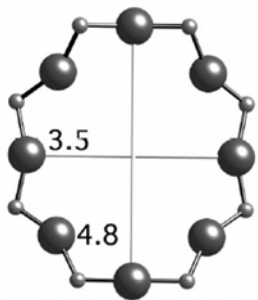
(c) 3-D pore structure of the SAPO STA-7



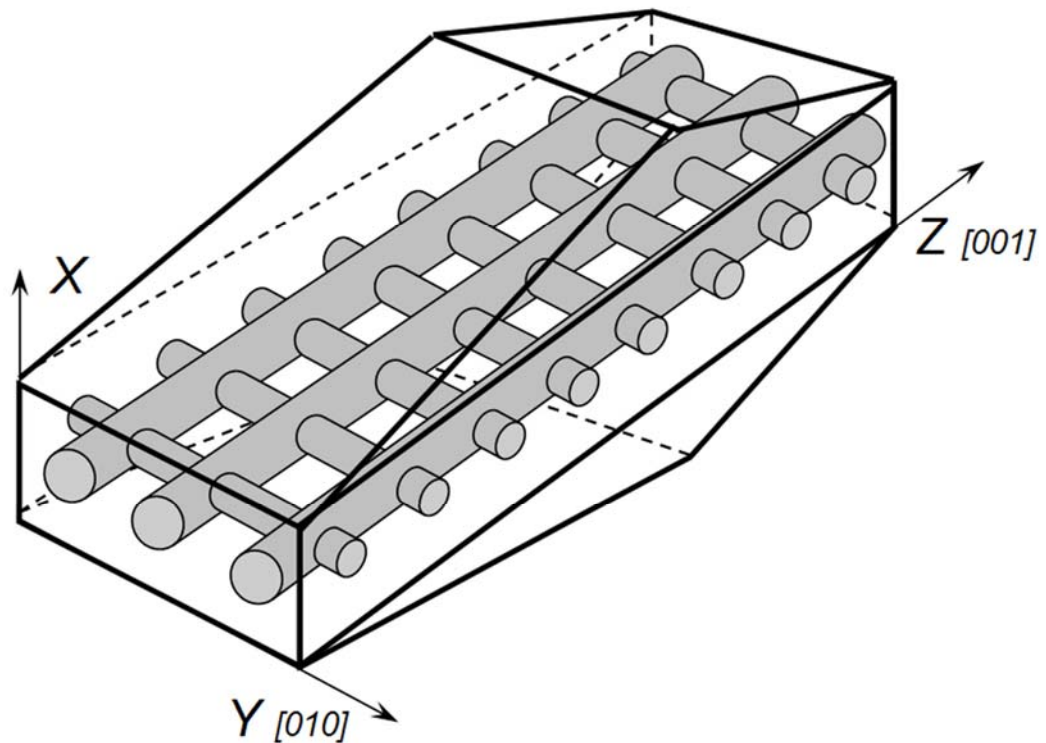
2-D pore structure of Ferrierite crystals



