

# Solid-State Diffusion and NMR

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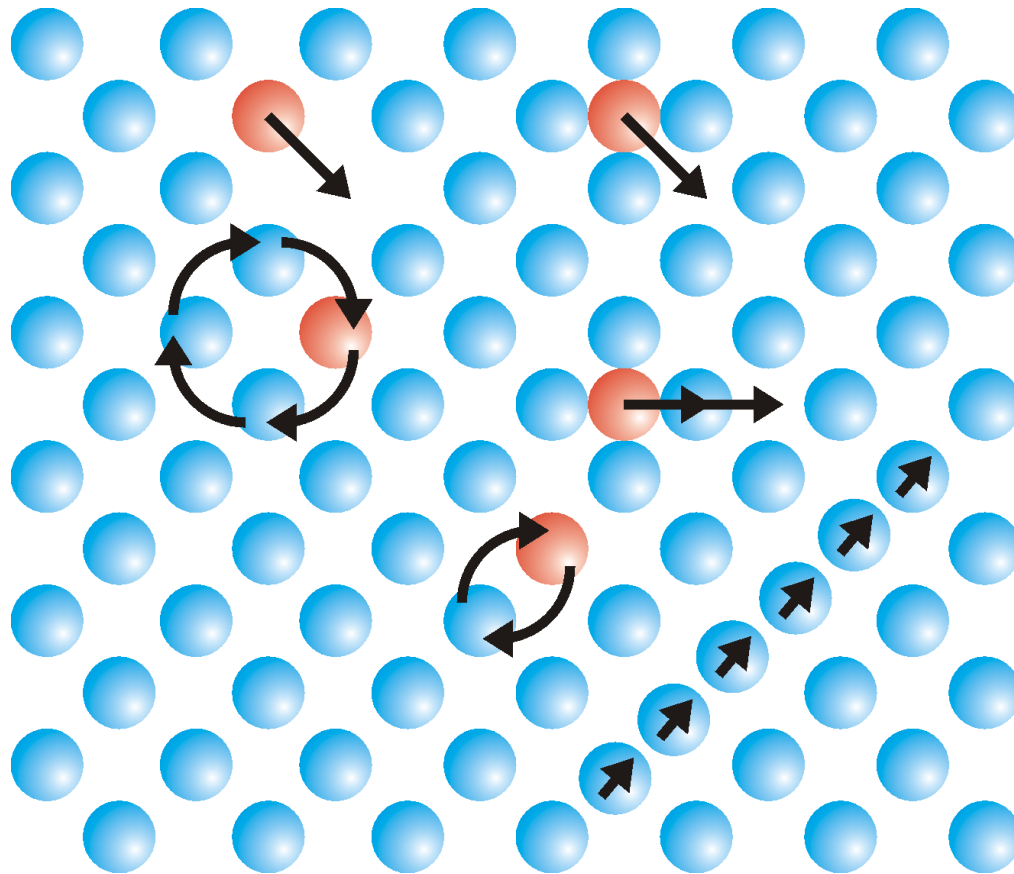
Diffusion Fundamentals, Leipzig, 23 Sept. 2005

# Introduction

- Diffusivity in Solids as Compared to Liquids and Gases

	$D / \text{m}^2 \text{s}^{-1}$	<i>time</i> for 1 cm	
• Gases	$10^{-4}$	1 s	} RT
• Liquids	$10^{-9}$	1 d	
• Solids	$< 10^{-13}$	$> 30 \text{ a}$	} $< T_m$
• Interfaces/ Surfaces	$< 10^{-9}$	$> 1 \text{ d}$	

- Reason for Slow Diffusion in Solids:  
Formation of Defects is needed



Activation Energy

$$D \sim e^{-\overbrace{(E_F + E_M)} / k_B T}$$

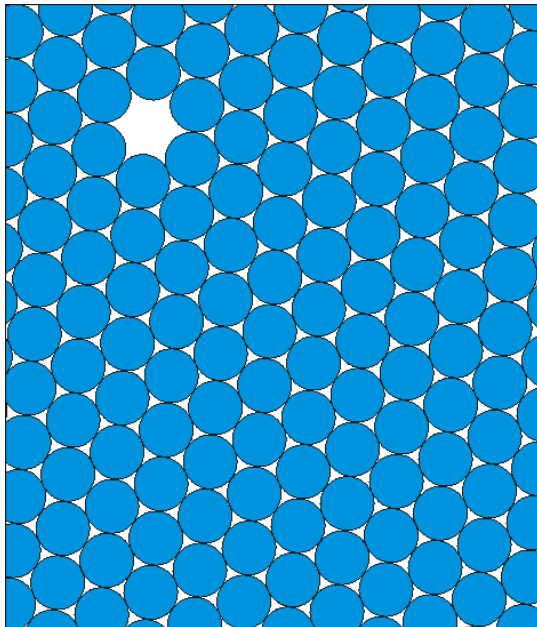
$E_F > 0$  for Solids

$E_F \approx 0$  for Liquids,  
Gases

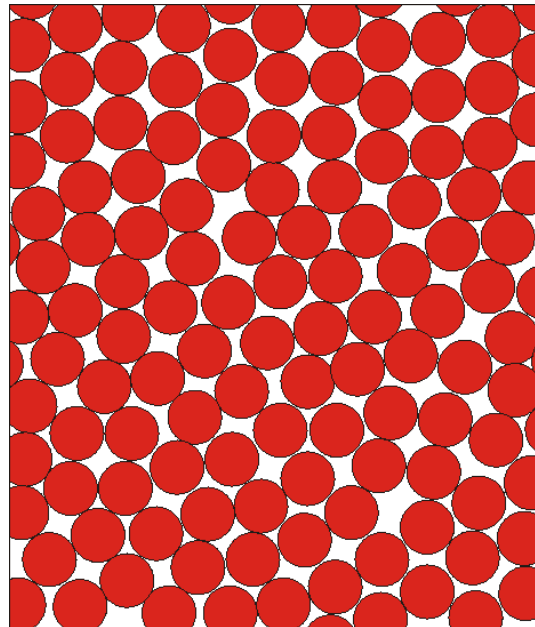
after Philibert: „Diffusion et Transport  
de Matière dans les Solides“ (1985)

- Overview: Defective Solids

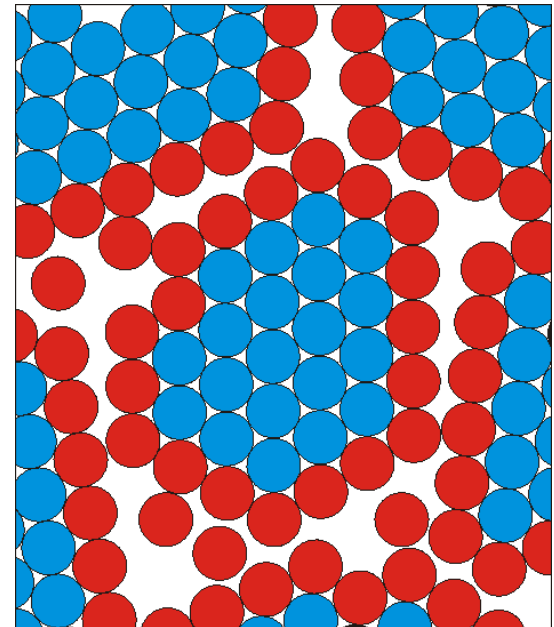
single  
crystalline



amorphous

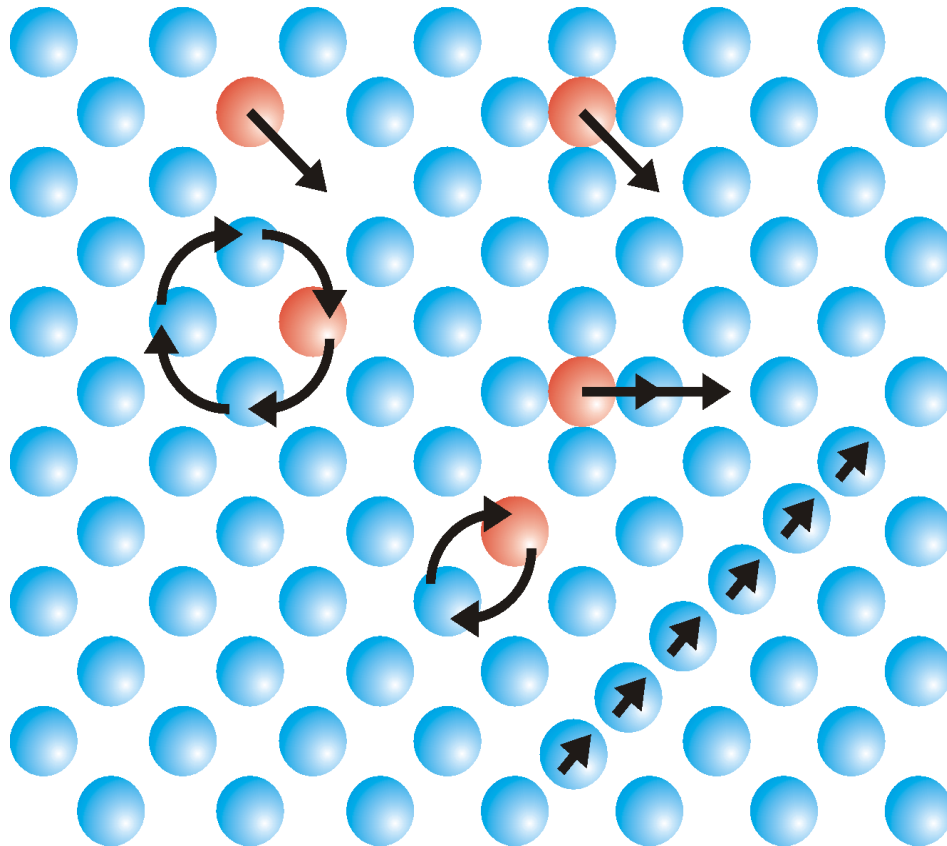


nano-  
crystalline

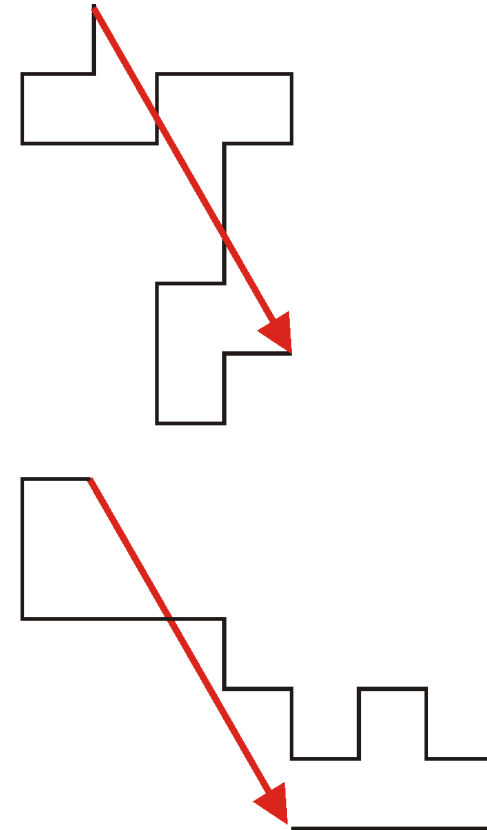


(see Talk: Chadwick)

- Microscopic and Macroscopic Aspects of Diffusion



elementary jumps



macroscopic transport

- Microscopic and Macroscopic Diffusion Quantities

Jump rate  $\tau^{-1} \cdot \frac{r^2}{6} \cdot f = D^T$  Tracer diffusivity

Correlation factor  $f \leq 1 \Rightarrow$  Diff.mechanism (see Talk: Murch)

