

Synthesis of zeolite crystallites for diffusion measurements

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Main Activities in Previous Synthesis Research: Synthesis, Modification, Characterization

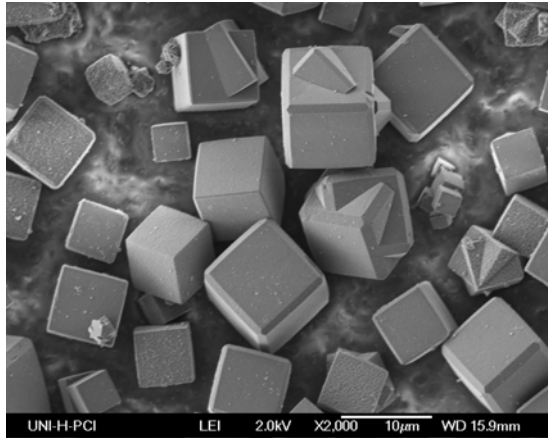
- Synthesis of zeolite A:
 - various crystal sizes and restricted size distributions: 5, 10, 20, 40 μm ;
- Synthesis of zeolite X:
 - various crystal sizes and restricted size distributions: 15, 50 and 80 μm ;
- Synthesis of molecular sieve $\text{AlPO}_4\text{-5}$:
 - large crystal size up to 500 μm , but unsatisfactory size distribution;
- Synthesis of extra-large crystals of silicalite-1: 200 μm (W. Schmidt, Mülheim);
- Synthesis of large crystals of zeolite Y: 5 μm (R. Gläser, Stuttgart);
- Synthesis of extra-large crystals of ferrierite: 200 μm (R. Gläser, Stuttgart);
- Synthesis of the MOF manganese formate of controlled morphology and as film.

- Modifications of zeolites A and X:
 - Ion-exchange: Ca^{2+} in LTA, La^{3+} in FAU;
 - Magnification of surface barriers via liquid/vapor phase chemical deposition (LCD/VCD) of TEOS;
 - Removal of surface barriers via chemical etching with HF.

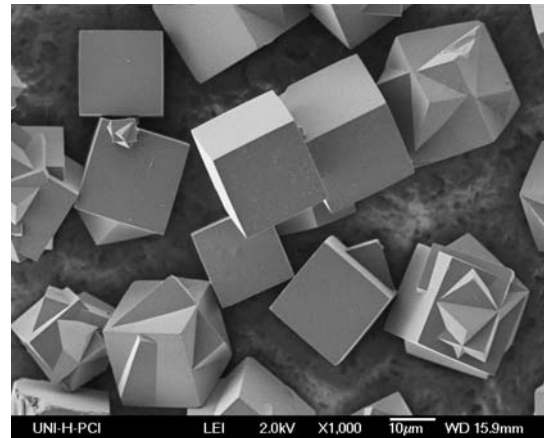
- Structural characterization:
 - SEM, TEM;
 - XRD, synchrotron XRD, neutron diffraction.

Highlights of Previous Synthesis Research: Zeolite LTA

Standardized methods established for zeolites A of various crystal sizes and restricted size distributions.



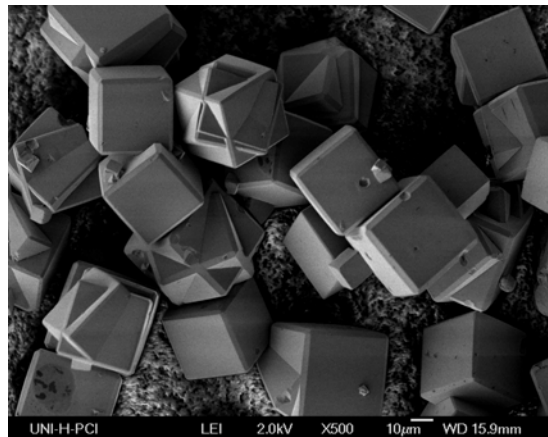
5 µm



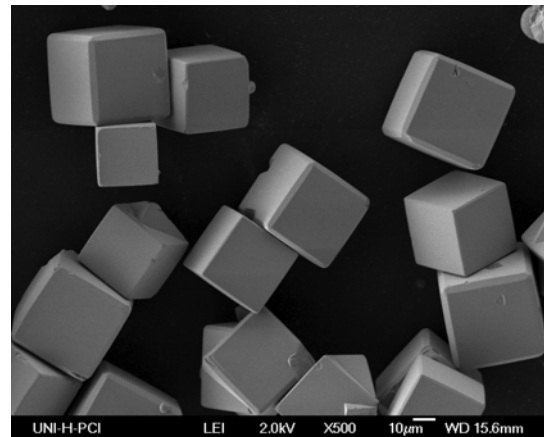
20 µm

X. Yang, D. Albrecht, J. Caro
Micropor. Mesopor. Mater.
90 (2006) 53.

A. Möller, X. Yang, J. Caro,
R. Staudt, CIT, submitted.



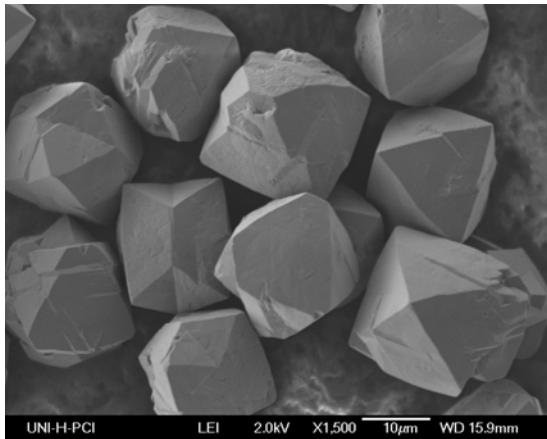
30 µm



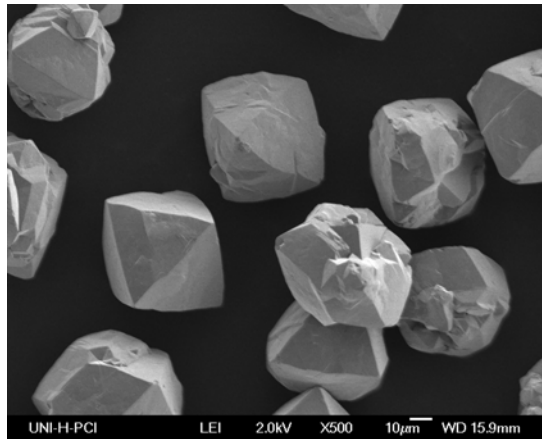
40 µm

Highlights of Previous Synthesis Research: Zeolite FAU (X)

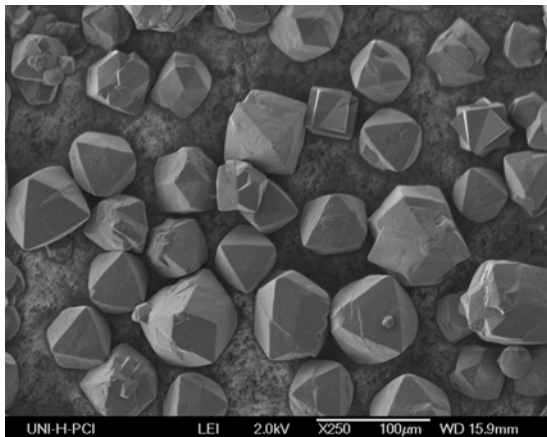
Standardized methods established for zeolites X of various crystal sizes and restricted size distributions.