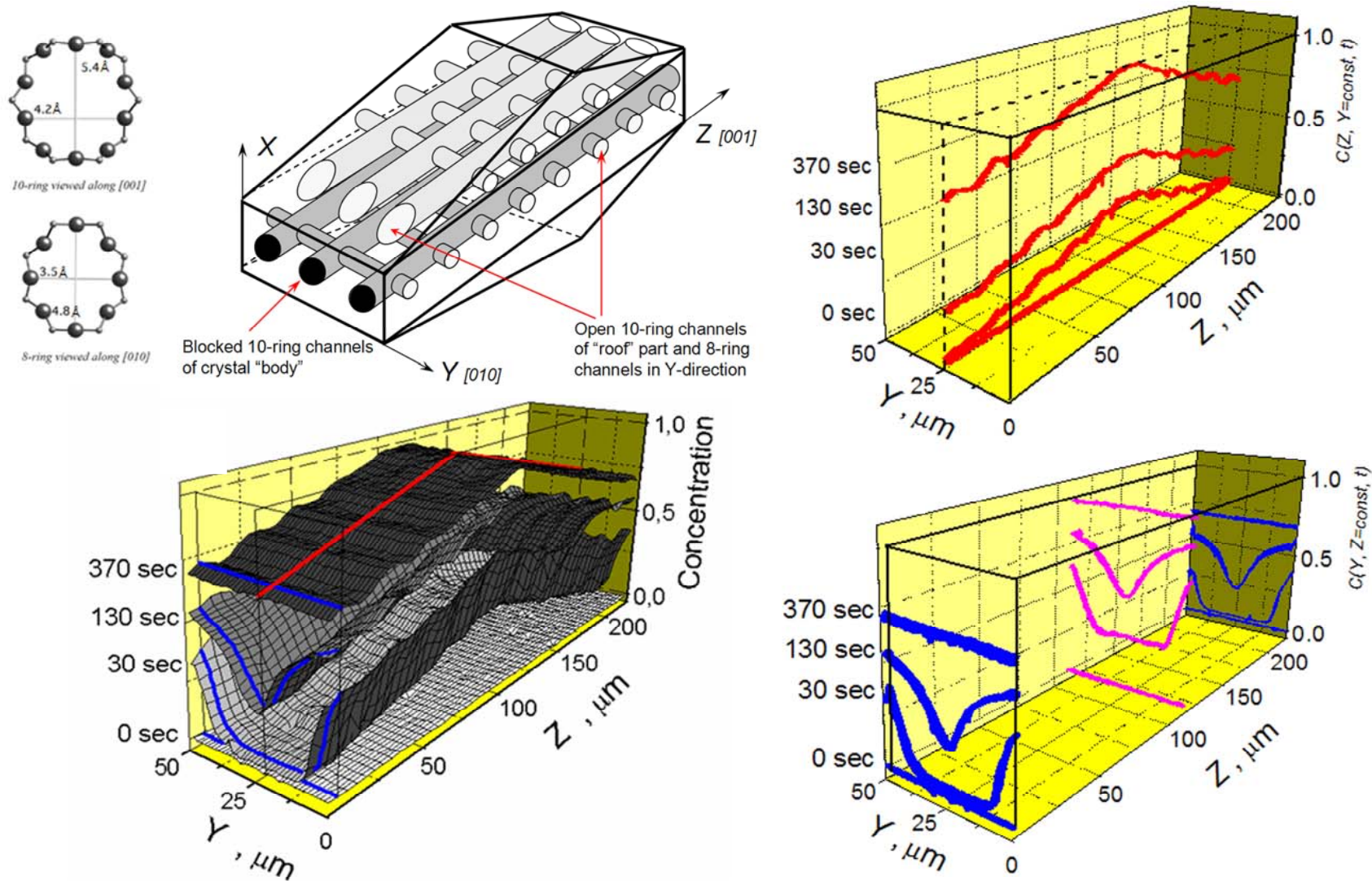
10-ring viewed along [001]



8-ring viewed along [010]