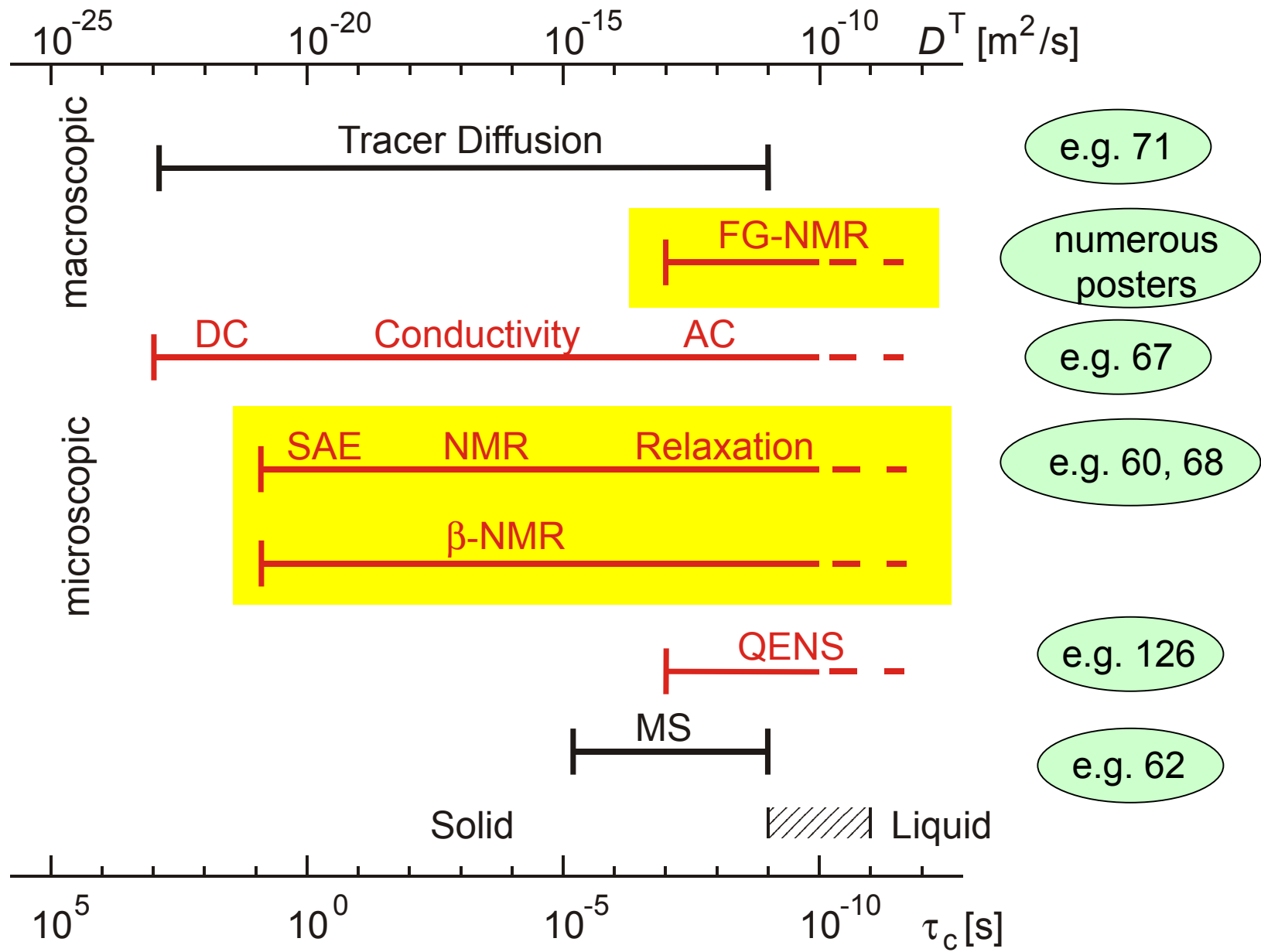
$$\tau^{-1} \approx 10^6 \text{ s}^{-1} \text{ at RT}$$

Temperature dep.  $\Rightarrow E_A$  (depends on time window)

# Experimental Methods

<i>Microscopic</i>	<i>Macroscopic</i>
<ul style="list-style-type: none"><li>• NMR Relaxation / Lineshape Spin alignment echo</li><li>• <math>\beta</math>-radiation detected NMR</li><li>• Quasielastic neutron scattering</li></ul>	<ul style="list-style-type: none"><li>• Field gradient NMR Pulsed / Static</li><li>Radioactive tracer</li><li>Ion beam analysis</li></ul>
<ul style="list-style-type: none"><li>• AC conductivity</li></ul>	<ul style="list-style-type: none"><li>• DC conductivity</li></ul>

# • Ranges of Diffusivities and Correlation Times

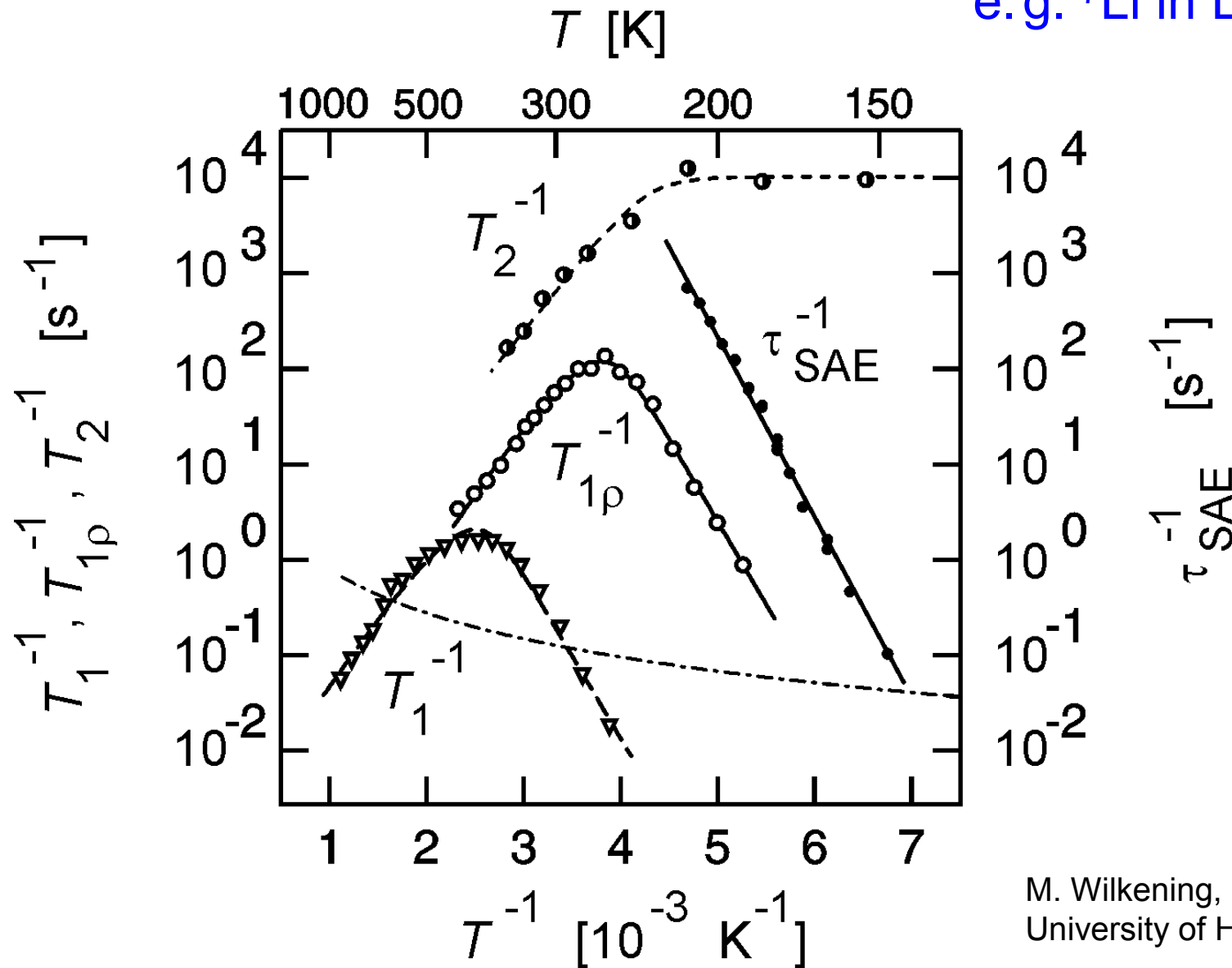




- Overview NMR

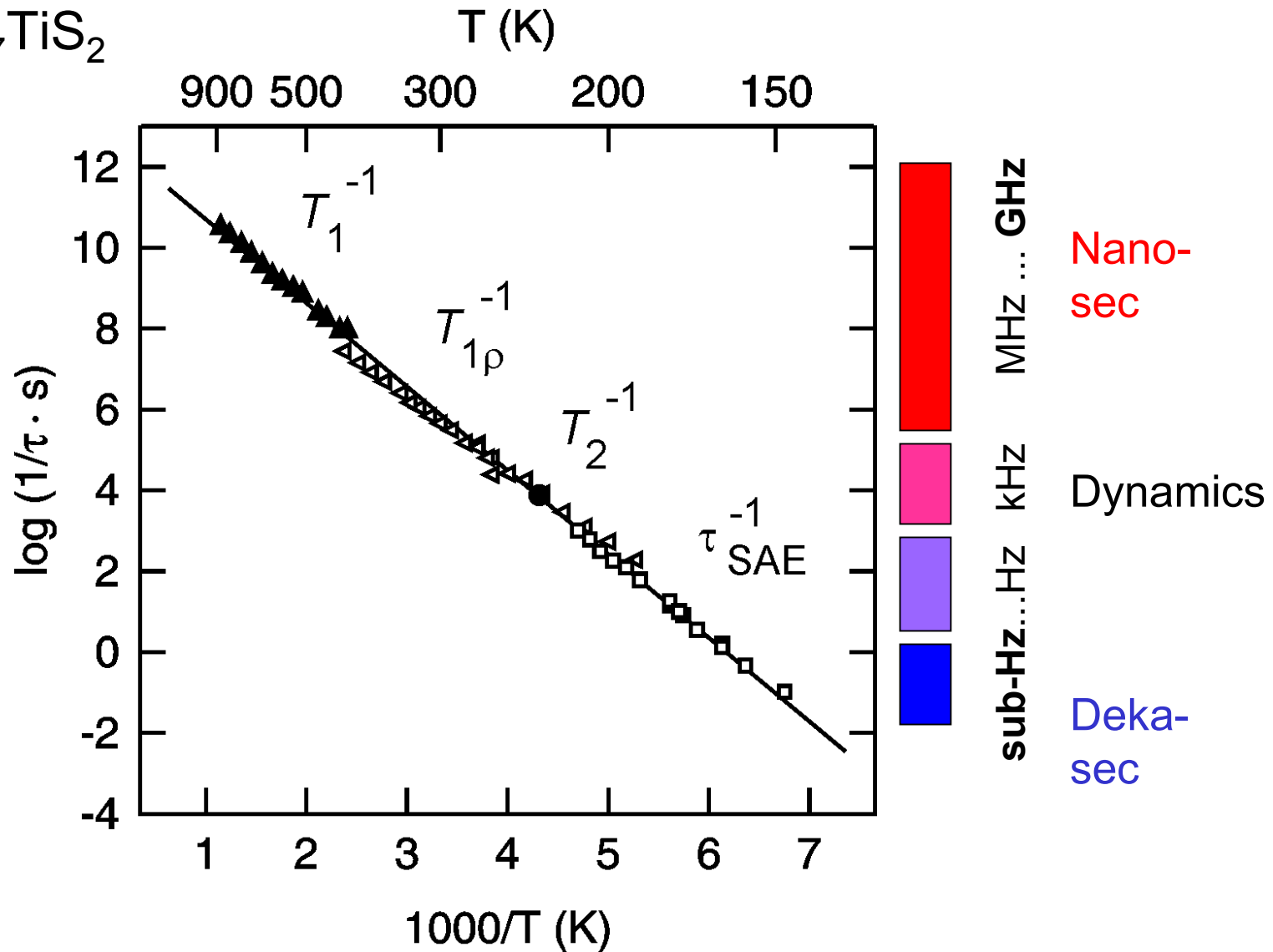
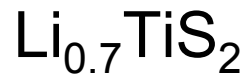
Spin-Lattice Relaxation, Spin-Spin Relaxation;  
Spin-Alignment Echo