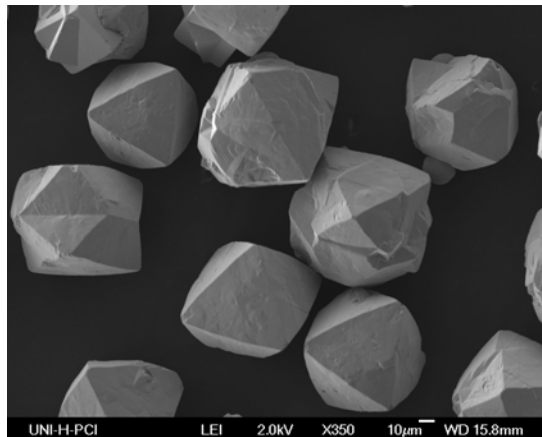
~15 µm



~ 50 µm



~50 µm

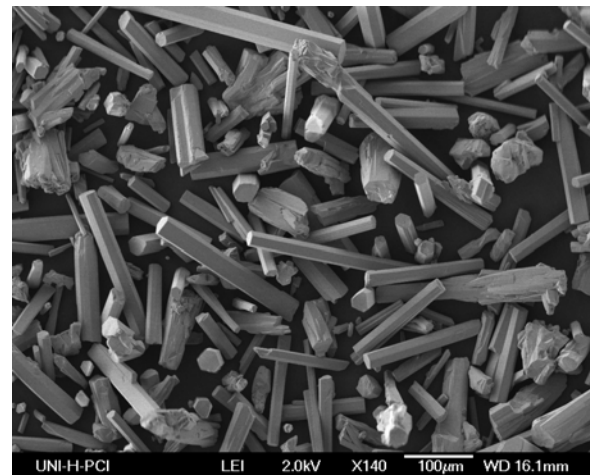
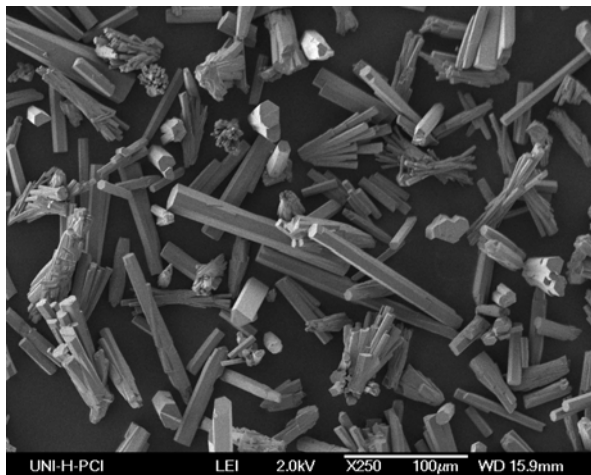
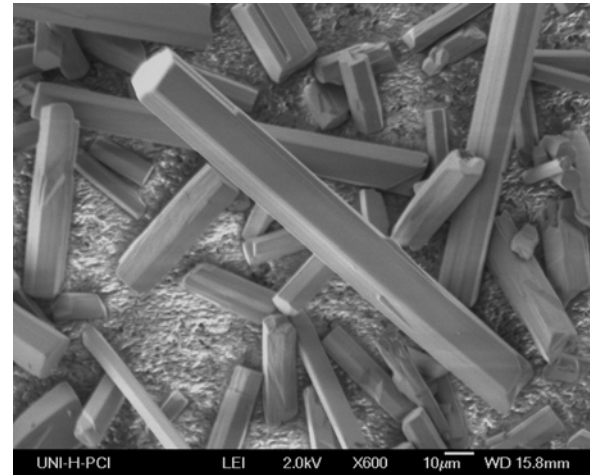
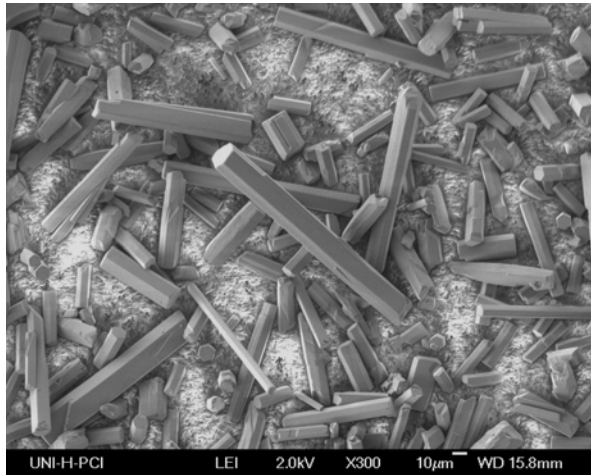


~80 µm

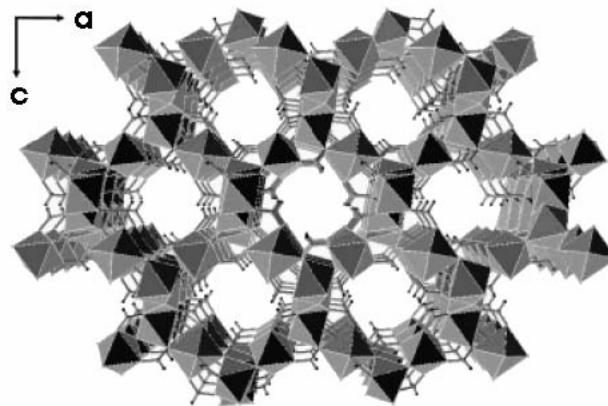
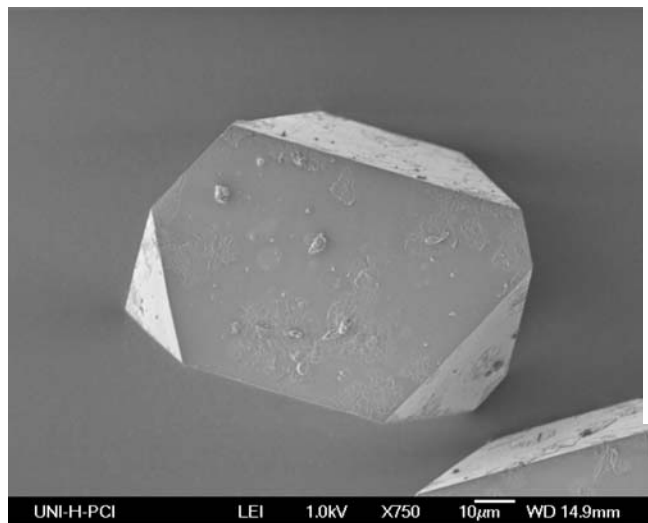
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Highlights of Previous Synthesis Research: Aluminophosphate $\text{AlPO}_4\text{-5}$