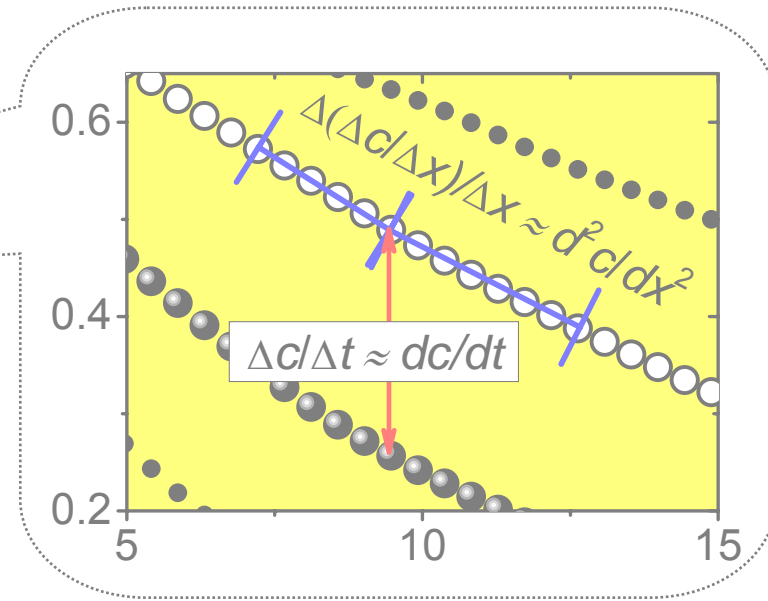
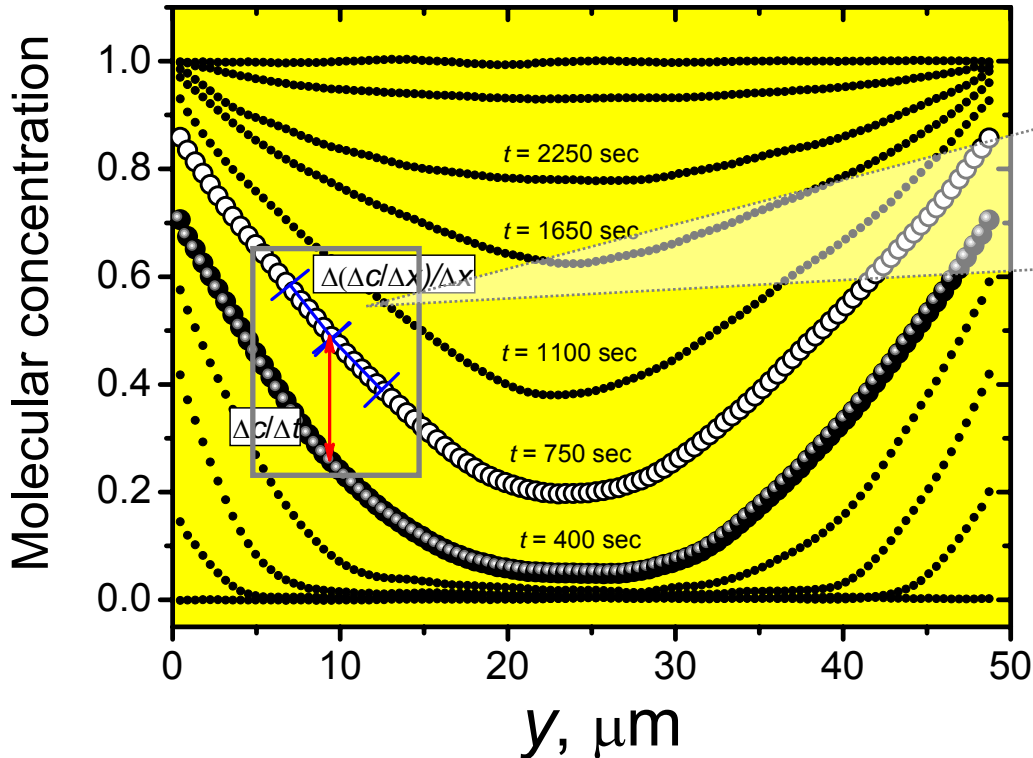


Directions of methanol adsorption in Ferrierite crystal



Local diffusion and surface resistance

Adsorption of Methanol (0 → 5 mbar) into Ferrierite



Fick's Second Law:

$$\frac{\partial c}{\partial t} = \frac{\partial}{\partial x} \left[D(c, x) \frac{\partial c}{\partial x} \right] = \cancel{D(c, x)} \frac{\partial^2 c}{\partial x^2} + \frac{\partial D(c, x)}{\partial c} \cdot \left(\frac{\partial c}{\partial x} \right)^2$$