e.g.  ${}^7\text{Li}$  in  $\text{Li}_{0.7}\text{TiS}_2$

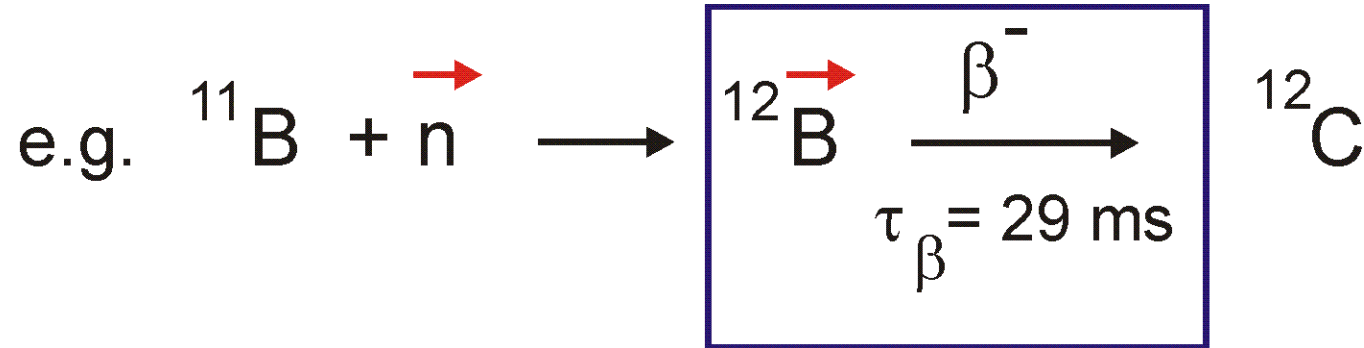


M. Wilkening, PhD thesis,  
University of Hannover, 2005.

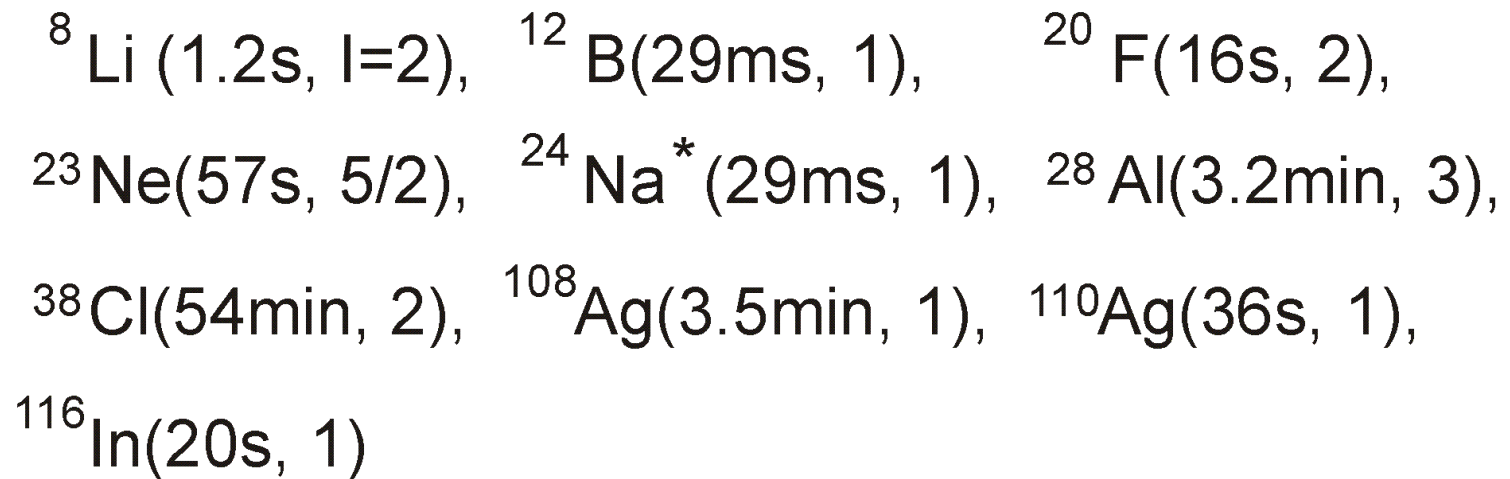
- Motional Correlation Rates



# Beta-NMR: Principle (1)

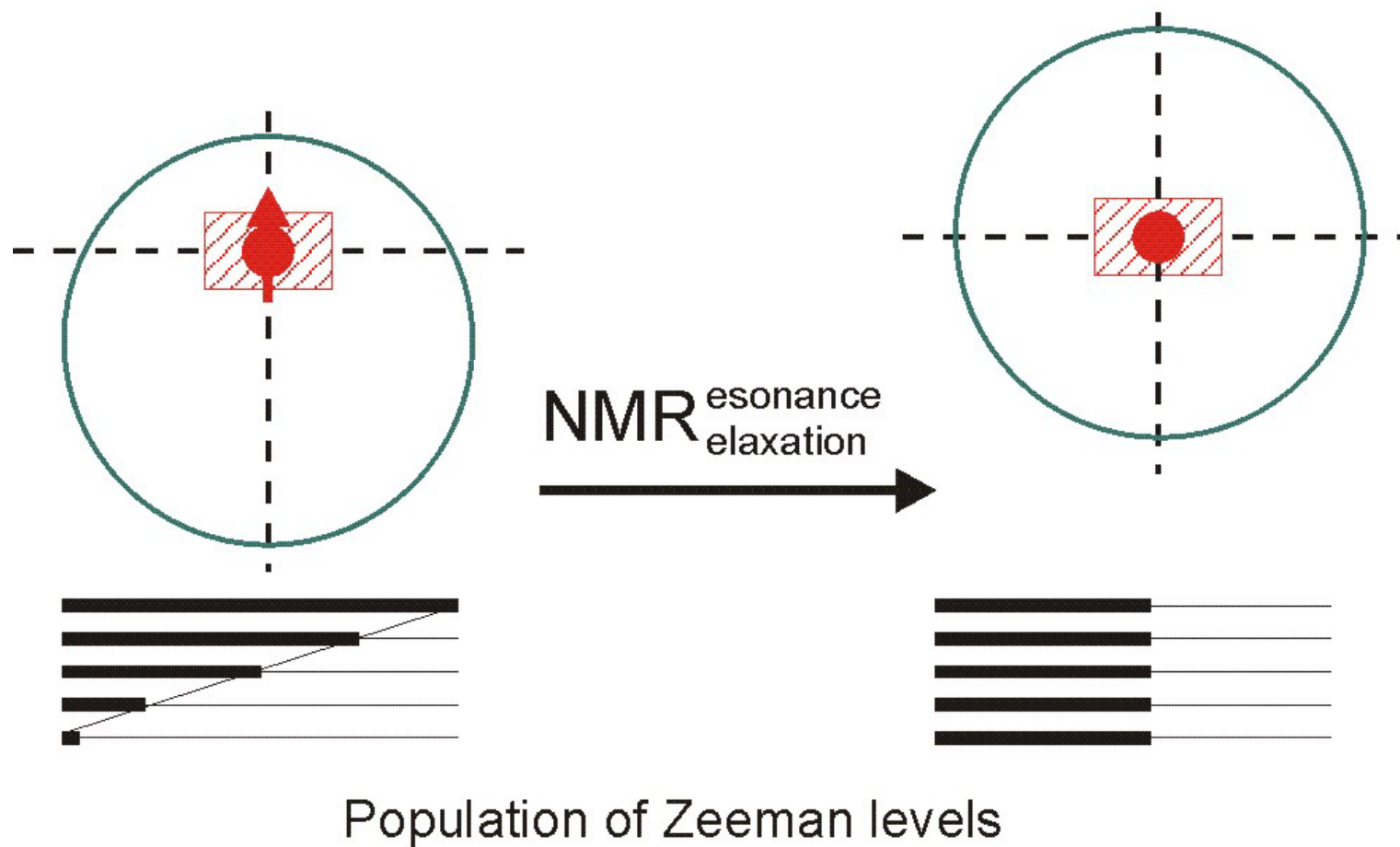


Probes used so far:

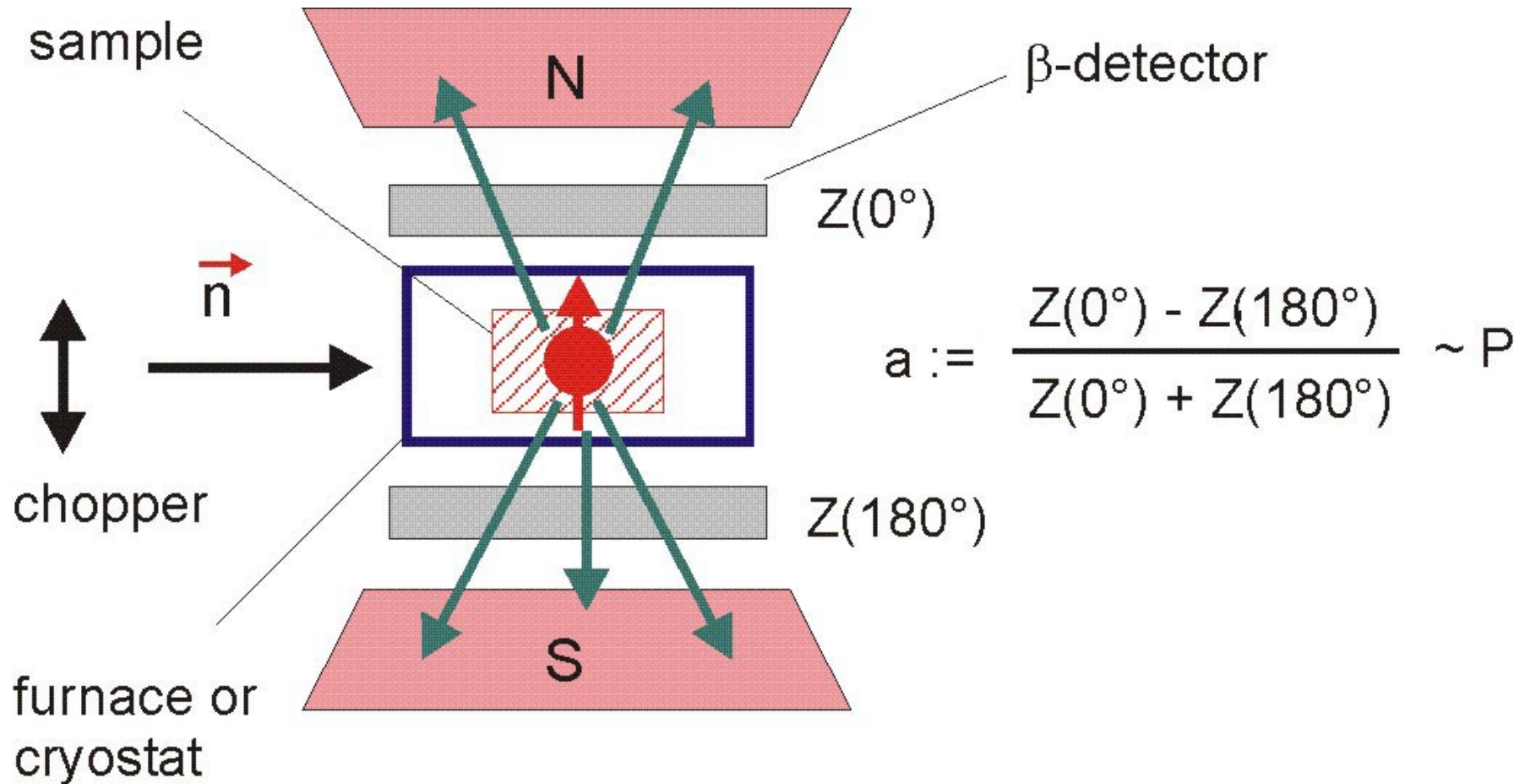


## Beta-NMR: Principle (2)

Angular distribution of  $\beta$ -radiation  
asymmetric as long as nuclei are polarized

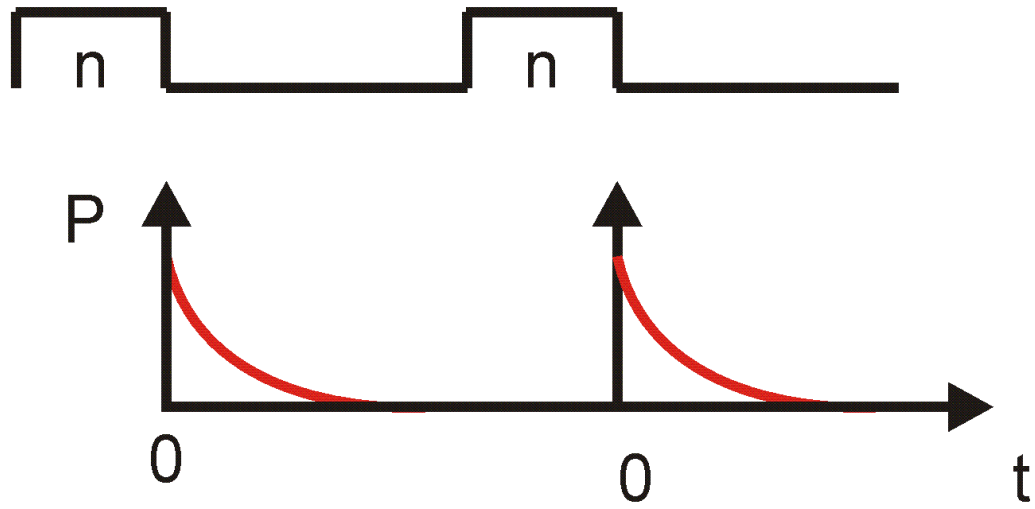


# Beta-NMR: Setup



# Beta-NMR: Operating Modes

- transients  $P(t)$  after n-activation pulses : spin-lattice relaxation (SLR)