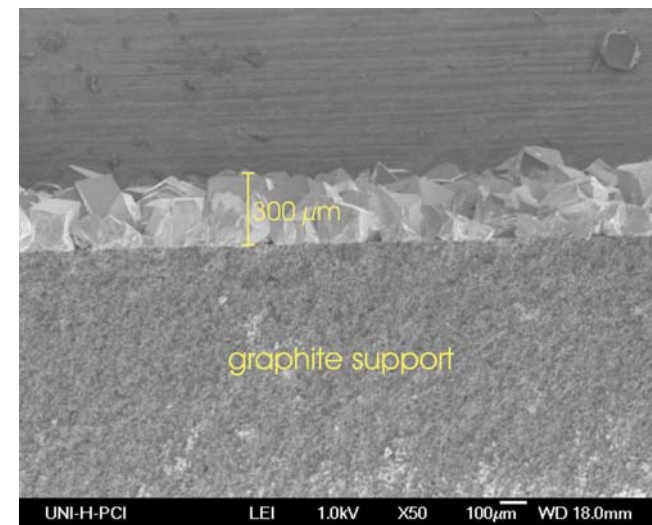
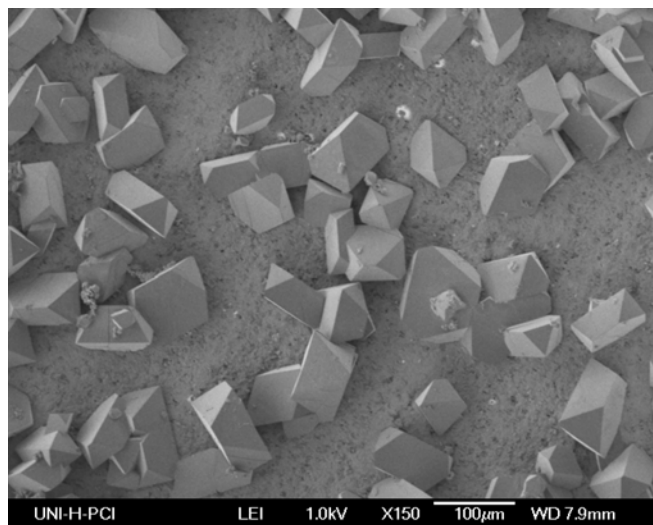
$\text{AlPO}_4\text{-5}$: synthesis of crystals with shape of hexagonal prism of lengths ca. 500 μm has been achieved; Problems with broad distributions of crystal sizes and crystal intergrowth have to be solved.



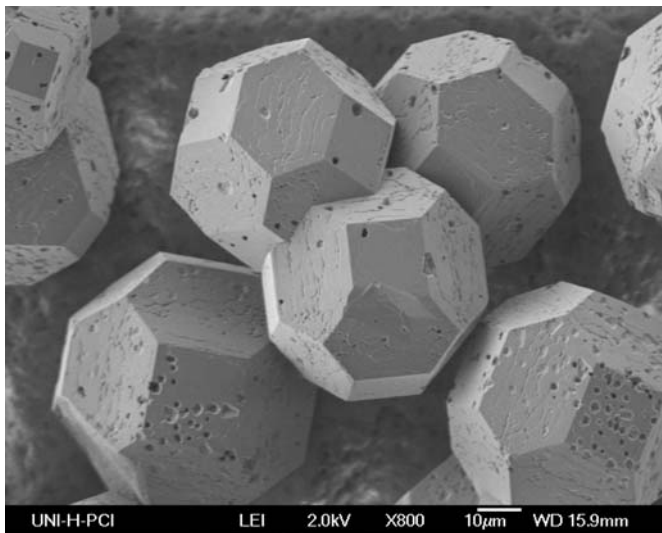
Highlights of Previous Synthesis Research: MOF Manganese Formate: Crystals and Film



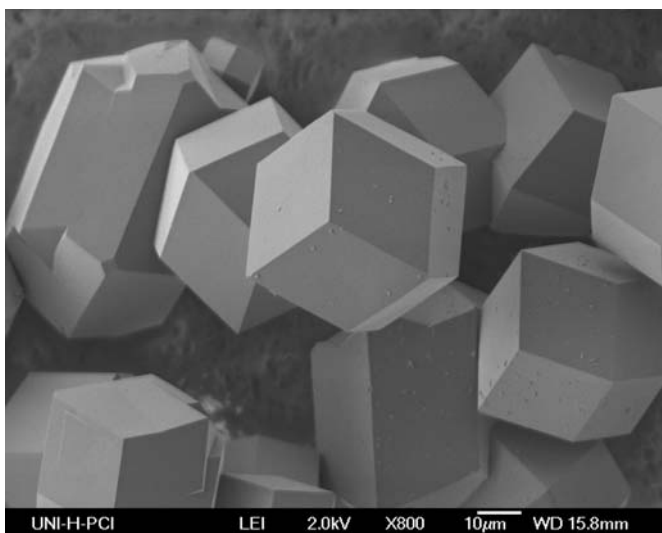
M. Arnold, P. Kortunov,
D.J. Jones, Y. Nedellec,
J. Kärger, J. Caro,
Europ. J. Inorg. Chem.,
in press.