$x = 24 \mu\text{m}$

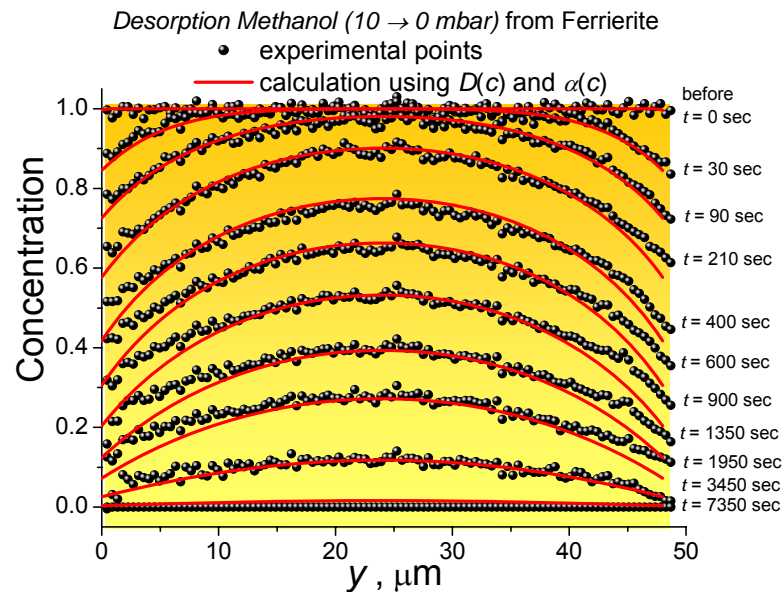
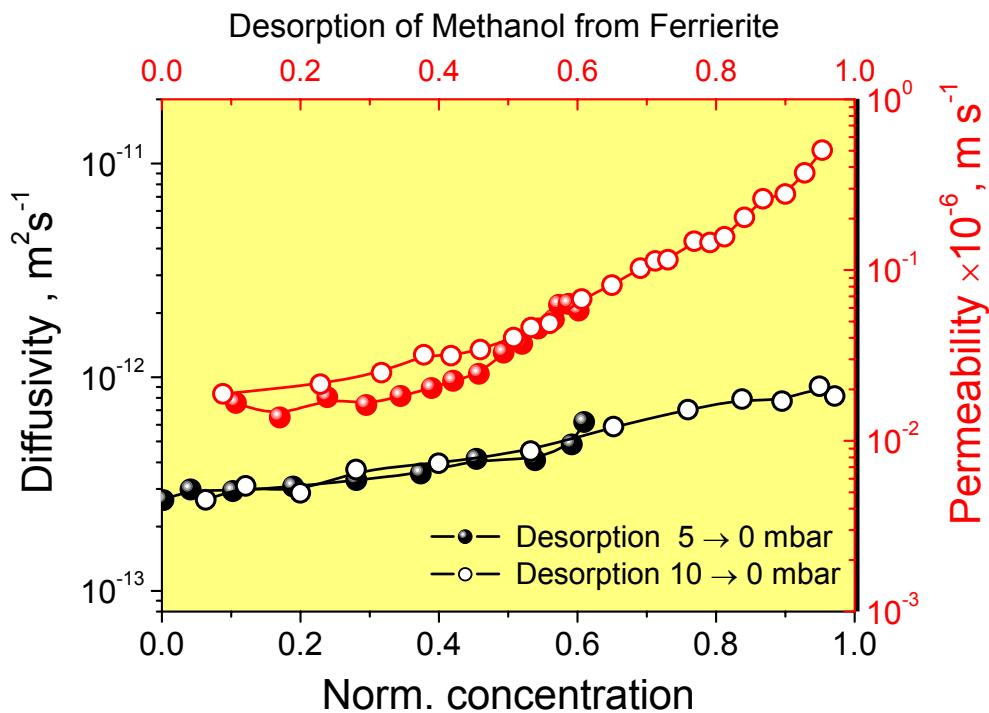
Fick's First Law:

$$j = \alpha (c_{eq} - c(t)) = D(c(t)) \cdot \frac{dc}{dx}$$

Concentration dependence of the molecular diffusivity

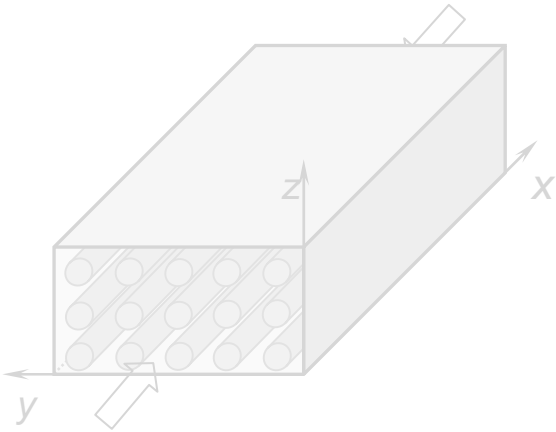
Local DIFFUSION

TEST

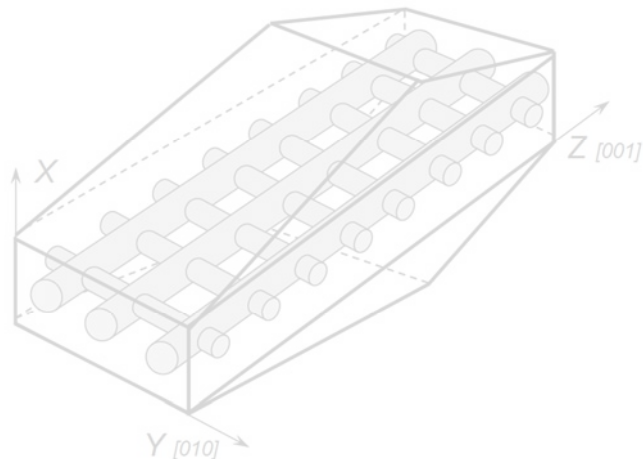


Outline

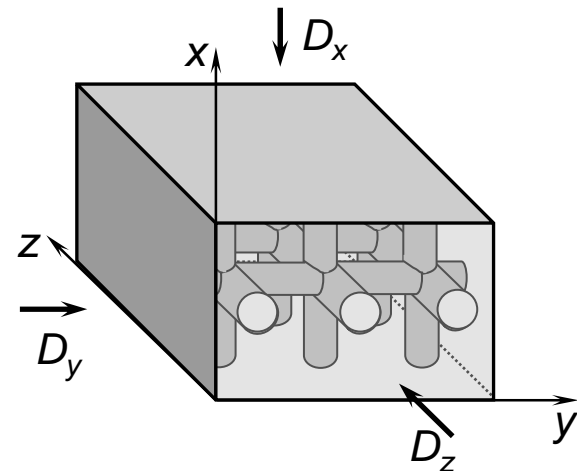