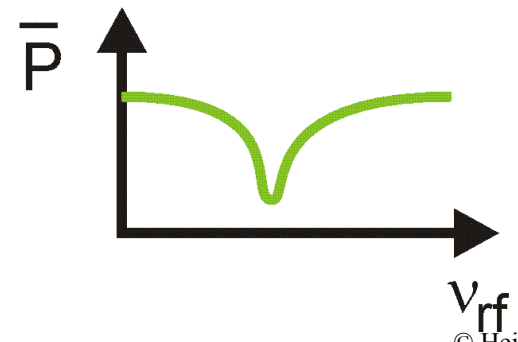


$$P = P_0 \cdot e^{-t/T_1}$$

$T_1 = \text{SLR time}$

- stationary  $\bar{P}(v_{rf})$  : resonance spectra

$$\bar{P} \sim \bar{a} = \frac{a_0}{1 + \tau_\beta / T_1}$$



# Beta-NMR: Some Features and Implications (1)

- P ( $\approx 10\%$ ) independent of Boltzmann factor  
→ low B, high T accessible
- SLR measurements do *not* require rf fields  
→ B easily variable  
→ no skin effect: metallic samples/containers
- SLR time window:  $0.01 \tau_{\beta} < T_1 < 100 \tau_{\beta}$

## Beta-NMR: Some Features and Implications (2)

- Concentration of probes extremely small ( $1:10^{18}$ )
  - probes surrounded only by *unlike* nuclei
  - no spin diffusion
    - no SLR by distant paramagnetic impurities
    - inequivalent sites: inhomogeneous SLR

- Complementary probes

e.g. Q=0 for NMR

$^{19}\text{F}$  (100%)

$^{107}\text{Ag}, ^{109}\text{Ag}$  (52%+48%)

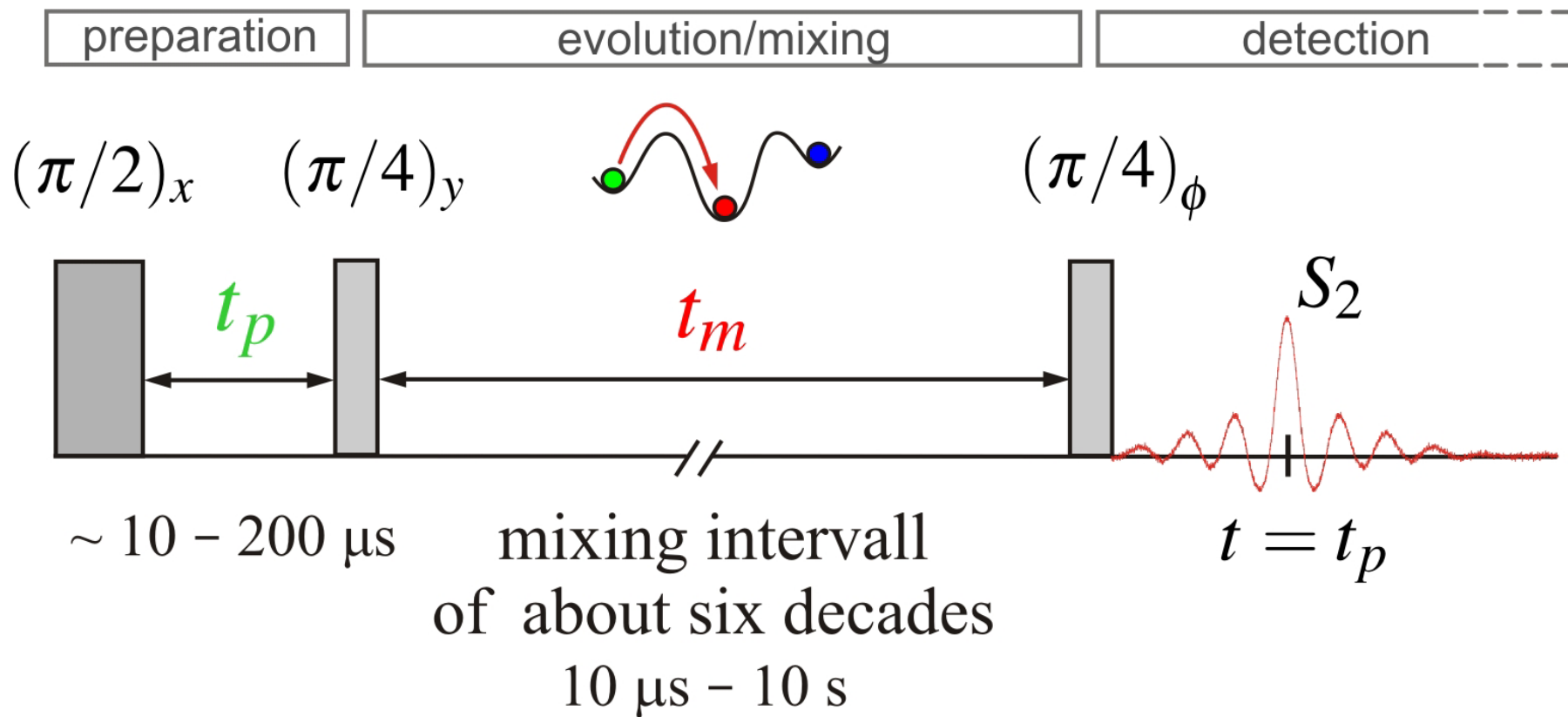
Q $\neq$ 0 for  $\beta$ -NMR probe

$^{20}\text{F}$

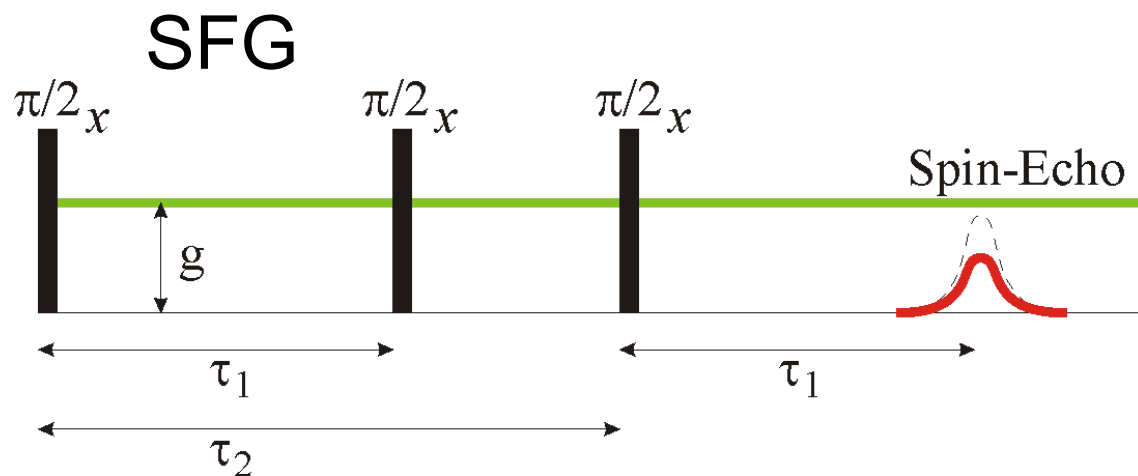
$^{108}\text{Ag}, ^{110}\text{Ag}$



# Multiple Time NMR: Spin-Alignment Echo (SAE)



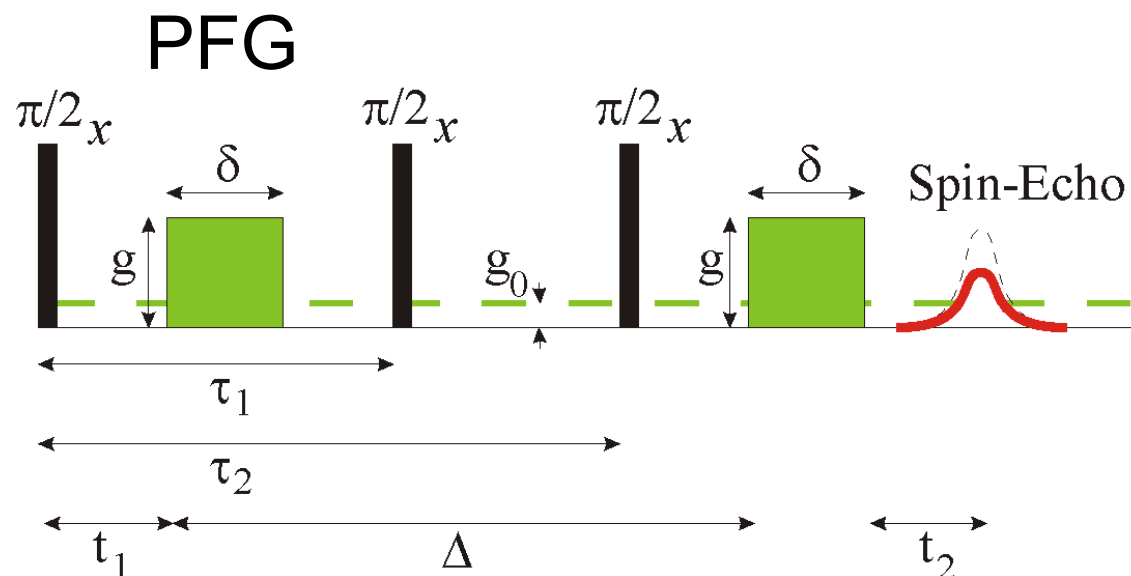
# Macroscopic Diffusion Measurement. in a Field Gradient



$$\frac{M(\tau_1 + \tau_2)}{M(0)/2} =$$

$$\exp\left(-\frac{\tau_2 - \tau_1}{T_1} - \frac{2\tau_1}{T_2}\right)$$

$$\cdot \exp\left(-D^\top \gamma^2 g^2 \tau_1^2 \left(\tau_2 - \frac{\tau_1}{3}\right)\right)$$



$$\frac{M(\tau_1 + \tau_2)}{M(0)/2} =$$

$$\exp\left(-\frac{\tau_2 - \tau_1}{T_1} - \frac{2\tau_1}{T_2}\right)$$

$$\cdot \exp\left(-D^\top \gamma^2 g^2 \delta^2 \left(\Delta - \frac{\delta}{3}\right)\right)$$

# Case Studies:

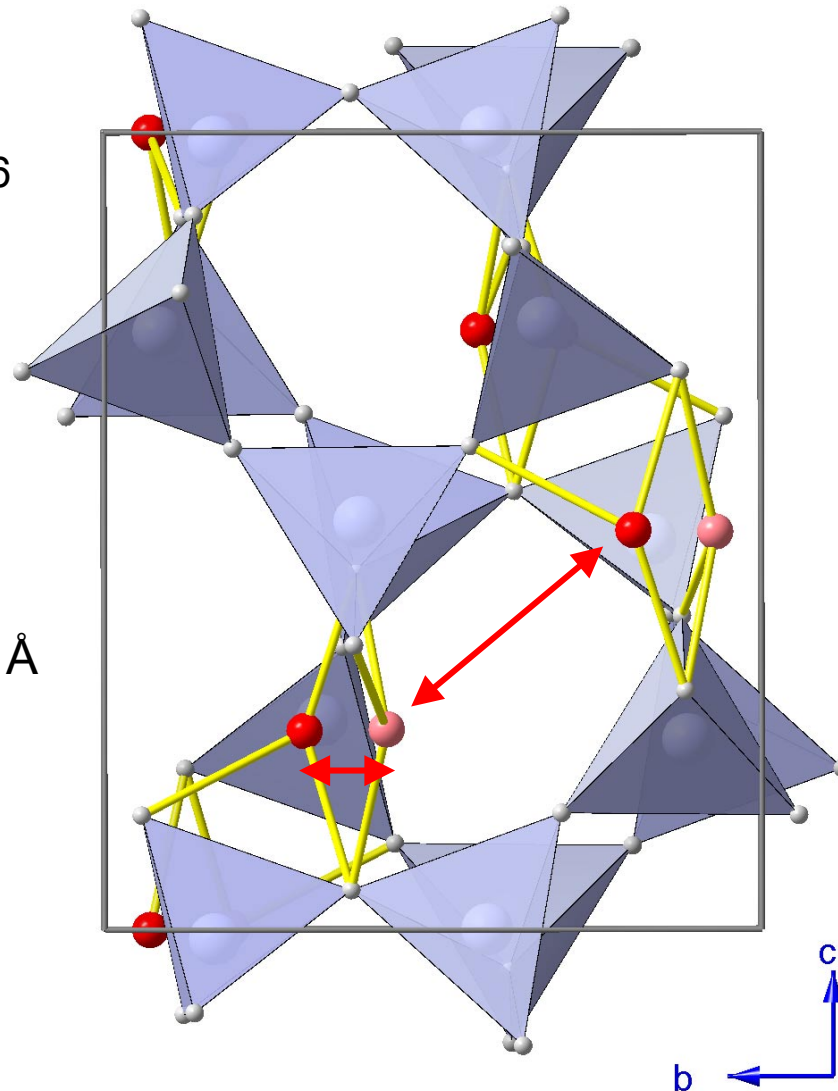
- Glassy and Crystalline Spodumene  $\text{LiAlSi}_2\text{O}_6$