Highlights of Previous Synthesis Research: Germanosilicate Sodalite

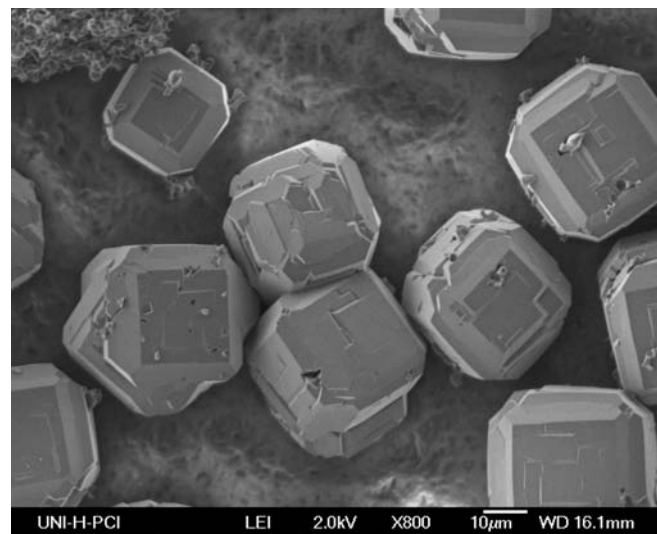


Synthesized in anhydrous EG solution, Si/Ge~5



Synthesized in anhydrous EG solution, Si/Ge~3

X. Yang, D. Albrecht, J. Caro,
Micropor. Mesopor. Mater.,
accepted.



Synthesized in EG solution with fluoride ions
and small amount of water, Si/Ge~3

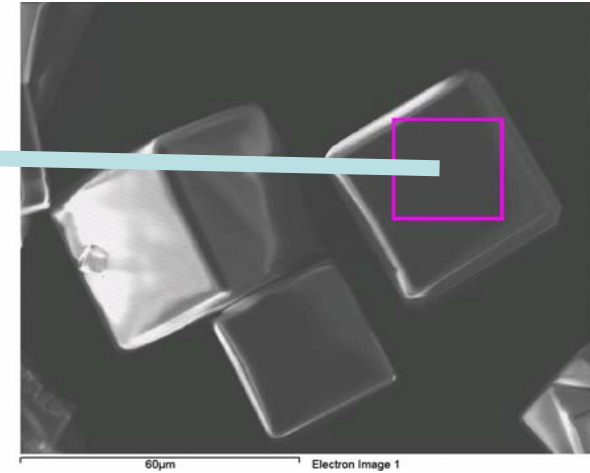
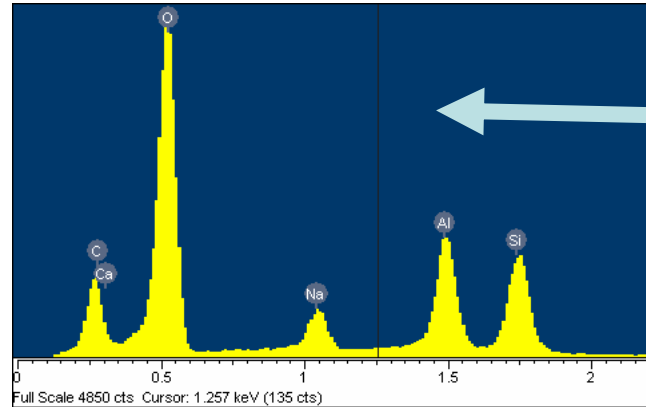
Highlights of Synthesis Research: Help from Partners silicalite-1, zeolite Y, ferrierite

Great help from partners: Extra-large crystals of

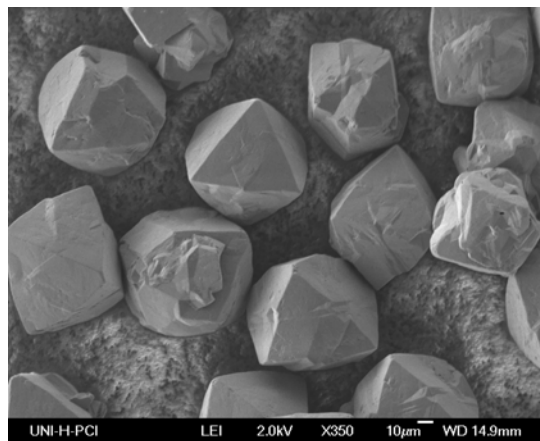
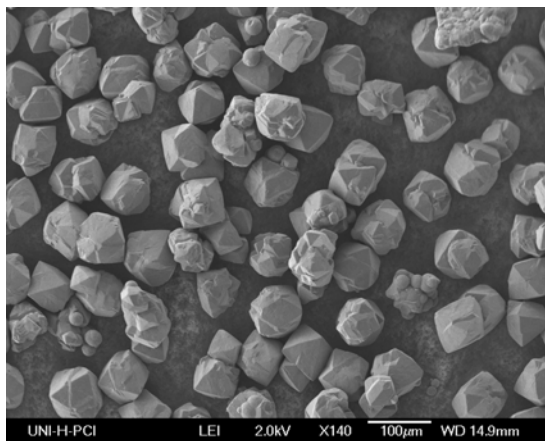
- silicalite-1 (200 μm in length) : W. Schmidt, Mülheim
- zeolite Y (5 μm diameter): R. Gläser, Stuttgart
- ferrierite (200 x 50 x 10 μm^3): R. Gläser, Stuttgart

Highlights of Synthesis Research: Post-synthesis Modifications

Ca²⁺-exchange
of zeolite NaA:
Quantitative
determination
of Ca-content
by SEM/EDX



CaNaA, edge-length ~ 40 μm

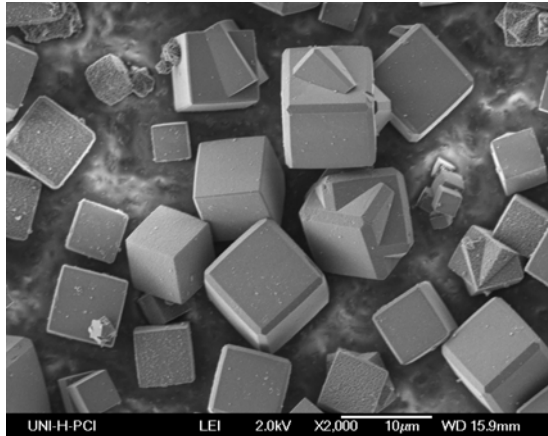


LaNaX, edge-length ~ 70 μm

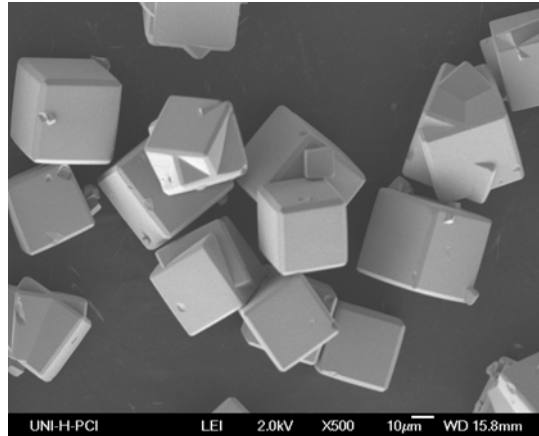
After 3 times
La³⁺-exchange
of zeolite NaX:
SEM shows
crystals are
un-damaged

Highlights of Synthesis Research: Strengthen surface barriers for diffusions through LCD/VCD of TEOS

Before TEOS coating

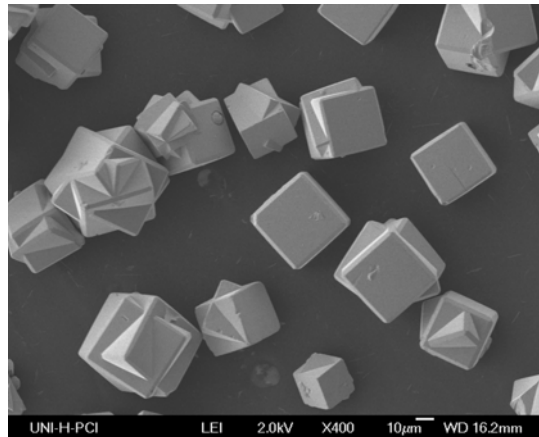
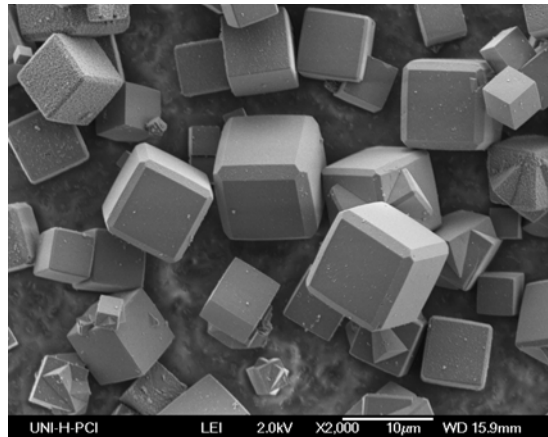