(a) 1-D pore structure of the MOF crystals



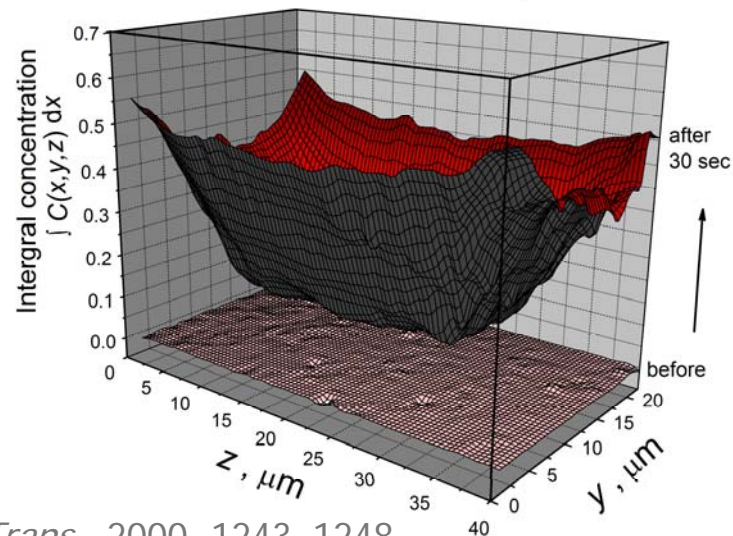
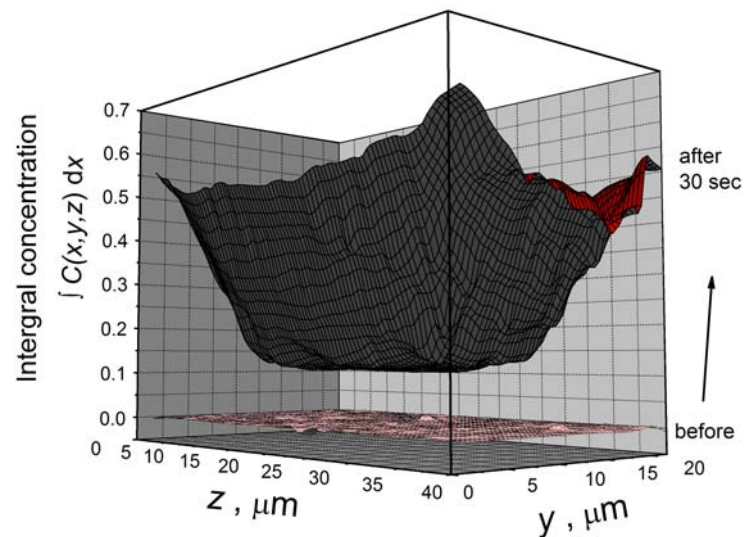
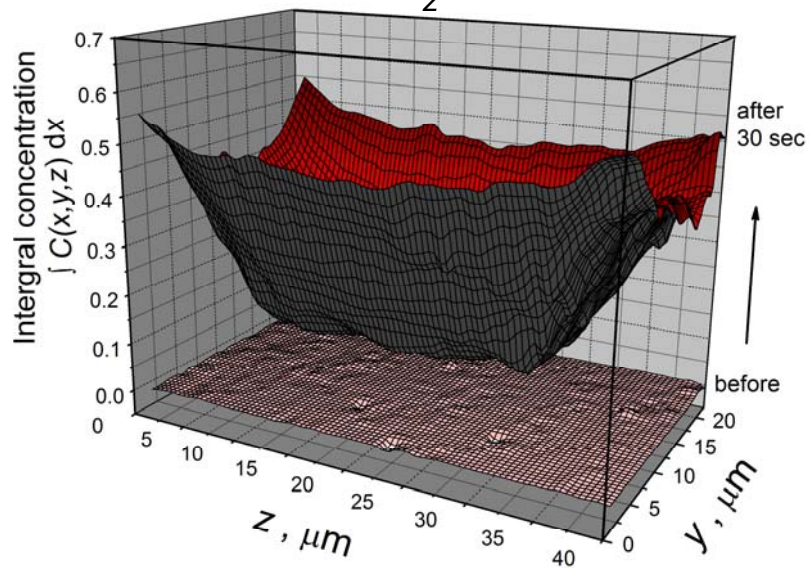
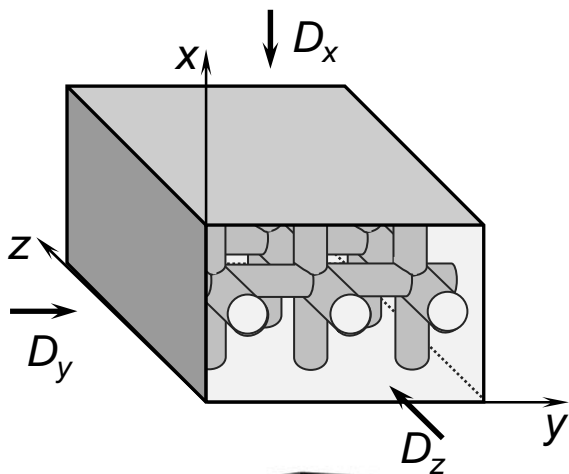
(b) 2-D pore structure of the Ferrierite zeolite