$\beta\text{-LiAlSi}_2\text{O}_6$

$P4_32_12$

$a = b = 7.541 \text{ \AA}$

$c = 9.159 \text{ \AA}$

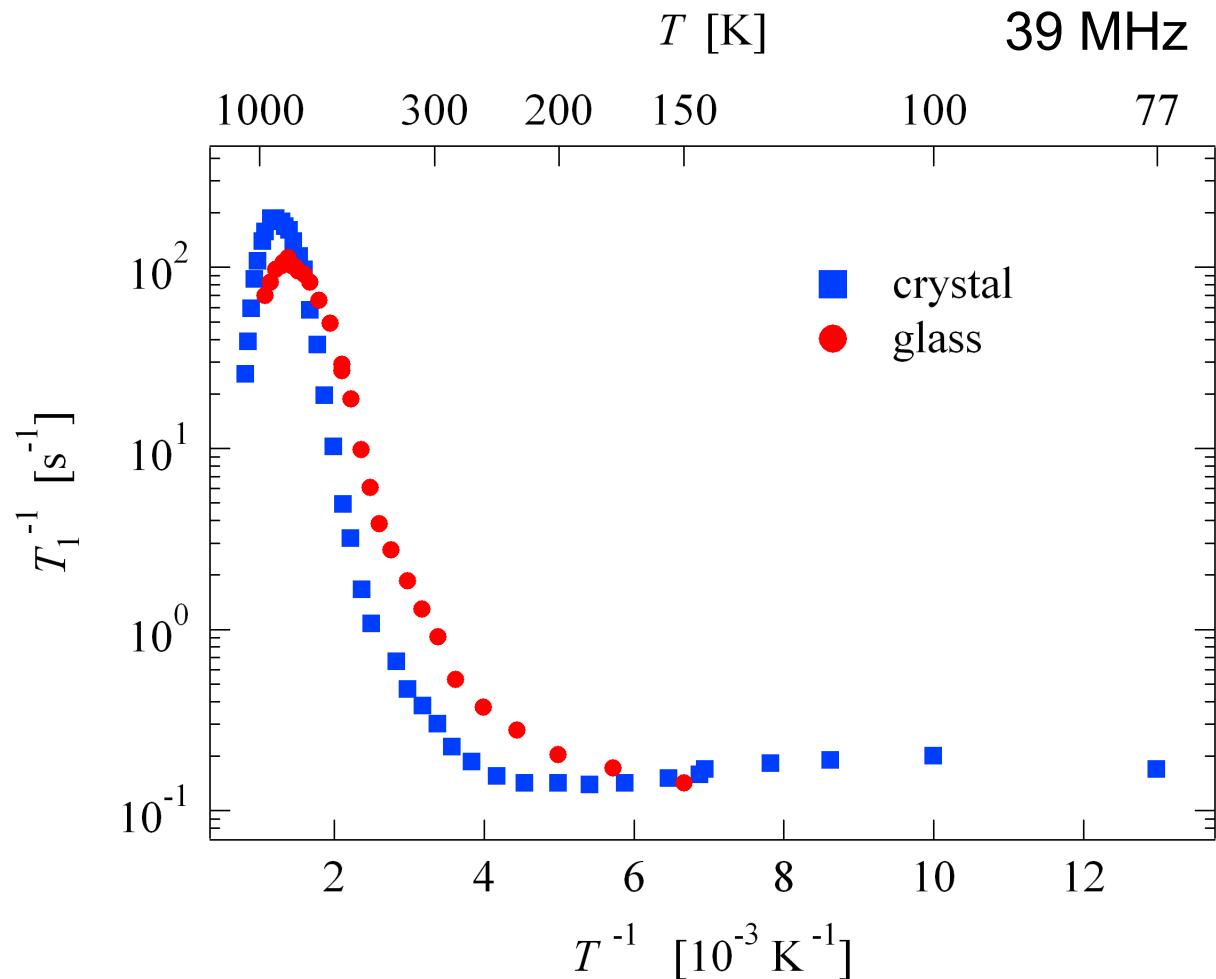


4 pairs of Li sites per unit cell (distance of neighbored paires:  $4.5 \text{ \AA}$ ):

only one site of each pair occupied (distance only  $1.3 \text{ \AA}$ )

→ long-range and short-range jumps of Li ions ?

- $^7\text{Li}$  Spin-Lattice Relaxation  
in Glassy and Crystalline Spodumene  $\text{LiAlSi}_2\text{O}_6$



high- $T$  peak:

long-range  
Li diffusion

faster in the glass

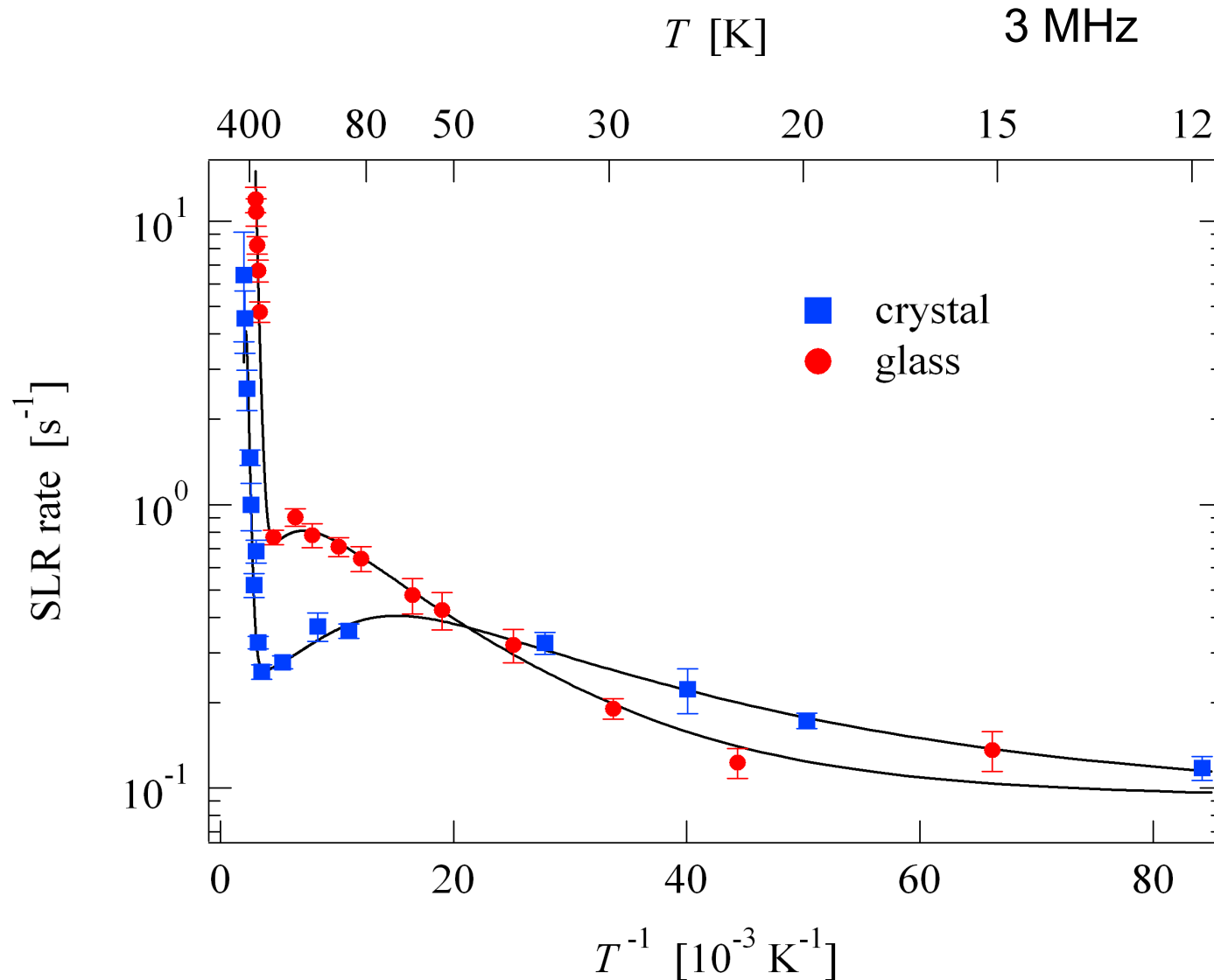
$$E_A(\text{glass}) = 0.34\text{eV}$$

$$E_A(\text{cryst.}) = 0.50\text{eV}$$

low- $T$  peak:

short-range  
(local) jumps

# $^8\text{Li}$ $\beta$ -NMR Spin-Lattice Relaxation in Glassy and Crystalline Spodumene $\text{LiAlSi}_2\text{O}_6$



low- $T$  peak:

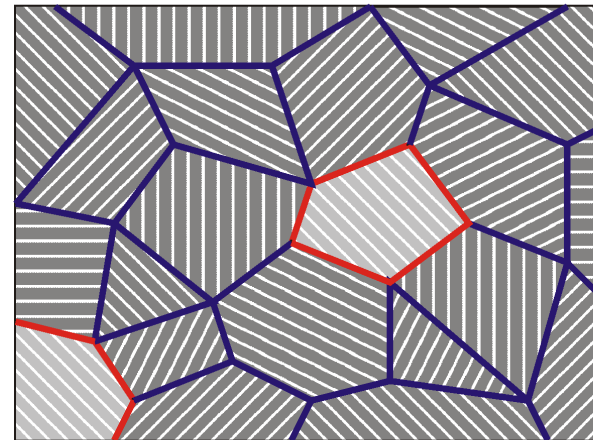
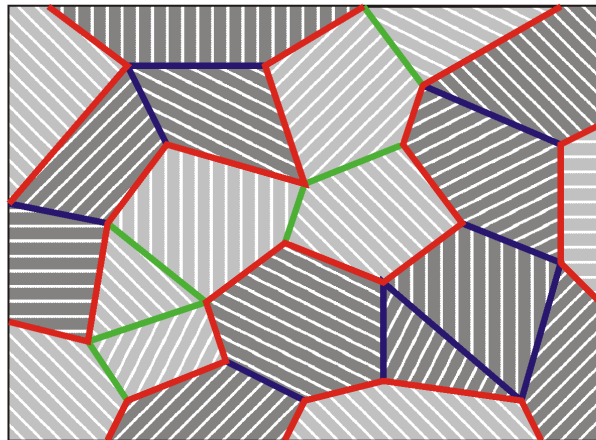
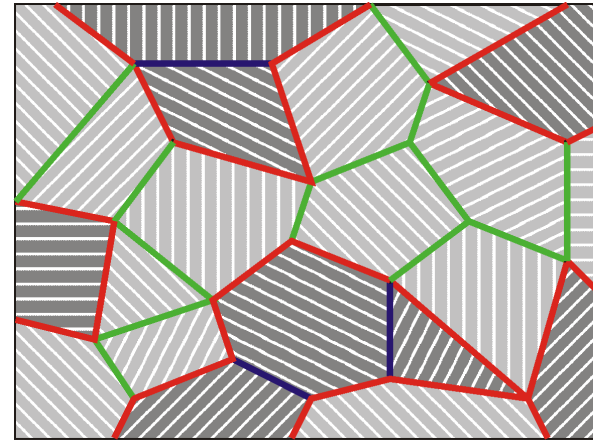
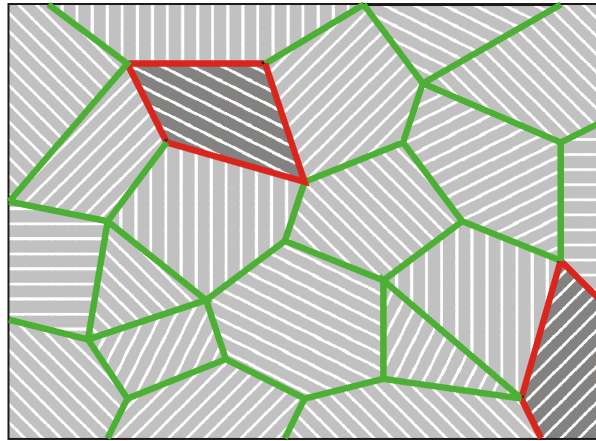
localized  
Li motion  
between the  
pair sites in  
the crystal

$$E_A \approx 50 \text{ meV}$$

From:

„Diffusion in Condensed Matter  
- Methods, Materials, Models“,  
P. Heitjans, J. Kärger (Eds.),  
Springer, Berlin 2005

- Nanocrystalline Composites



Ionic Conductor Grain



Insulator Grain

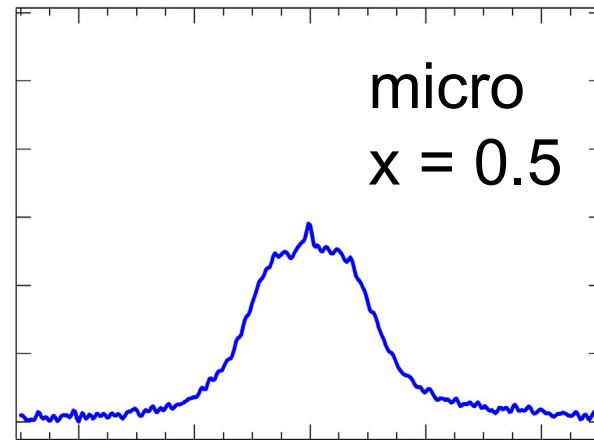
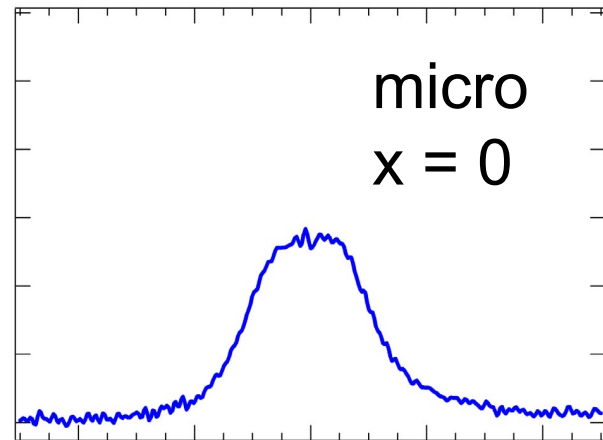
Interface between

— Insulators

— Ionic Conductors

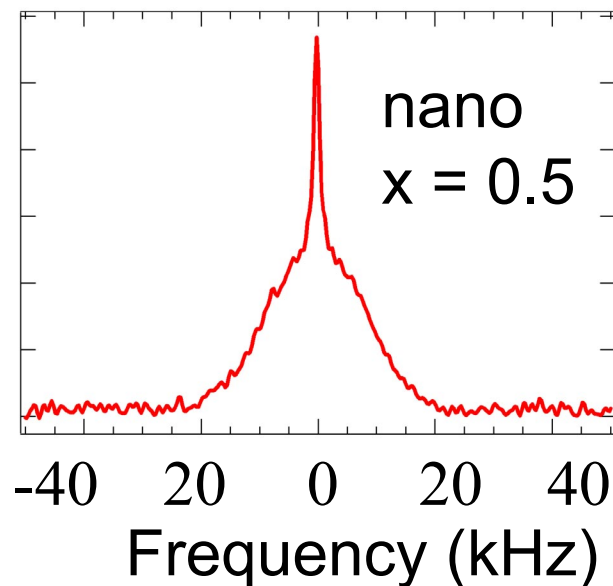
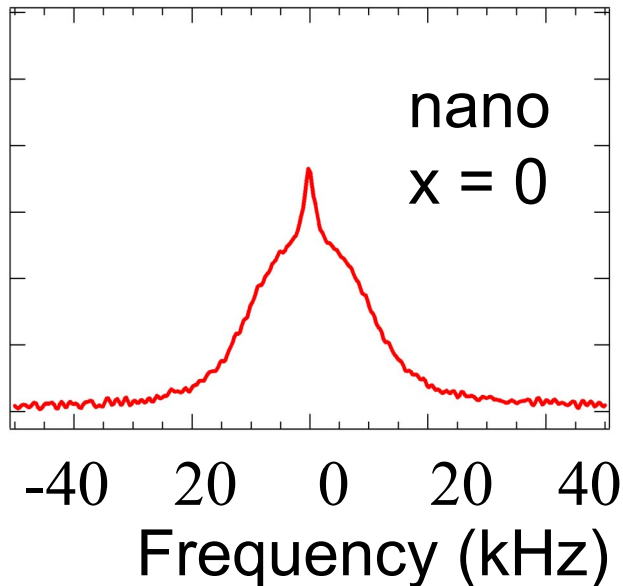
— Ionic Conductor & Insulator

- ${}^7\text{Li}$  NMR Lineshapes:  $(1-x)\text{Li}_2\text{O}:x\text{B}_2\text{O}_3$



$T = 433$  K

**micro:**  
one-component  
line

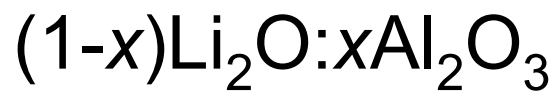


**nano:**  
two-component  
line

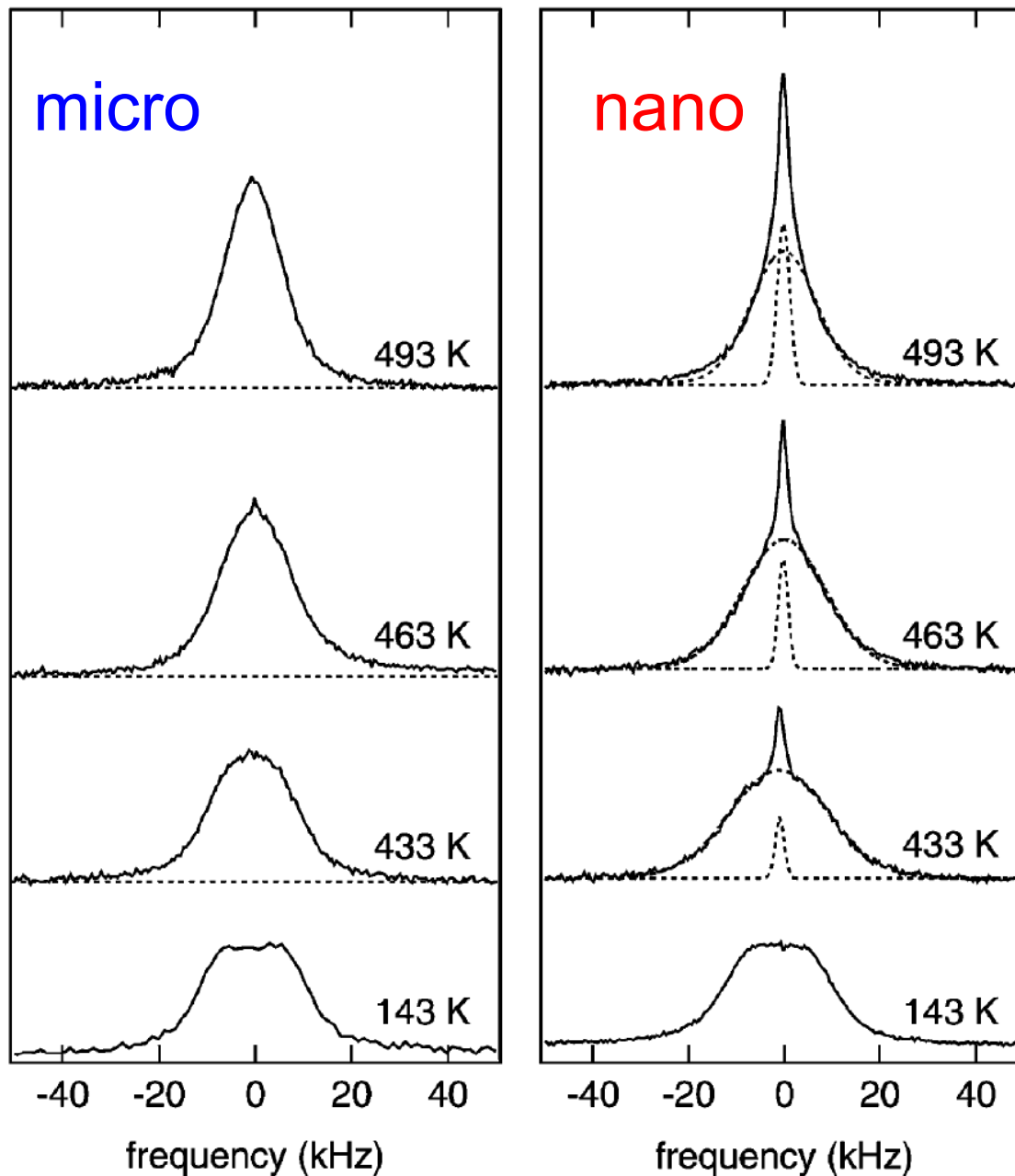
S. Indris et al.,  
J. Non-Cryst. Solids  
307-310 (2002) 555

# $^7\text{Li}$ -NMR

Lineshapes:



$x=0.5$

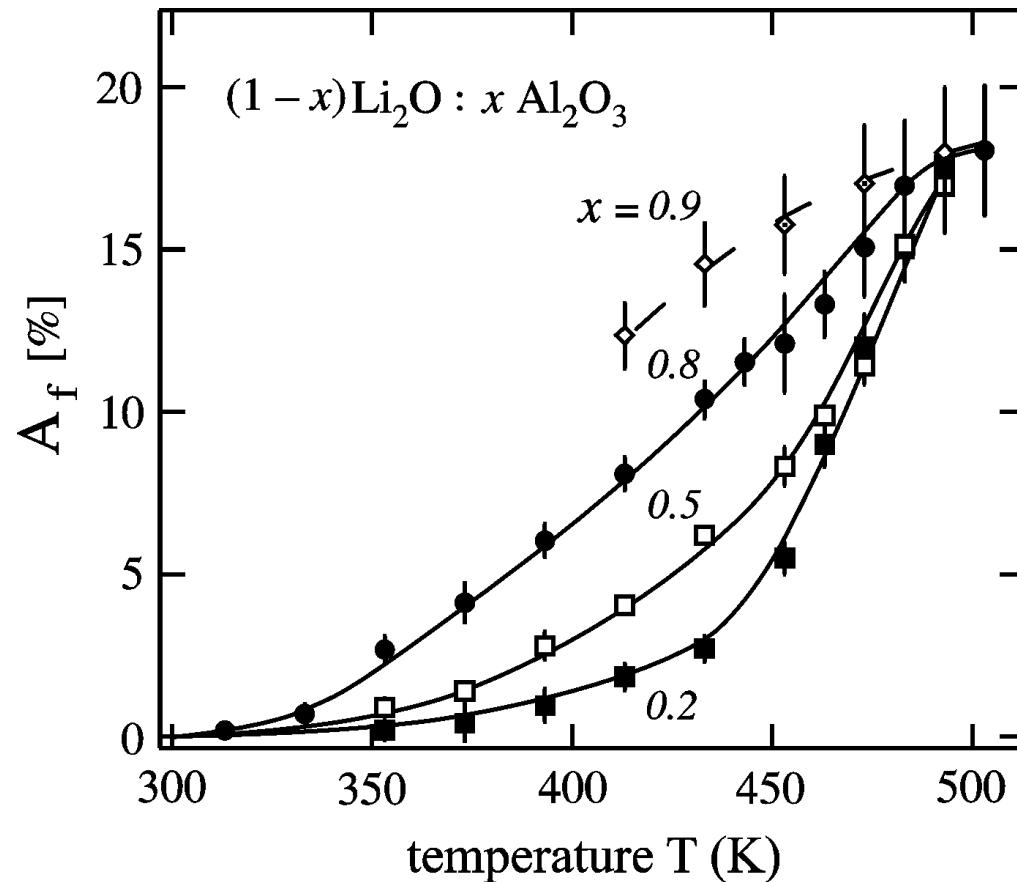


M. Wilkening et al.,  
Phys. Chem. Chem. Phys.  
5 (2003) 2225



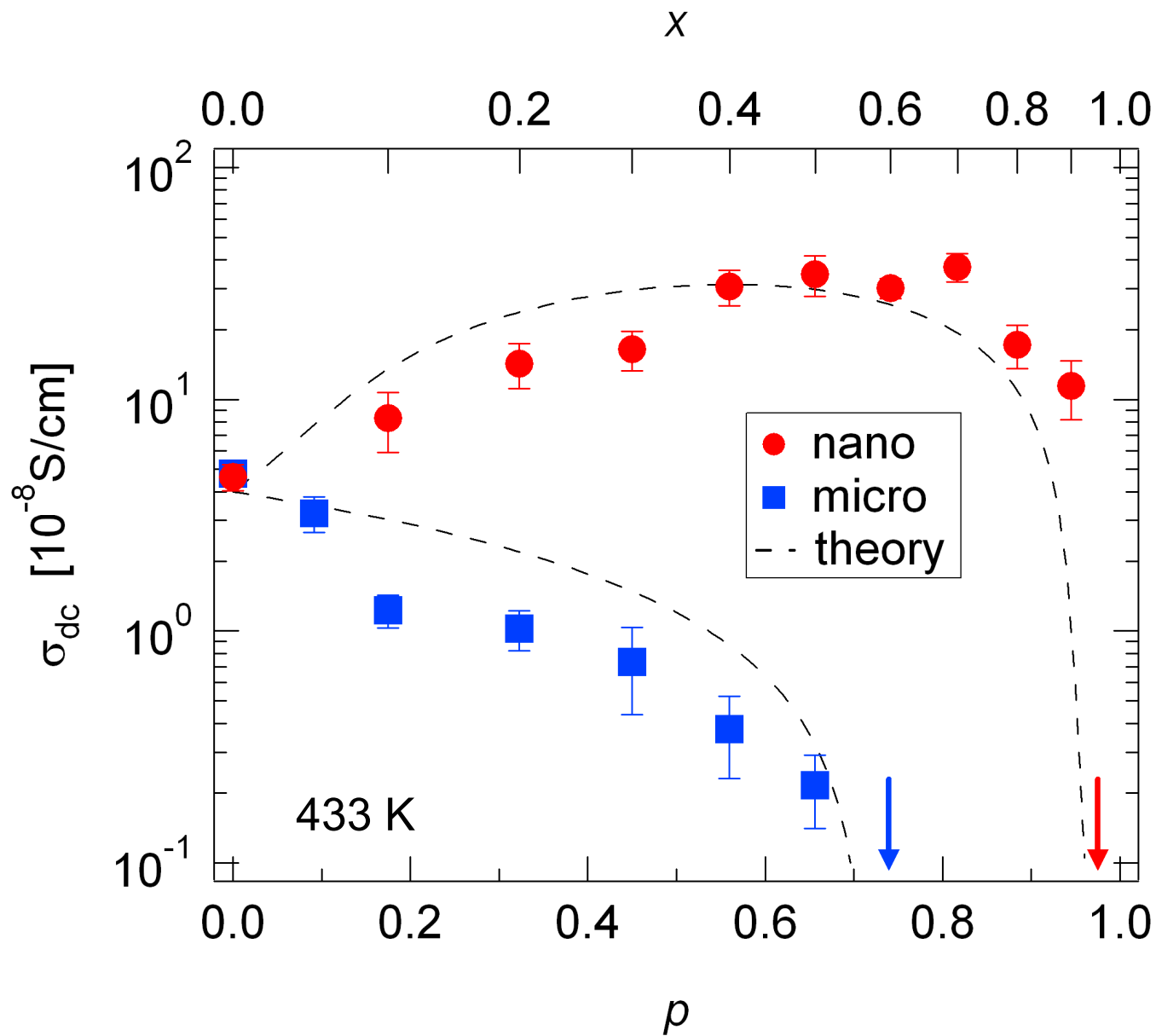
$^7\text{Li}$ -NMR Lineshape: **nanocryst.**  $(1-x)\text{Li}_2\text{O}:x\text{Al}_2\text{O}_3$

Fraction of mobile  $\text{Li}^+$



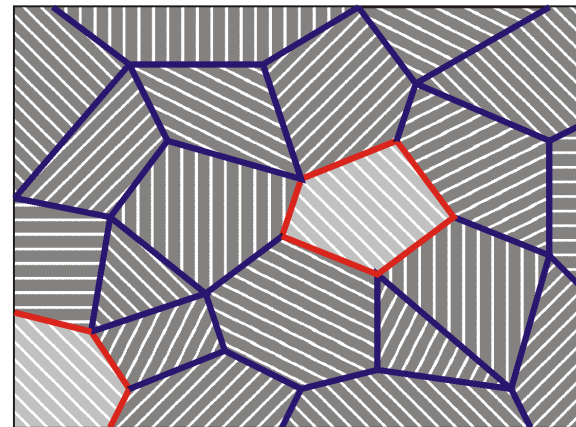
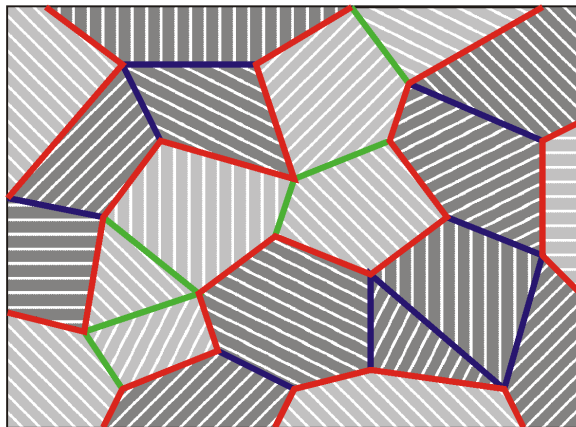
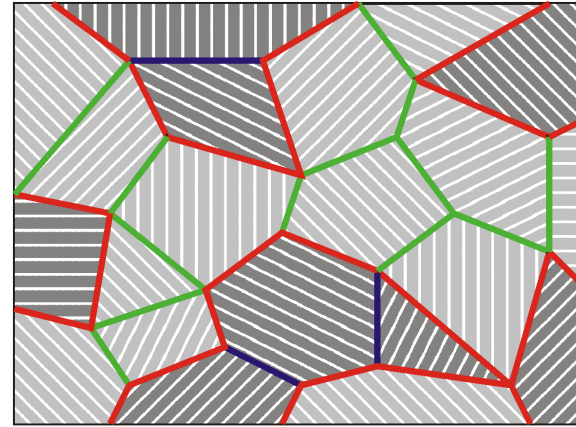
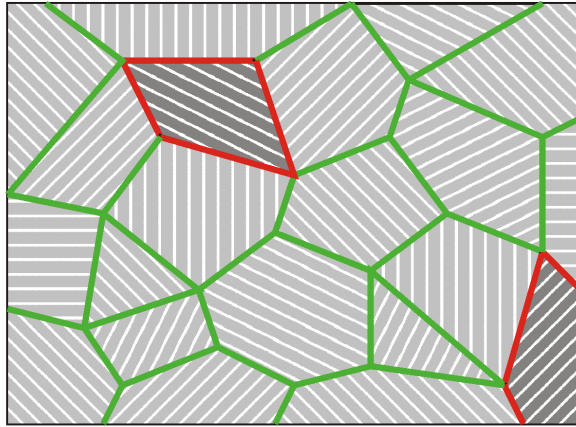
- ⇒ fast ions are located in the interfaces between ionic conductor and insulator
- ⇒ conductivity increases with insulator content  $x$
- ⇒ possible route to design fast solid electrolytes

# DC Conductivity: $(1-x) \text{Li}_2\text{O}:x\text{B}_2\text{O}_3$ $x=0\dots 1$






S. Indris et al.,  
Phys. Rev. Lett.  
87 (2000) 2889.

# Percolation Model



—  
 $\infty$   
 $\sigma_{dc}$

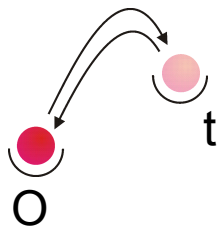
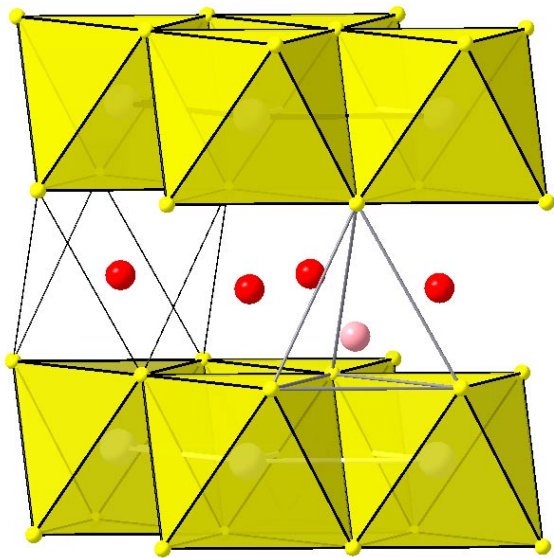
— /   
 $\infty$   
 $A_f$

 Ionic Conductor Grain  
 Insulator Grain

Interface between  
 — Insulators  
 — Ionic Conductors  
 — Ionic Conductor & Insulator

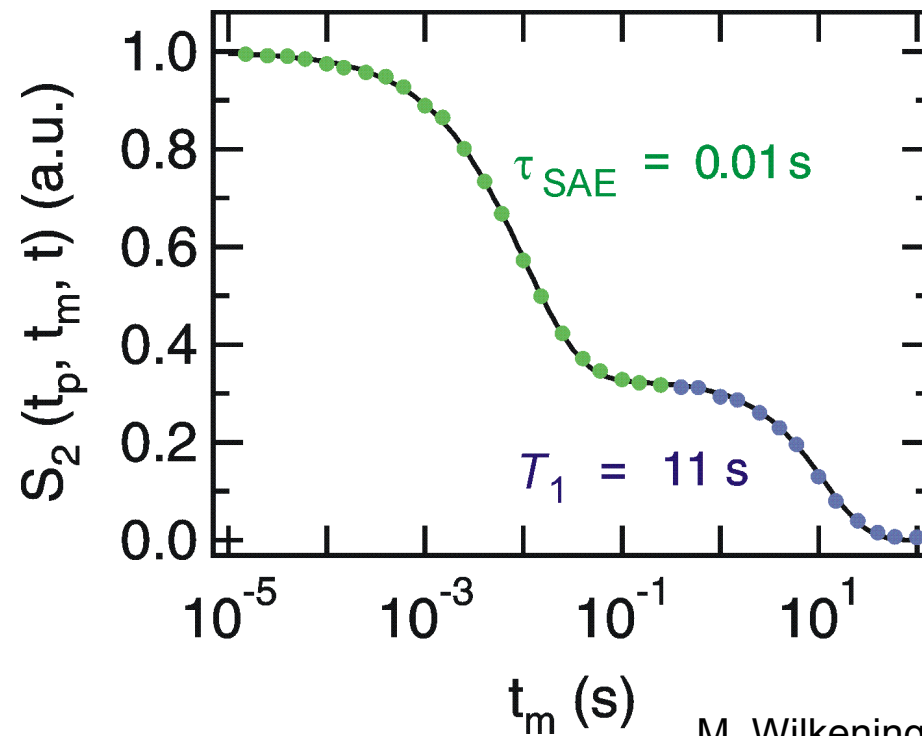
- $^7\text{Li}$  Spin-Alignment Echo

$$S_2(t_p, t_m) \propto \langle \sin(\omega_Q(0)t_p) \sin(\omega_Q(t_m)t_p) \rangle \exp\left(-\frac{t_m}{T_{1Q}}\right)$$