5 μm



35 μm

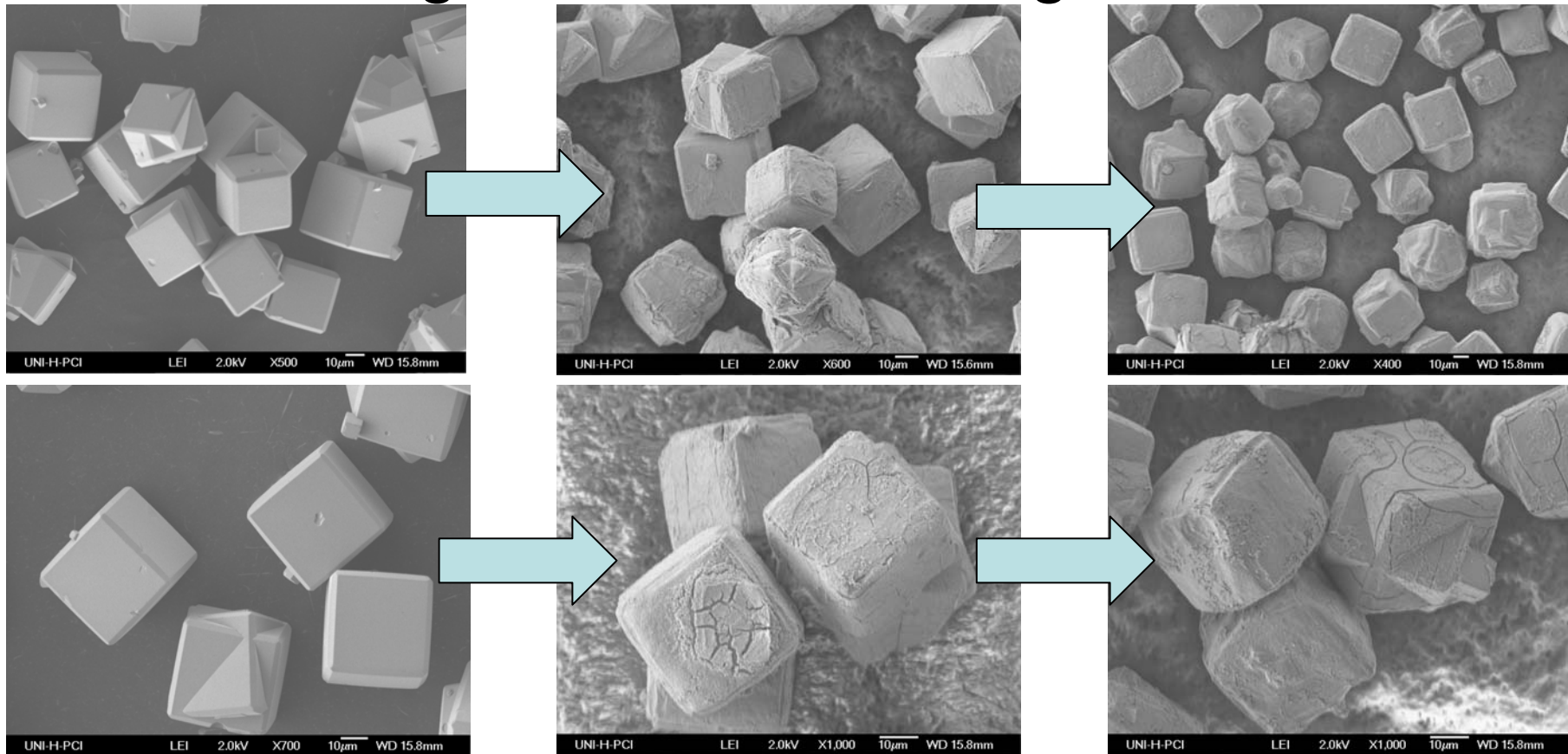
After TEOS coating



No foreign particles
are formed (silica)

Homogeneous silica
coating

Highlights of Previous Synthesis Research: Removal of surface barriers for diffusions through chemical etching with HF



Starting: XY293_1 CaNaA
1 g, calcined at 500° C.
Exposed to n-C₆ at RT for 1 week

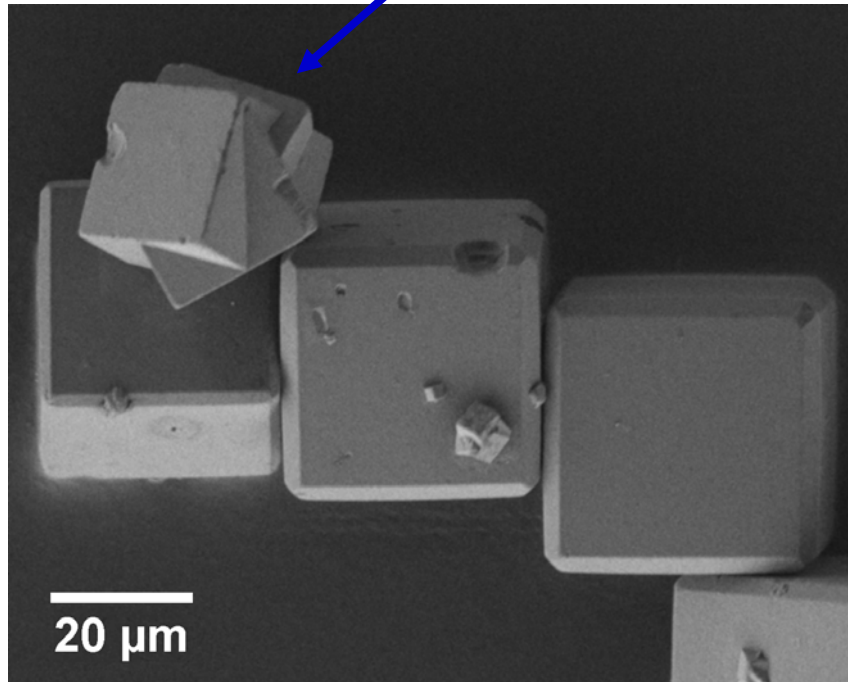
Stirred in 1 g 39% HF+12 g acetone
for 5 min.