(c) 3-D pore structure of the SAPO STA-7

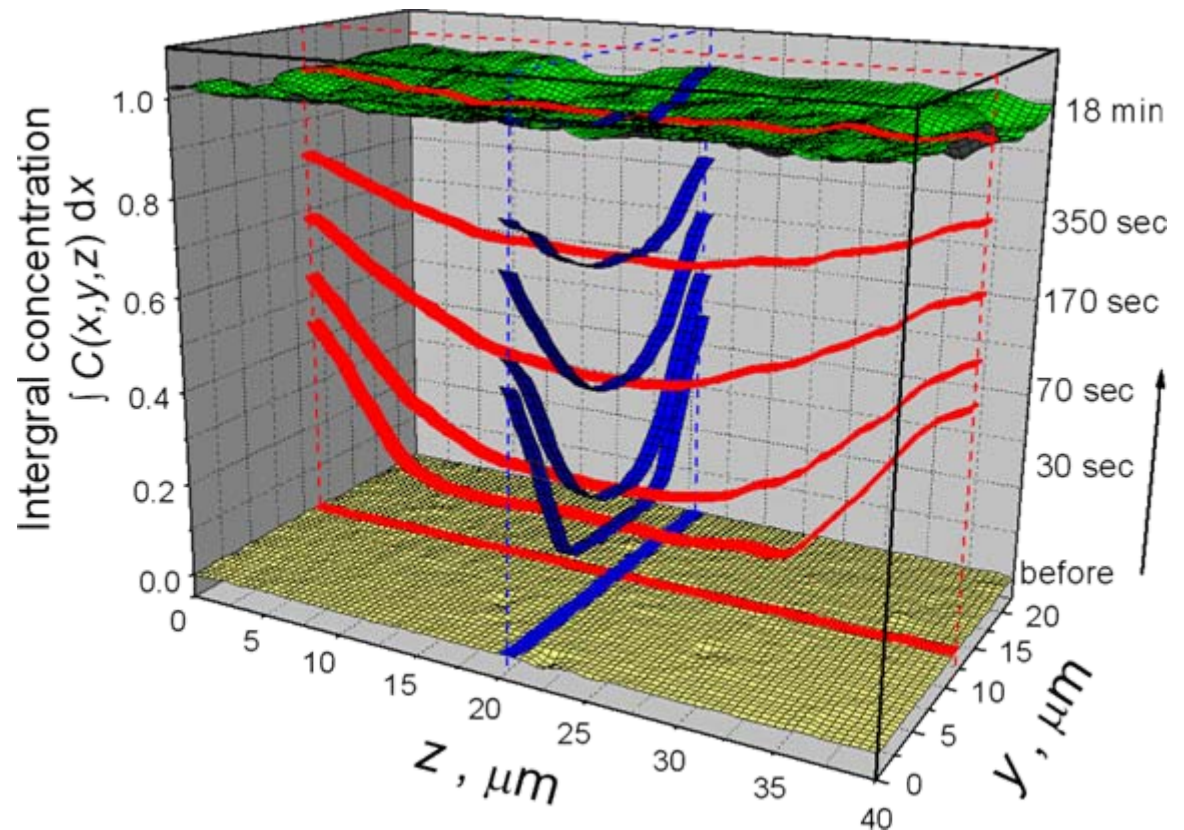
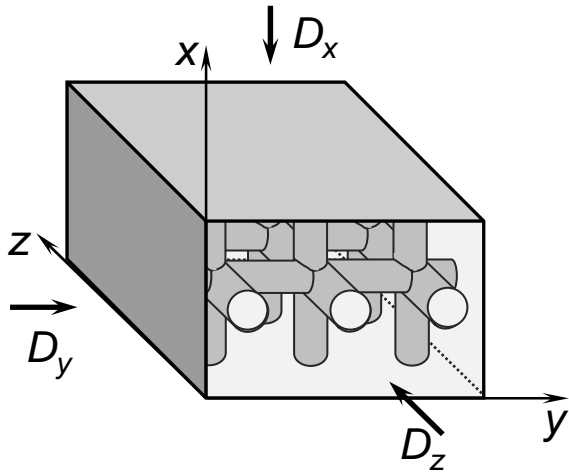


3-D pore structure of STA-7 materials (*)