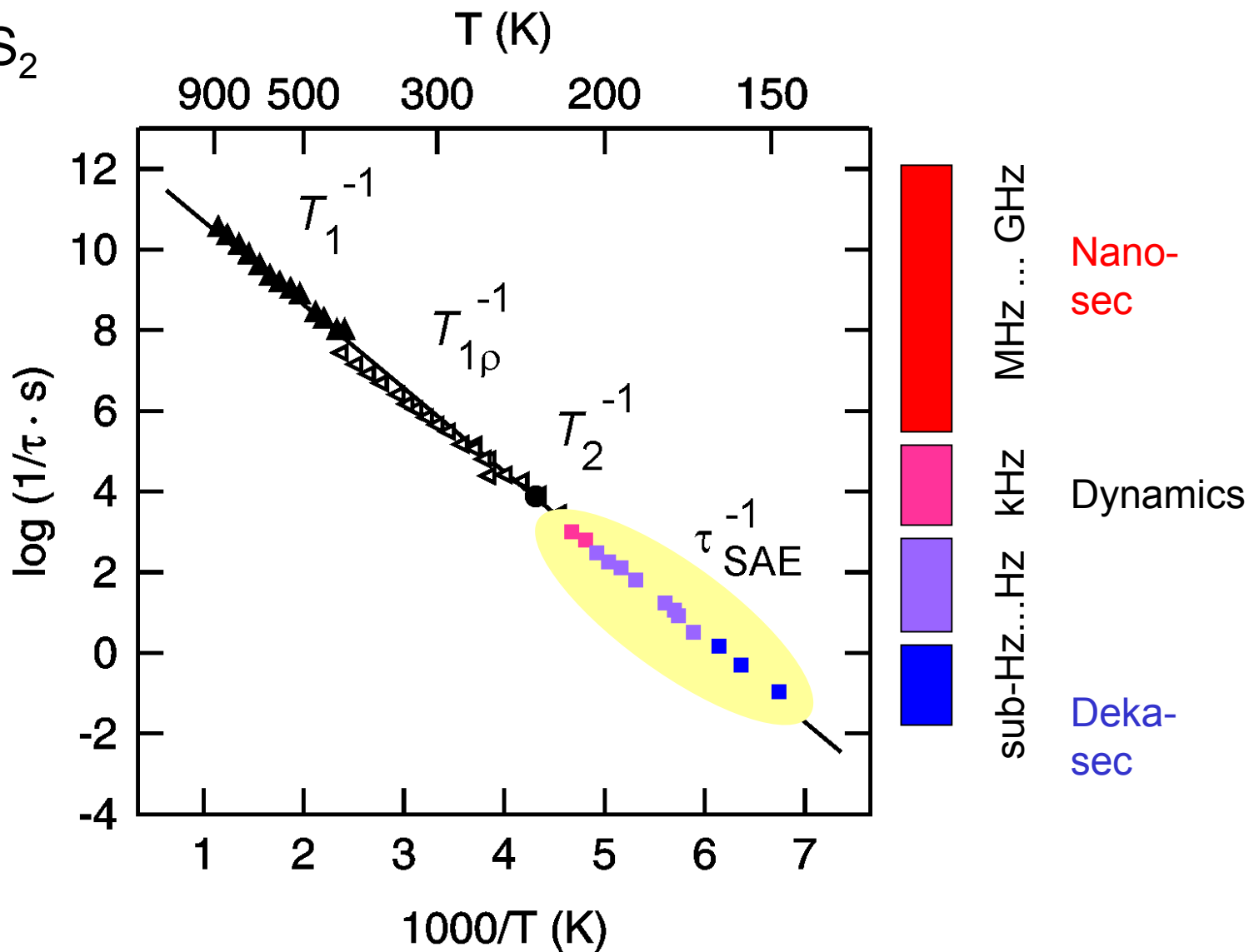
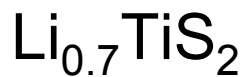
local  $\text{Li}^+$   
hopping

$\text{Li}_{0.7}\text{TiS}_2$  193 K, 155 MHz, 15  $\mu\text{s}$

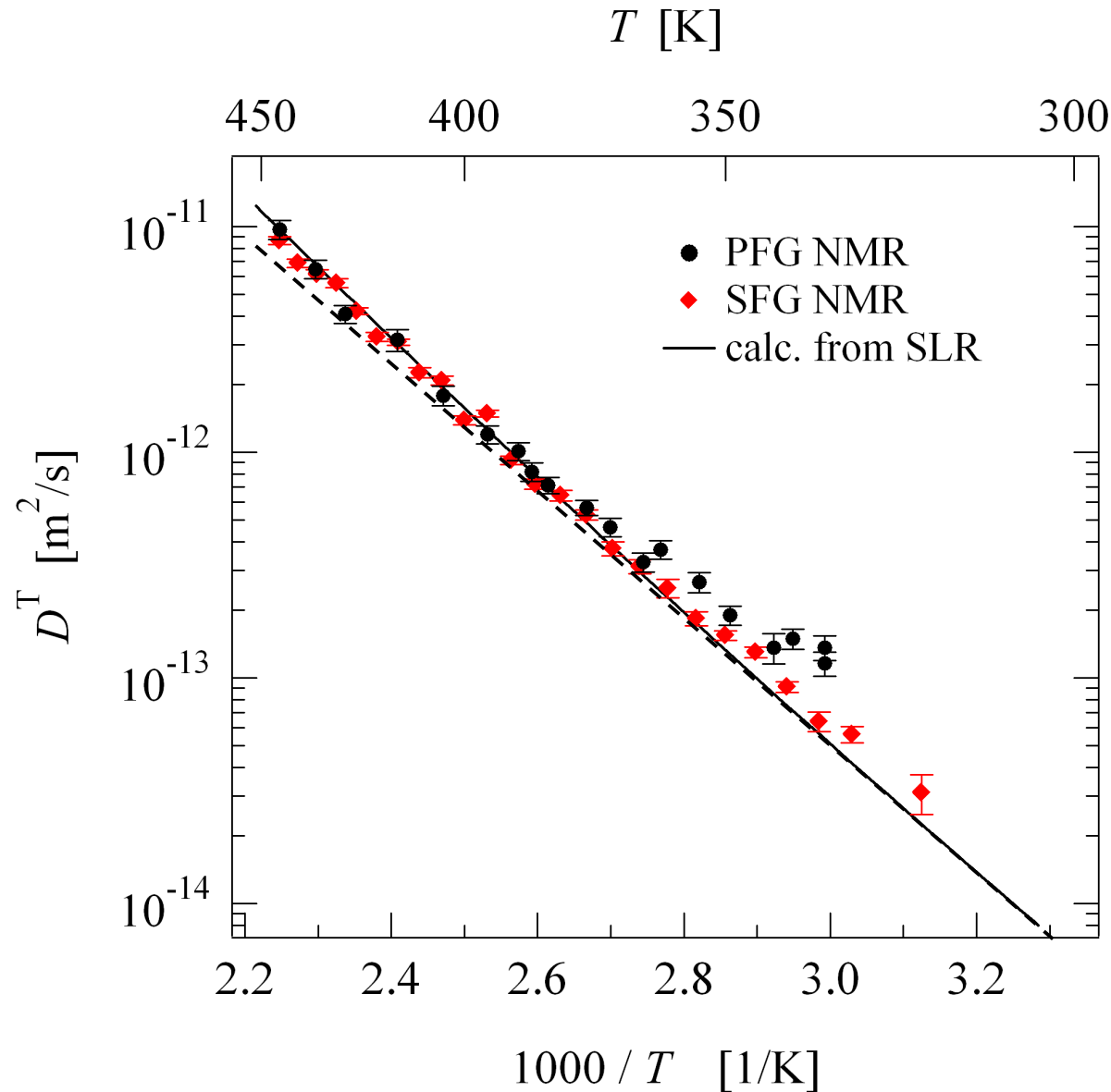


M. Wilkening, PhD thesis,  
University of Hannover, 2005.

# Motional Correlation Rates



- $^7\text{Li}$  SFG and PFG NMR  
on Solid Lithium as Simple Test Case



$D^T$  measured down to  
about  $10^{-14}$   $\text{m}^2/\text{s}$

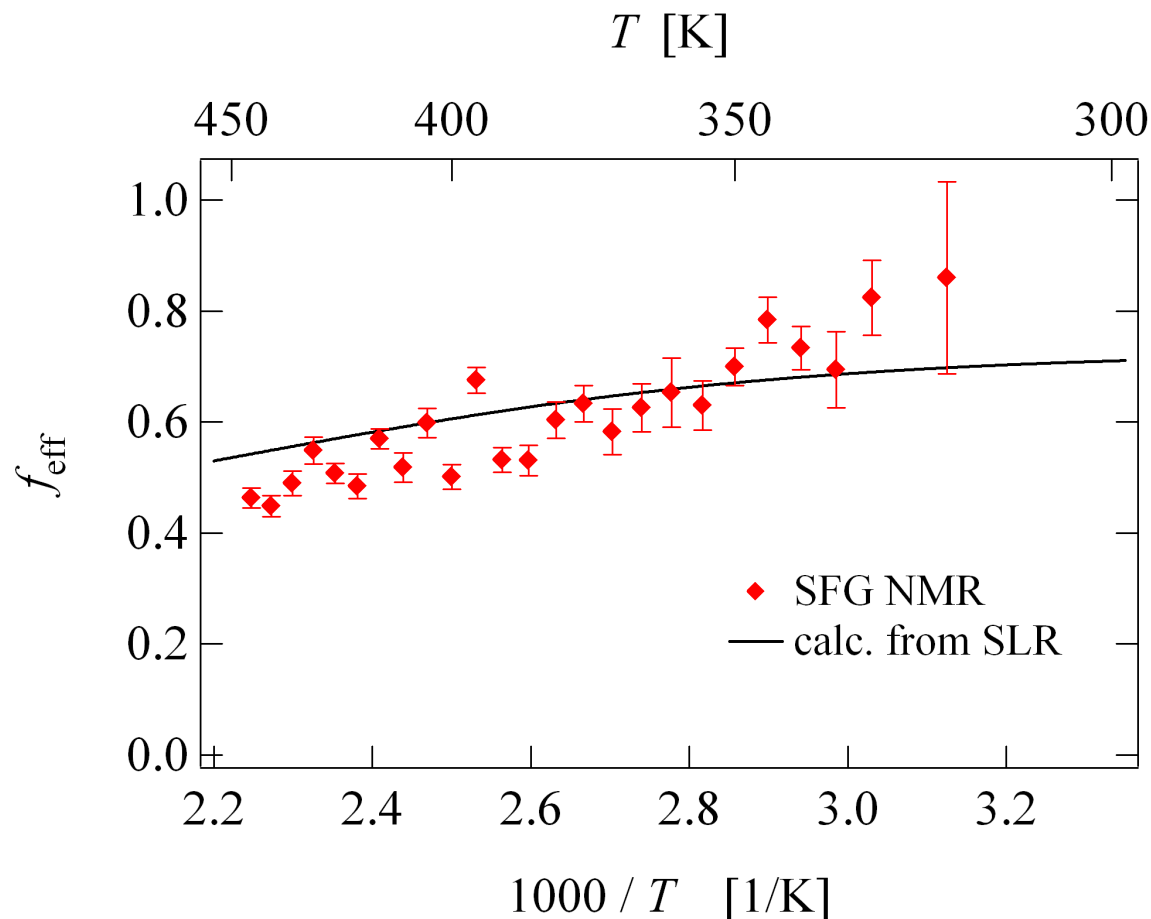
Comparison with  $T_1$ :

$D^T$  calculated from  
 $^8\text{Li}$  SLR data  
assuming  
Monovacancy-  
Divacancy-Mechanism

D. M. Fischer et al.,  
Solid State NMR, 26 (2004) 74

# $^7\text{Li}$ SFG NMR on Solid Lithium

effective correlation factor  $f_{\text{eff}} = \frac{D^{\text{T}}}{r^2 / 6\tau}$



$$D^{\text{T}} = f_{1\text{V}}D_{1\text{V}} + f_{2\text{V}}D_{2\text{V}}$$

$$f_{1\text{V}} = 0.727$$

$$f_{2\text{V}} = 0.347$$

(Mehrer 1973)

consistent with  
1V-2V mechanism

# Conclusion

- NMR provides arsenal of techniques

microscopic:  $T_1$ ,  $T_2$ ,  $T_{1\rho}$ ,  $\beta$ -NMR, SAE

macroscopic: SFG NMR, PFG NMR

- Used to measure jump rates ( $10^9 \dots 10^{-1} \text{ s}^{-1}$ ) and tracer diffusion coefficients ( $10^{-11} \dots 10^{-14} \text{ m}^2\text{s}^{-1}$ ) in

metals, glasses, ceramics, nanocrystals,  
intercalation compounds, solid electrolytes ...

- Comparison of microscopic and macroscopic diffusion parameters allows determination of diffusion mechanisms



# Acknowledgement

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DFG, BMBF, Land Niedersachsen