Stirred in 1 g 39% HF+12 g acetone
for 10 min.

1. Loss of ~1/2 total weight
2. Shape and size of the recovered crystals unchanged
3. Serious internal damages/voides/pores.

Highlights of Synthesis Research: Characterization by SEM

- occasional intergrown twins



XY293_1

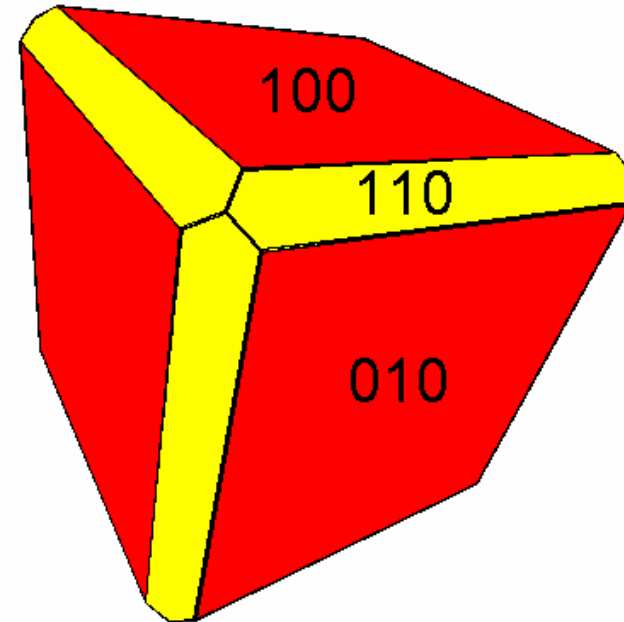
- cubes with truncated edges

6 faces of the type (100): $A_{100} = 80 \%$

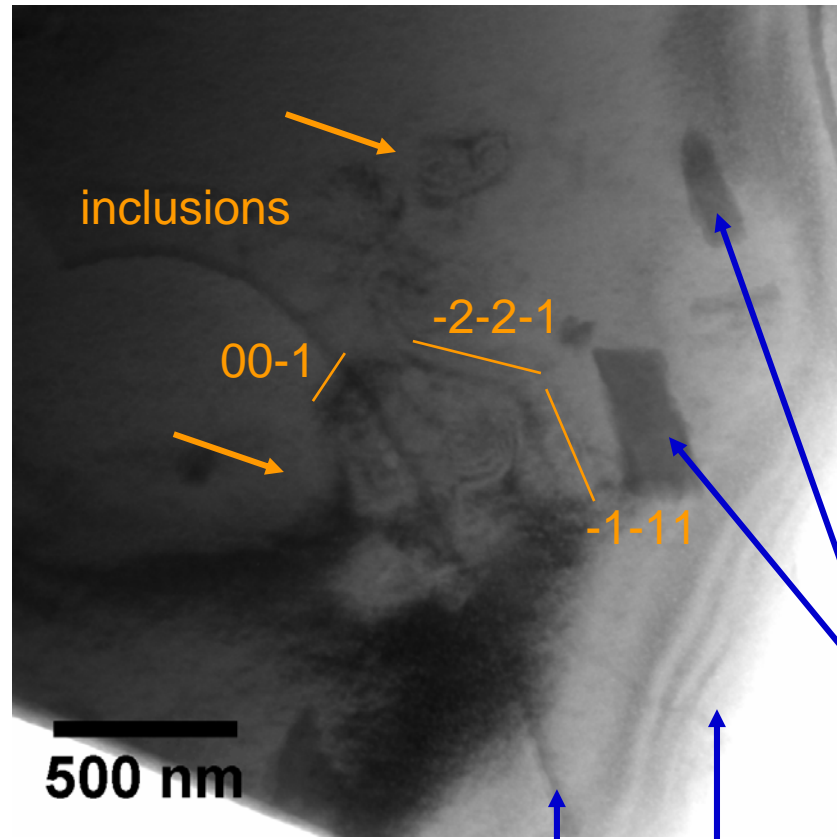
12 faces of the type (110): $A_{110} = 20 \%$

$$\Rightarrow \frac{A_{100}}{A_{110}} = 4 \quad (\text{estimated by SEM})$$

SEM Characterization: CaNaA



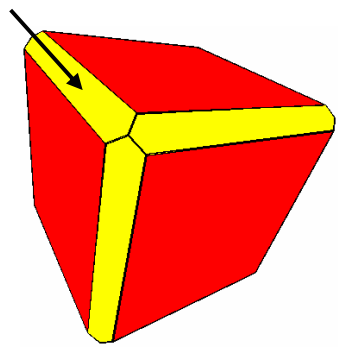
Highlights of Synthesis Research: Characterization by TEM



XY293_1

TEM
Characterization:
CaNaA

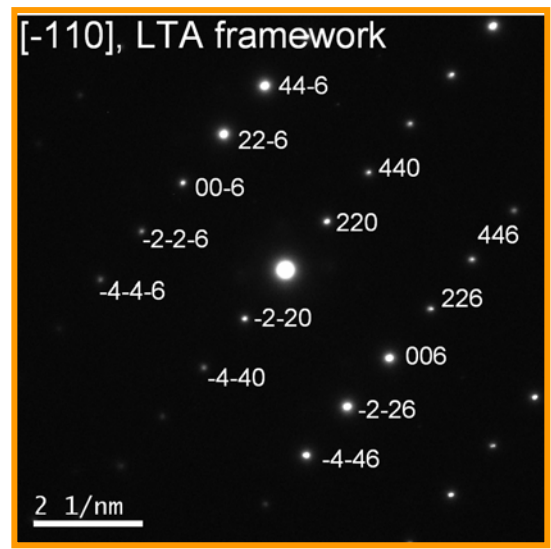
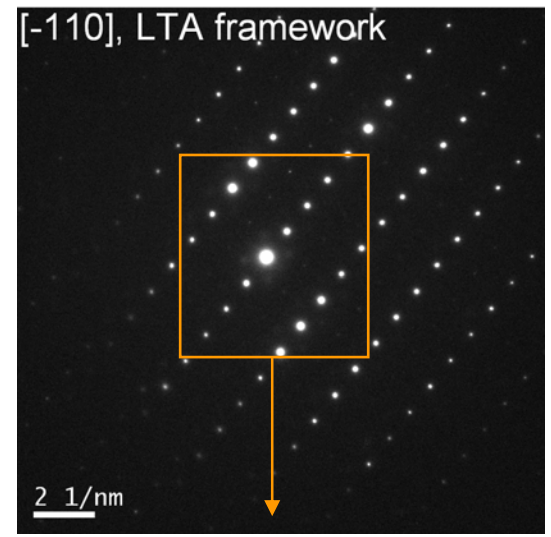
TEM view along
a normal of
a truncated edge