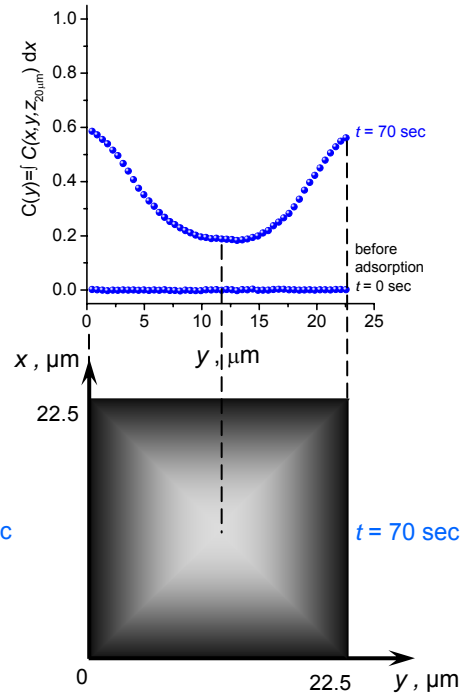
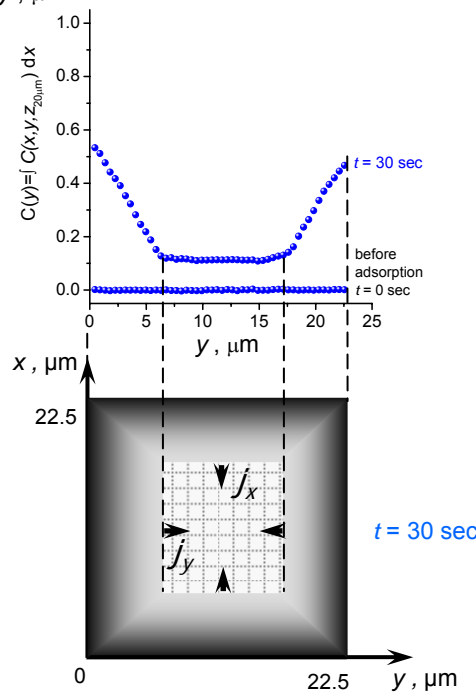
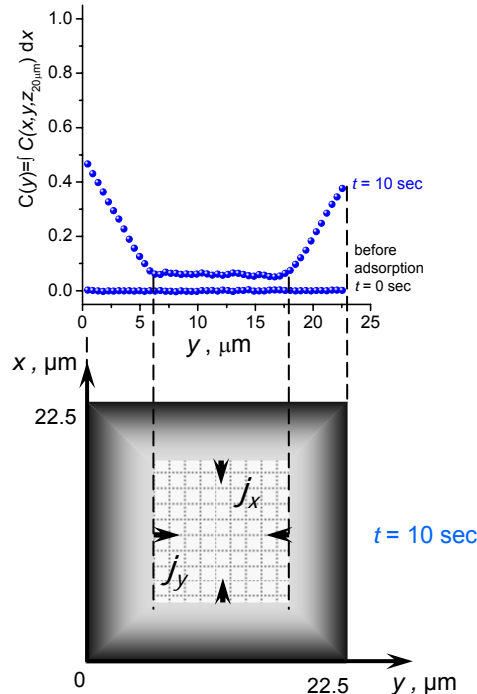
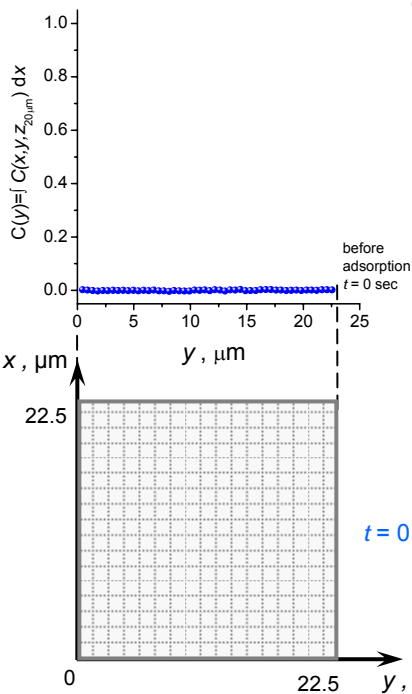
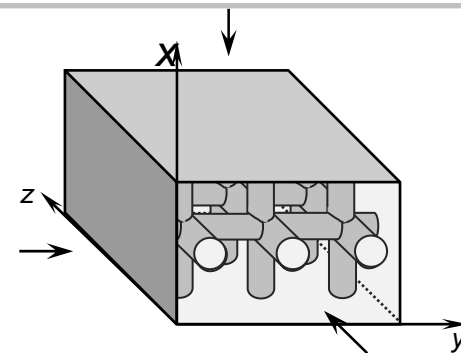
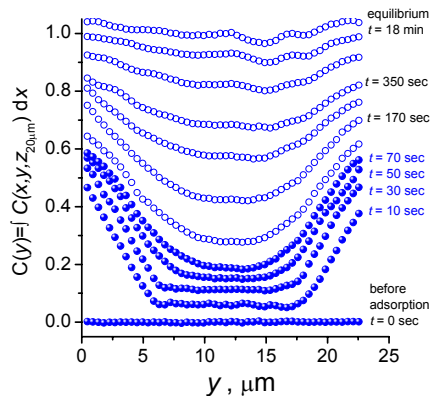
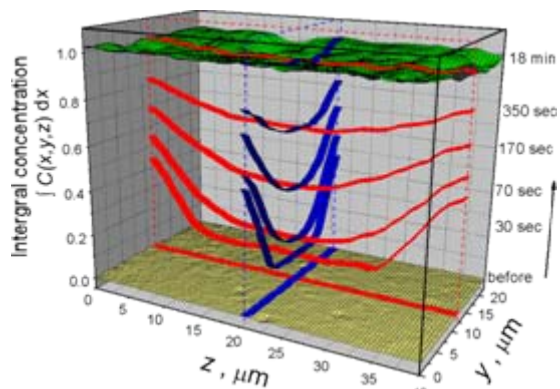


(*) Paul A. Wright, Martin J. Maple et al., *J. Chem. Soc., Dalton Trans.*, 2000, 1243–1248

3-D pore structure of STA-7 materials



Uptake in y - and x -directions



IFM perspectives...

$T_{\max} = 100^{\circ}\text{C}$ (204°C)

$T_{\min} = 0^{\circ}\text{C}$ (-50°C)