deposited
fragments due to
sample crushing

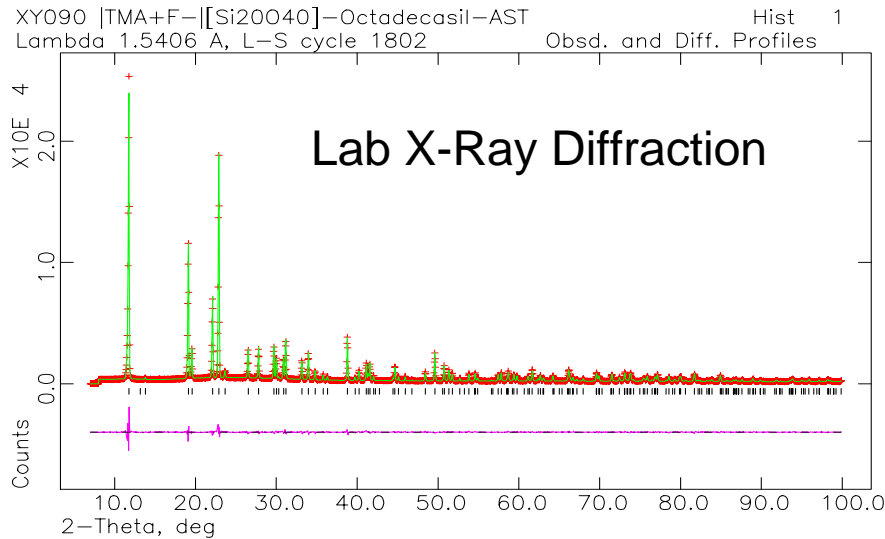
thickness fringes at
the edge of the crystal

edge of the carbon support film

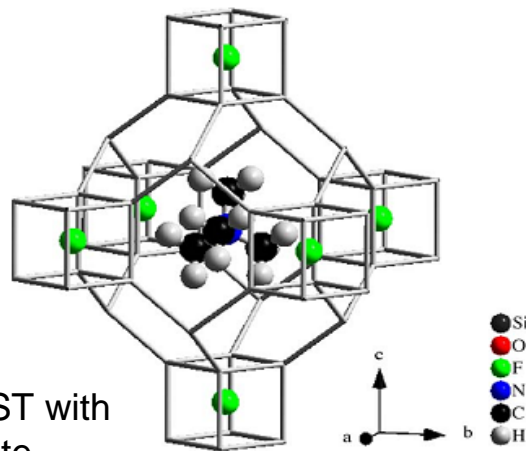
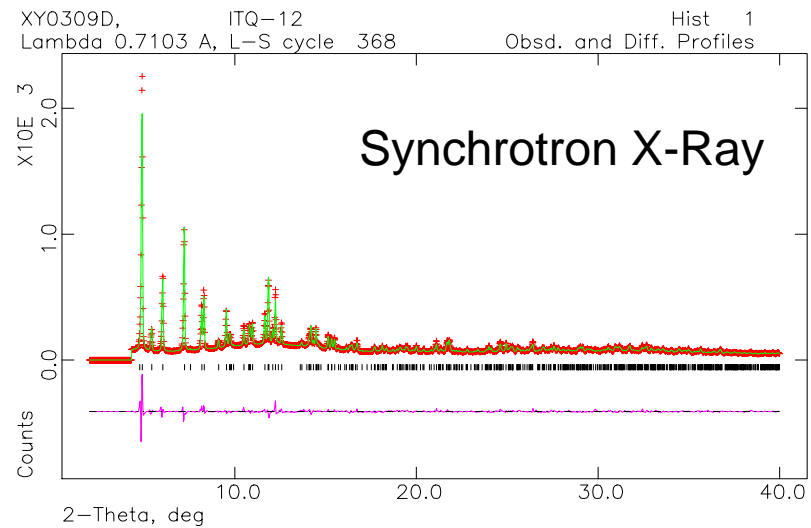
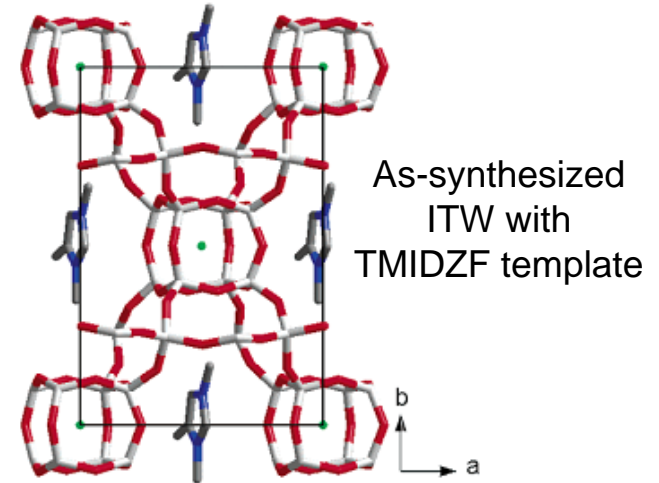


electron diffraction
shows crystallinity

Highlights of Synthesis Research: Crystallographic characterization of zeolitic host/guest systems



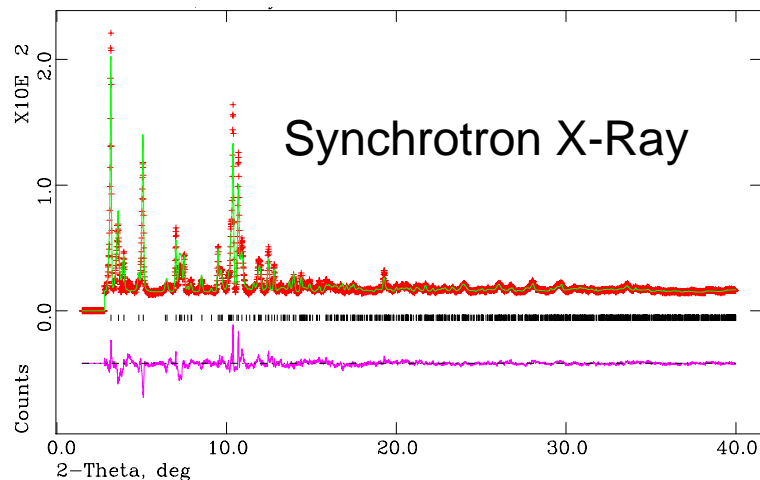
As-synthesized zeolites



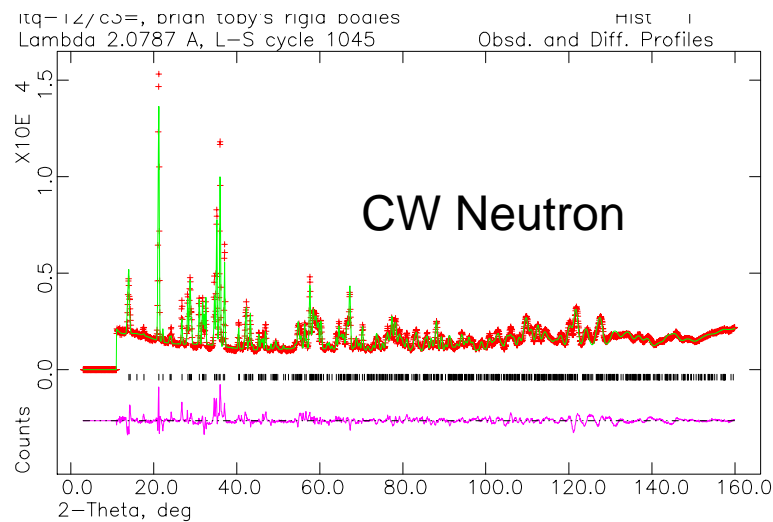
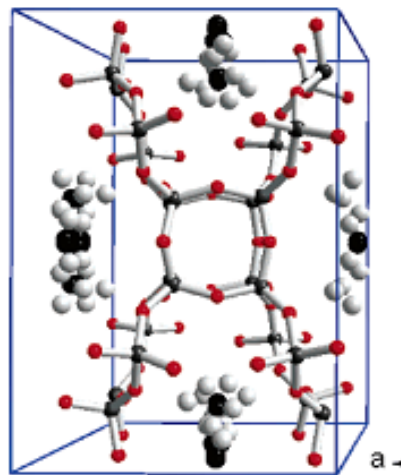
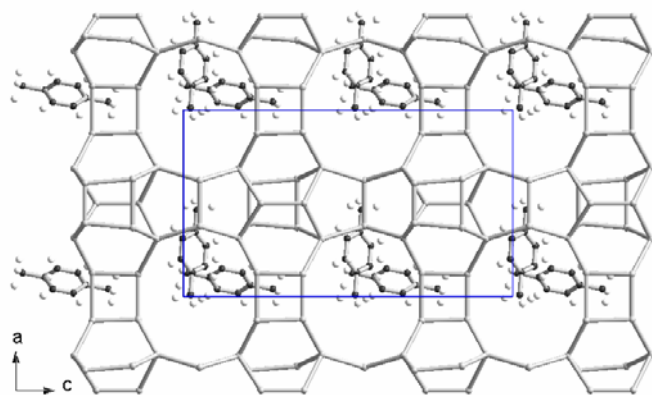
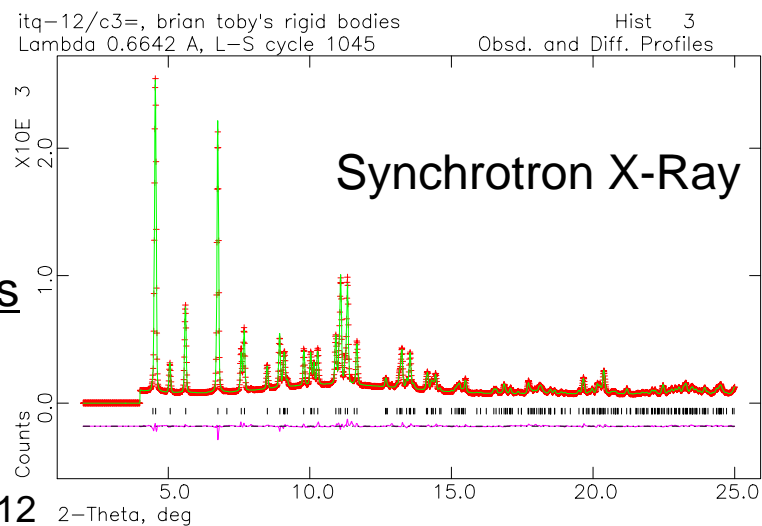
X. Yang, Materials Research Bulletin, 41 (2006) 54.

X. Yang, M.A. Camblor, Y. Lee, H. Liu, D.H. Olson, J. Am. Chem. Soc., 126 (2004) 10403.

Highlights of Synthesis Research: Crystallographic characterization of zeolitic host/guest systems



Adsorption Sites
in zeolites



X. Yang, B.H. Toby, M.A. Camblor, Y. Lee, D.H. Olson,
J. Phys. Chem. B, 109 (2005) 7894.

D.H. Olson, X. Yang, M.A. Camblor, J. Phys. Chem. B, 108 (2004) 11044.

Planned Activities vis. Achieved Results in the previous project 2004-2006

Planned	Achieved	Remark
NaCaA	NaA and NaCaA, 5-40 μm	Hannover, successful
Cation Free A	Not achieved	Guth's AIPO-LTA, or Corma's ITQ-29
Silicalite-1/ZSM-5	Extra-large crystals	Mühlheim
NaX	Extra-large crystals 15, 50 and 80 μm	Hannover, successful
NaY	Large crystals 5 μm	Stuttgart
AIPO ₄ -5	Up to 500 μm	Crystal size distribution unsatisfactory
Ferrierite	Extra-large crystal	Stuttgart

Aims of the Synthesis Work in the Second Period

- Synthesis of materials of the same framework types but of different chemical compositions;
- Control of crystal sizes and size distributions;
- Modification of surface and bulk microstructures.

Targets:

- Self-diffusivity measurements;
- Multi-component diffusion.

Working Programme Second Period

First periodicity
1-1.5 years

overlaps

Second periodicity
1.5-2.5 years

Framework Type	Material	Remark	Pore Dimension
LTA	Aluminosilicate = Linde type A (optional: ZK-4 and zeolite alpha with Si/Al up to 5)	Variable Si/Al ratio	3D, 8-ring
	Pure SiO ₂ = ITQ-29	Corma's LTA	
	Aluminium phosphate	Guth's AlPO ₄	
MFI	Pure SiO ₂ = silicalite-1	Pentasil	3D, 10-ring
	(optional: Aluminosilicate = ZSM-5)		
(optional: MEL)	Pure SiO ₂ = silicalite-2		
	Aluminosilicate = ZSM-11		
FAU	Aluminosilicate Zeolite X	Variable Si/Al ratio	3D, 12-ring
	Aluminosilicate Zeolite Y		
FER	Pure SiO ₂ ferrierite	Gies' invention	2D, 8 and 10-ring
	(optional: Aluminosilicate ferrierite)	Vaughan's work in 1966	
AFI	Aluminium phosphate = AlPO ₄ -5	Caro's standard	1D, 12-ring
	Aluminosilicate SSZ-24	Zones' invention	

Resources

Available in the Institute:

- SEM and TEM for microstructure analysis;
- Powder XRD with heatable sample stage for phase identification;
- Other instruments for chemical and physical (thermal) analyses, etc.

Purchased in the first period:

- Autoclaves for conditional experiments, 5 X 23 mL, 10 X 45 mL, 5 X 100 mL;
- Autoclave (mini-reactor) for scale-up, 1 X 300 mL.

Needed for the second period:

- Heating blocks, 2 X for 100 mL autoclaves, 5000 €;
- Autoclaves for conditional experiments, 10 X 100 mL, 20 000 €;
- Autoclave for scale-up, 1 X 1000 mL,
with stirring and heating accessories, 29 000 €.